

Final Workshop

Passive Sampler Intercomparison Exercise

C. Miège, N. Mazzella, S. Schiavone, A. Dabrin, M. Coquery: Cemagref - Lyon, Bordeaux

C Berho, J-P Ghestem: BRGM – Orleans

J-L Gonzalez, D Munaron, C. Tixier: Ifremer - La Seyne/Mer, Sète, Nantes

B. Lalere, S. Lardy-Fontan: LNE - Paris

B. Lepot: INERIS – Paris

C. Gonzalez: EMA - Ales

Results for Metals

A. Dabrin, J-P. Ghestem, J-L. Gonzalez, M. Coquery

▶ 10 expert laboratories



- 5 French and 5 other European countries laboratories (Italy, Spain, United Kingdom, Sweden, Norway)
- Various strategies :
 - Standard commercial or home-made passive samplers (PSs) : DGT open pores, DGT restrictive pores, Chemcatcher
 - With home-made exposure systems
 - Analytical treatment
 - Using diffusion coefficients/uptake rates from literature

Passive samplers and exposure durations

8 metals

- Cadmium*
- Chromium*
- Lead*
- Nickel*
- Manganese
- Zinc*
- Copper*
- Cobalt

devices

- DGT (Diffusive Gradient in Thin films)
Open pores
Restrictive pores
Chelex-100
- Chemcatcher
Empore chelating disk

→ 7 days



**Priority substances (WFD)*

***Substances of the ecological status**

- Tools were exposed in triplicates and field blanks (brought to the field but not exposed in waters) were used

► Sampling sites

- 2 contrasted environments

Coastal environment

Thau Lagoon (Hérault)
 27 April-5 May 2010



Former site of oyster farming

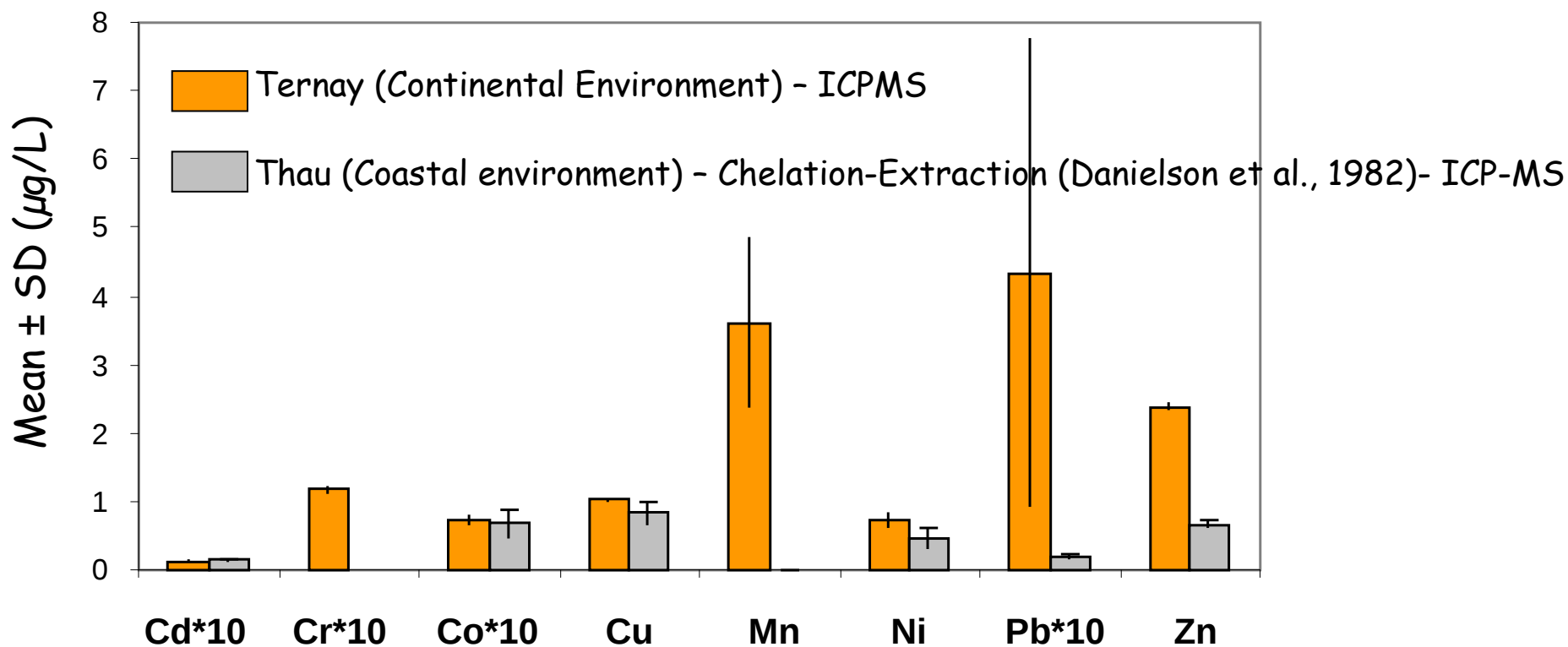
Continental environment

Rhône River Ternay site
 17-24 June 2010



► Spot sampling concentrations

3 spot sampling :
Start, during and at the end of the PSs deployment



Comparison of passive sampling concentrations from various tools and laboratories

	Ternay	Thau		
Number of participants	10	6	Nb results/ nb tools (%)	
Number of Tools	13	7		
Metals	Number of results		Ternay	Thau
Cd	12	7	92	100
Ni	13	7	100	100
Pb	12	6	92	86
Cu	13	7	100	100
Cr	11	7	85	100
Zn	10	5	77	71
Co	8	6	62	86
Mn	11	7	85	100

- Two times more results were obtained for the exercise at Ternay site than Thau
- Tools were lost or some laboratories did not give results for some metals
- Percentage of results compared with the number of tools :
 - Ternay : from 62 to 100 %
 - Thau : from 71 to 100%

▶ Statistical data treatment and methodology



- Arithmetic means and reproducibility standard deviations S_R (ISO 5725-2)

- Robust statistics: ISO 5725-5

No exclusion from laboratories with outliers results

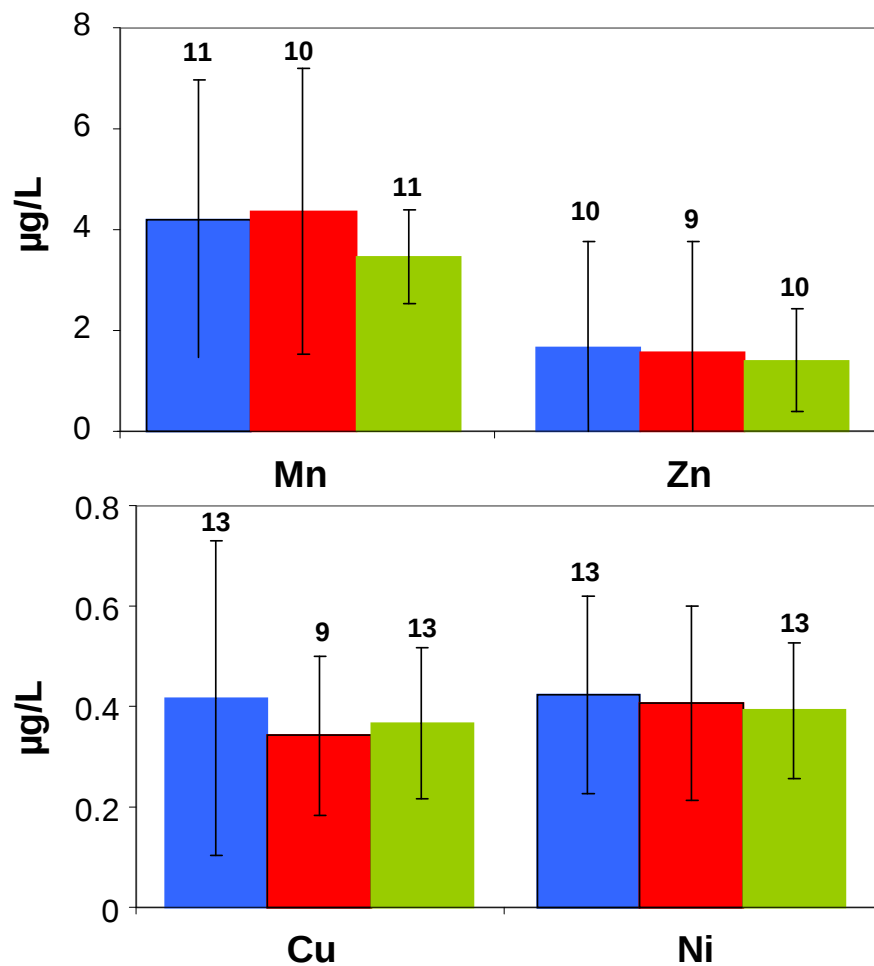
Data was processed to minimize the weight of suspect values

- Comparison of:

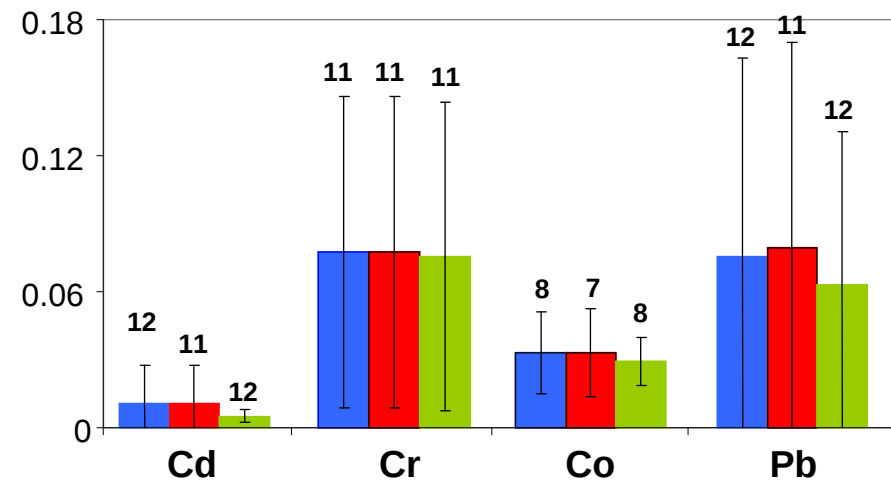
- Arithmetic means and S_R with data of all lab.
- Arithmetic means and S_R after elimination of QC outliers
- Robust means (x^*) and S_R with data of all lab.

Water concentrations ($\mu\text{g/L}$) for metals – passive samplers

• Ternay site:



■ Means and standard deviations (all participants)
■ Means and standard deviations (without QC outliers)
■ Robust statistic (all participants)



- Robust approach allows to decrease the means and the standard deviations

► Data dispersion of passive samplers

Passive sampler data							SWIFT-WFD proficiency Testing Exercise (2006)			
Metals	Ternay			Thau			LQ Water μg/L	Robust mean		
	Robust mean	RSD	n	Robust mean	RSD	n		Robust mean	RSD	n
	x* ± SD (μg/L)			x* ± SD (μg/L)				x* ± SD (μg/L)		
Cd	0.005 ± 0.003	58	12	0.027 ± 0.025	92	7	0.010	0.09 ± 0.08	89	27
Cr	0.076 ± 0.070	93	11	0.036 ± 0.029	80	7	0.050	1.73 ± 1.57	91	36
Cu	0.367 ± 0.153	42	13	0.233 ± 0.1089	47	7	0.050	4.15 ± 1.66	40	42
Mn	3.47 ± 0.99	28	11	7.48 ± 2.646	35	7	0.100	154 ± 17	11	47
Ni	0.392 ± 0.139	35	13	0.261 ± 0.1265	48	7	0.050	1.85 ± 1.40	75	32
Pb	0.063 ± 0.070	112	12	0.021 ± 0.012	58	6	0.010	1.20 ± 0.83	69	31
Zn	1.40 ± 1.10	79	10	3.15 ± 3.13	99	5	0.500	12.3 ± 2.8	23	39

- Comparison with a classical proficiency testing exercise (analytical) :

- Higher dispersion of PSs data for Pb, Zn, Mn
 - Similar dispersion of PSs for Cd, Cr, Cu
 - Lower dispersion for Ni

- However, much lower concentrations determined by passive samplers

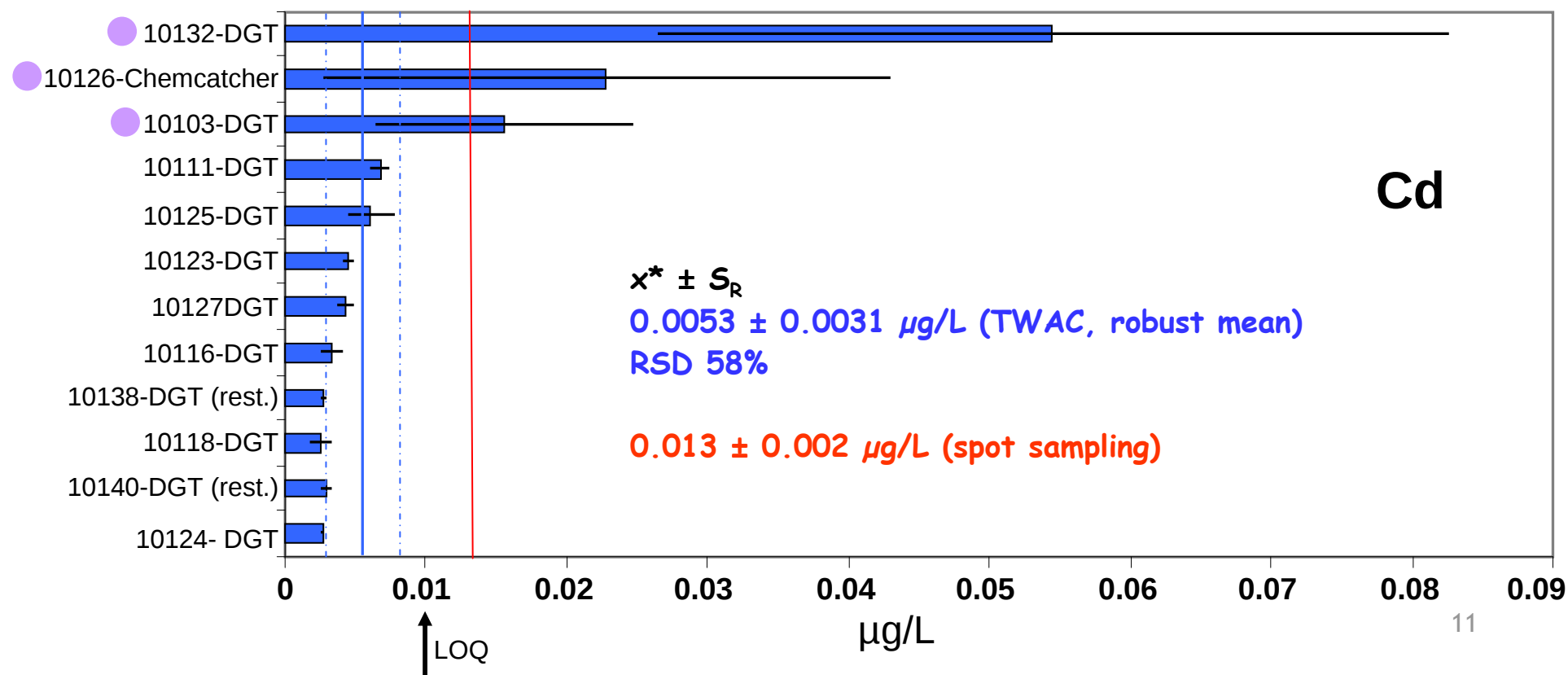
- Moreover, reproducibility for PS includes both analytical and sampling steps
Since analytical variability was low in this exercise (from 8 to 25%, from 4 to 44%), the dispersion was mainly due to PS step

Comparison of passive sampling results from various tool and lab

- For Ternay site:

Aberrant values

QC	● (Cochran test) within-laboratory variability
	● (Grubbs test) between-laboratory variability
PSs DATA	● (Cochran test) within-laboratory variability
	● (Grubbs test) between-laboratory variability

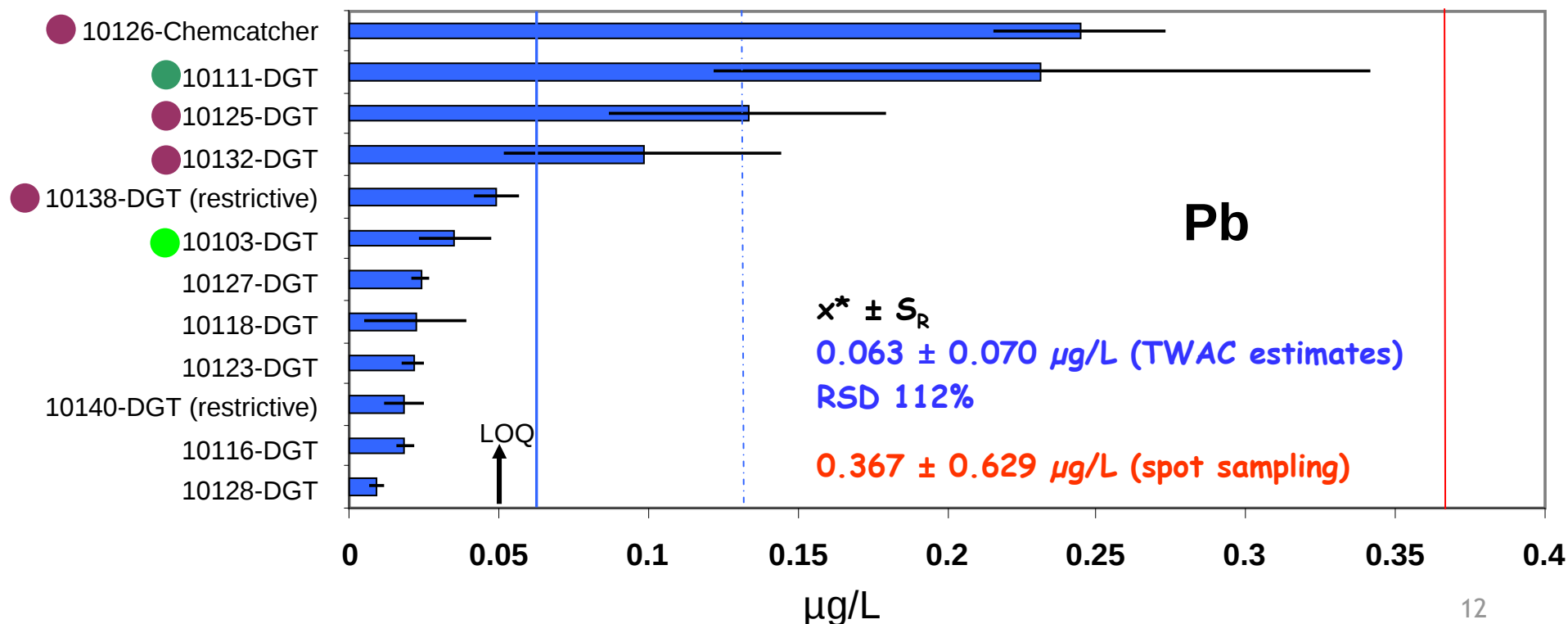


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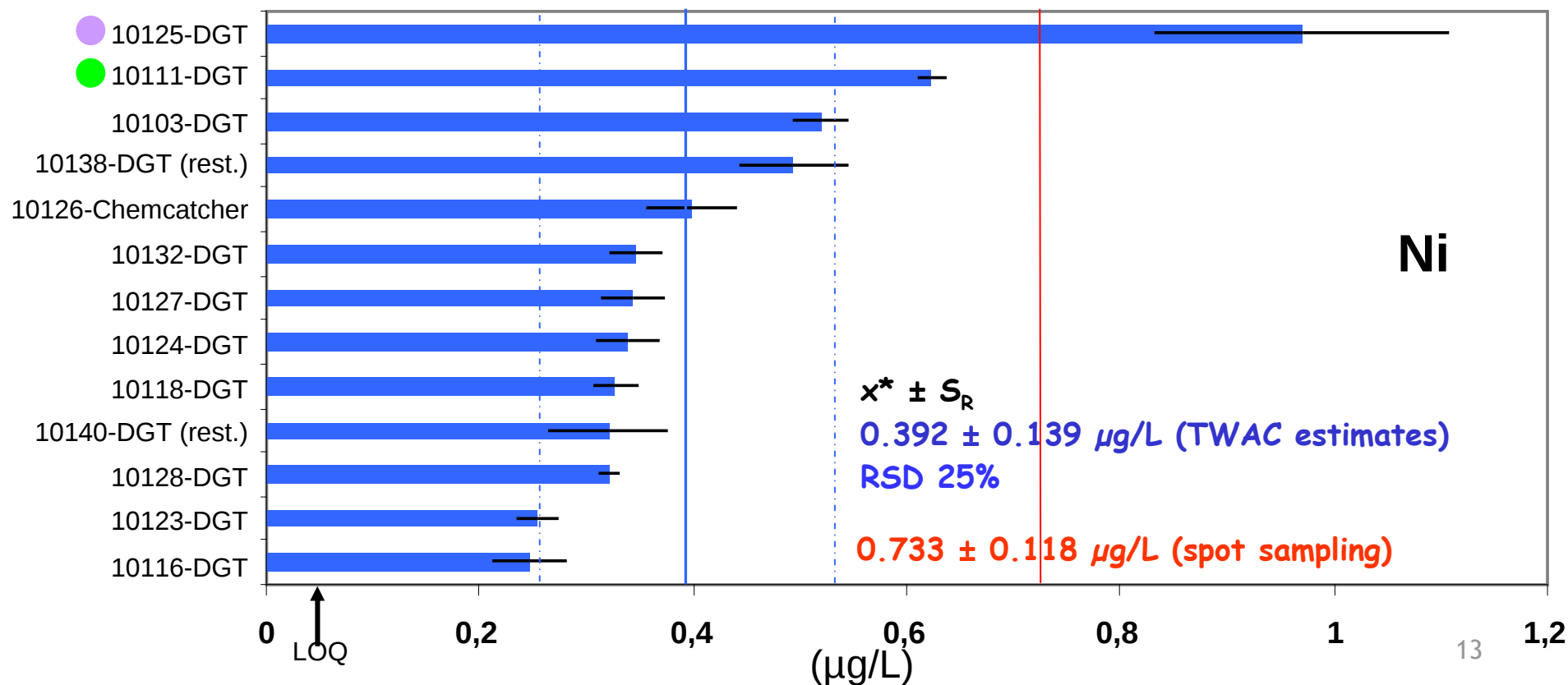


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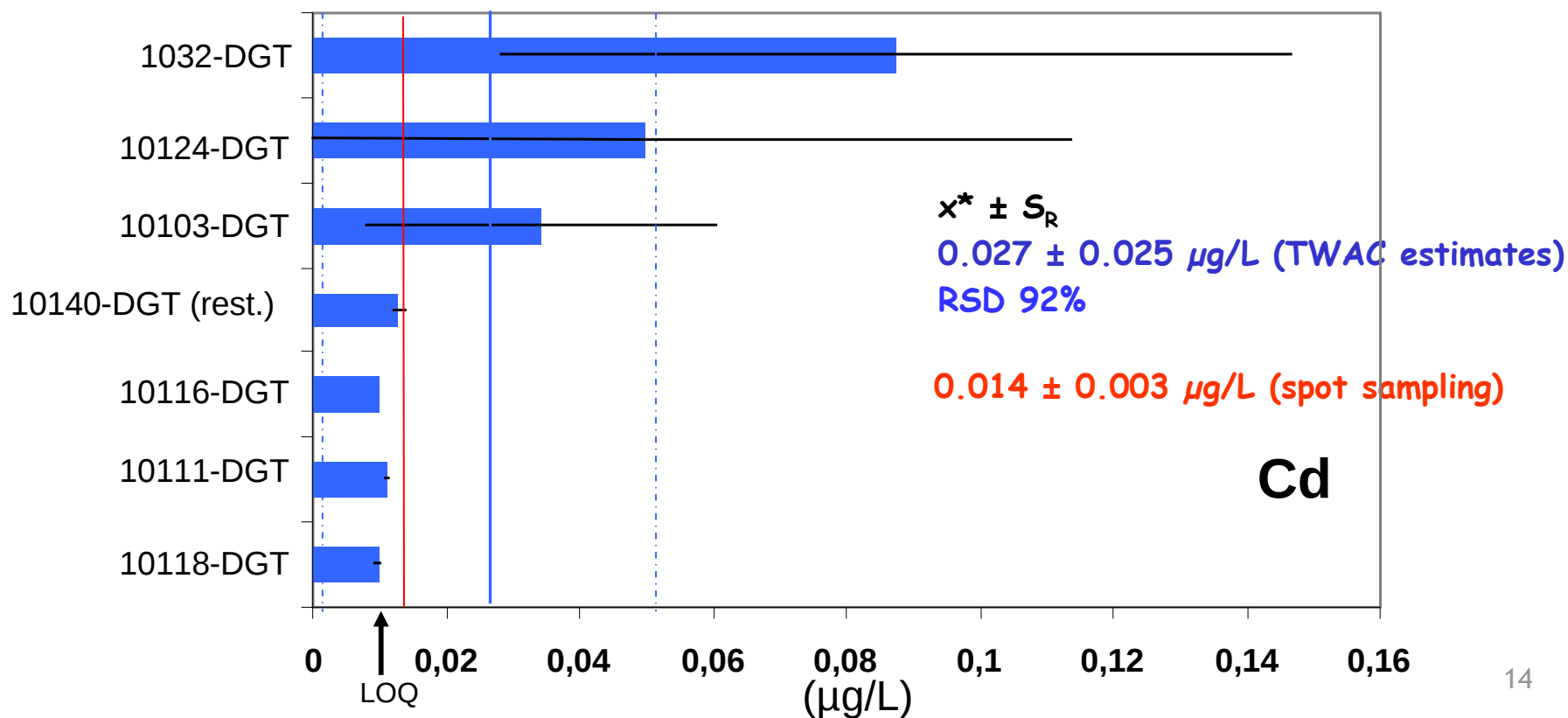


► Comparison of passive sampling results from various tool and lab

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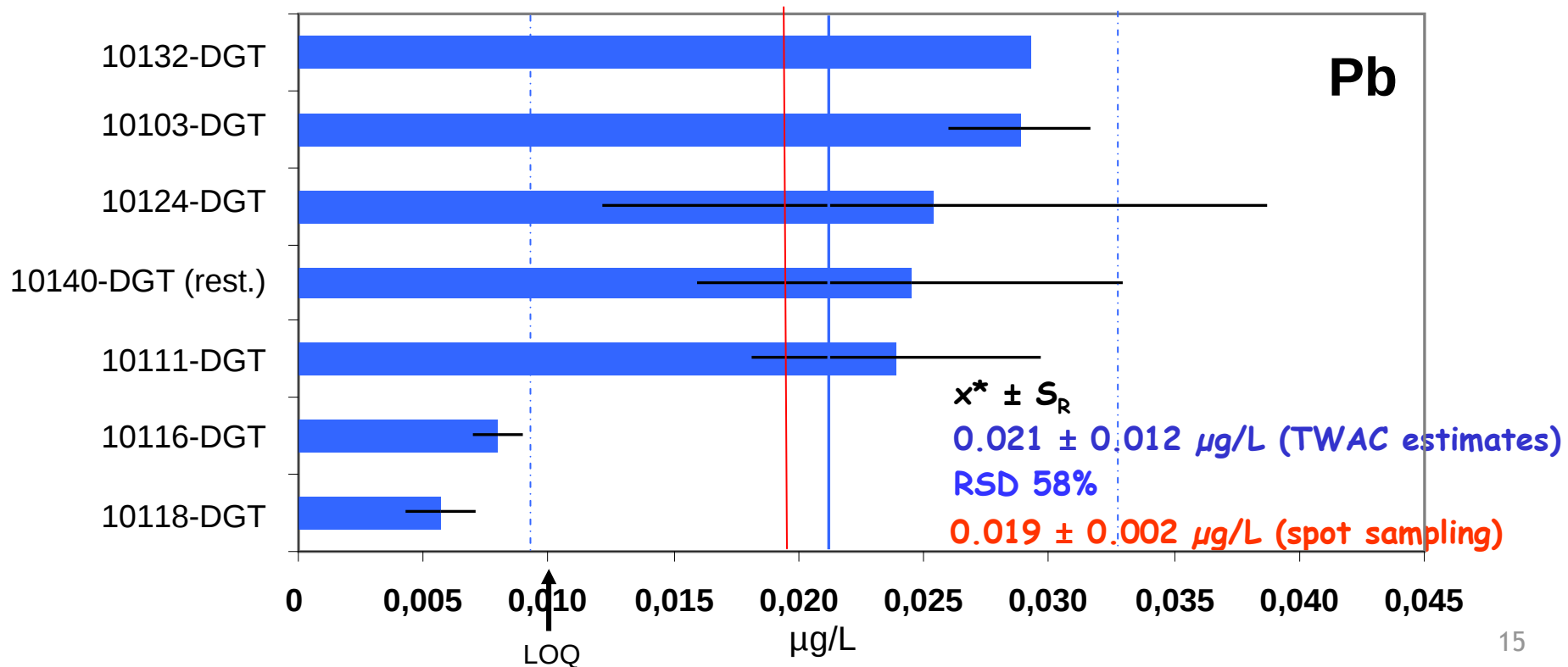


Comparison of passive sampling results from various tool and lab

- For Thau site:

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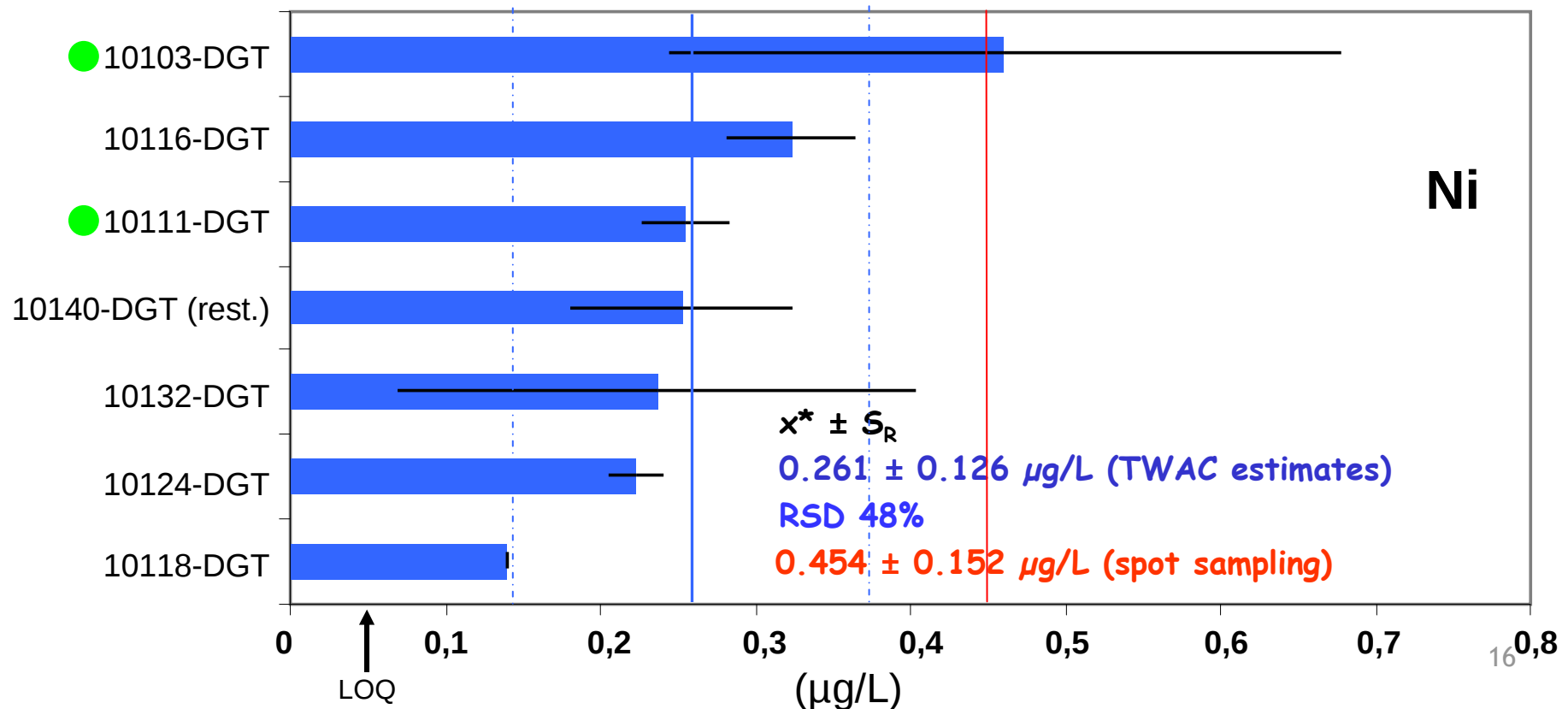


Comparison of passive sampling results from various tool and lab

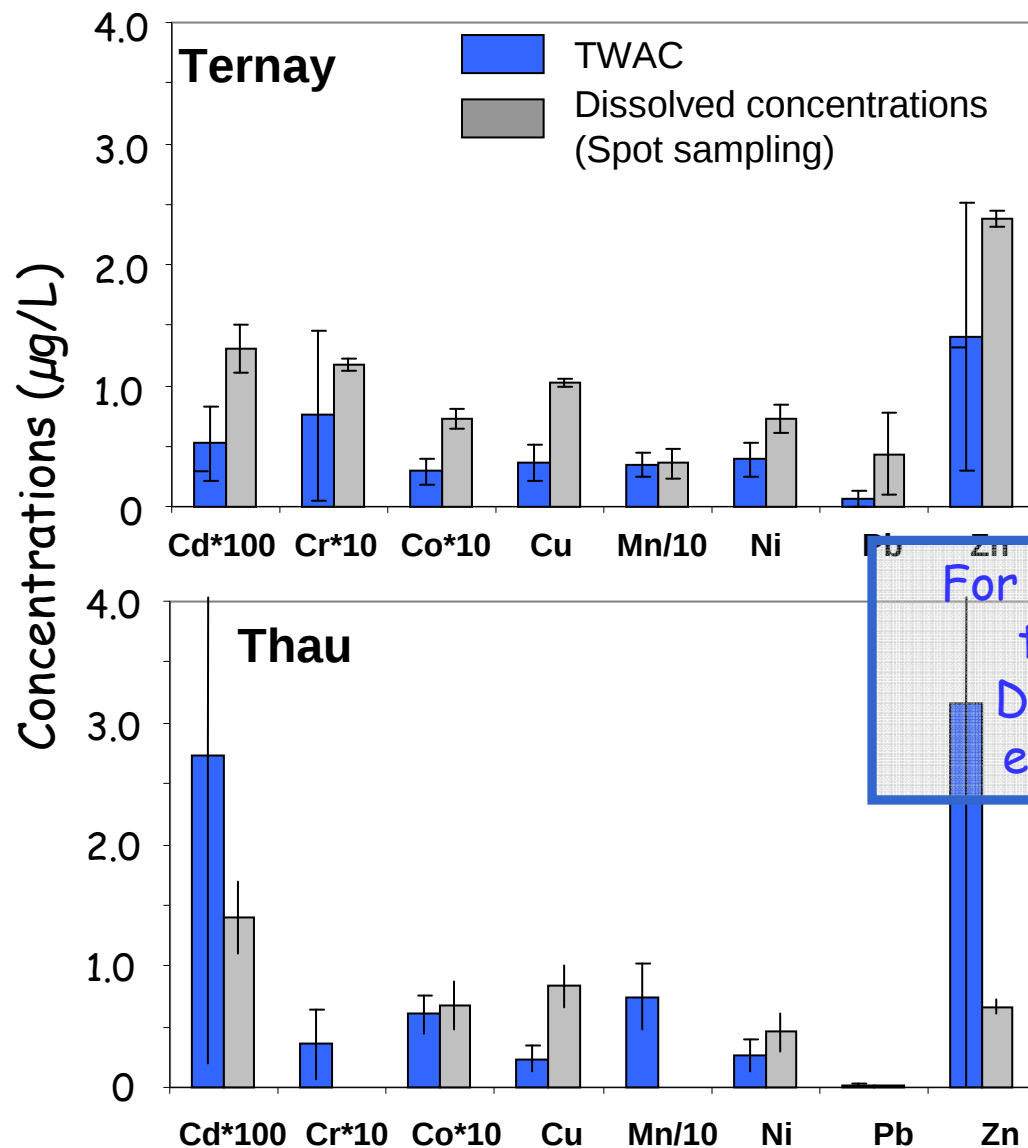
- For Thau site:

Aberrant values

QC	● (Cochran test) within-laboratory variability
	● (Grubbs test) between-laboratory variability
PSs DATA	● (Cochran test) within-laboratory variability
	● (Grubbs test) between-laboratory variability



Comparison of TWAC and spot sampling (Dissolved concentrations)



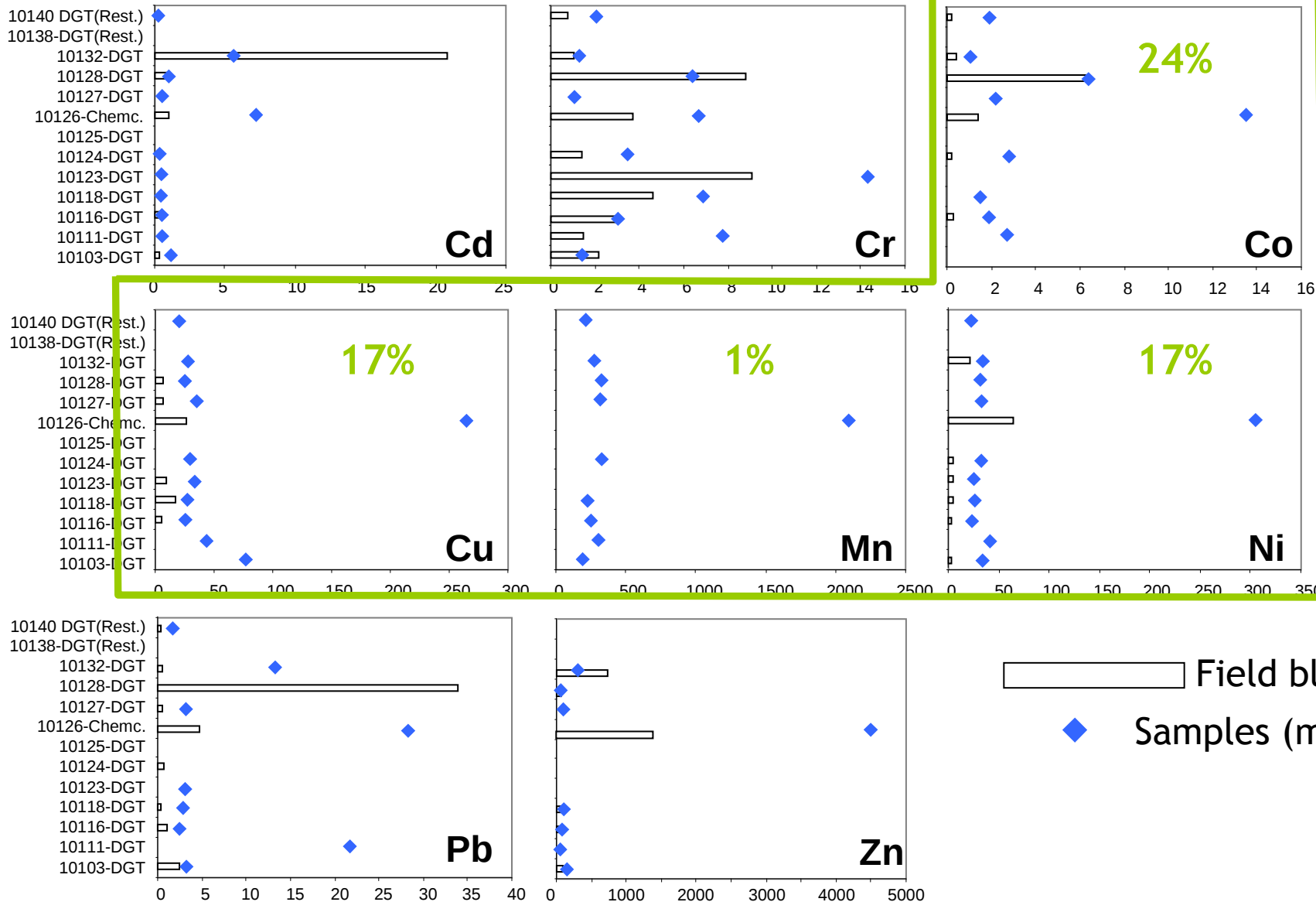
- 100 % of total dissolved Mn was sampled by PSs

- Only 35% of Cu was sampled by PSs

For metals, PSs only « see » a part of total dissolved concentrations, Depends on the metal and on the environmental conditions (DOM)

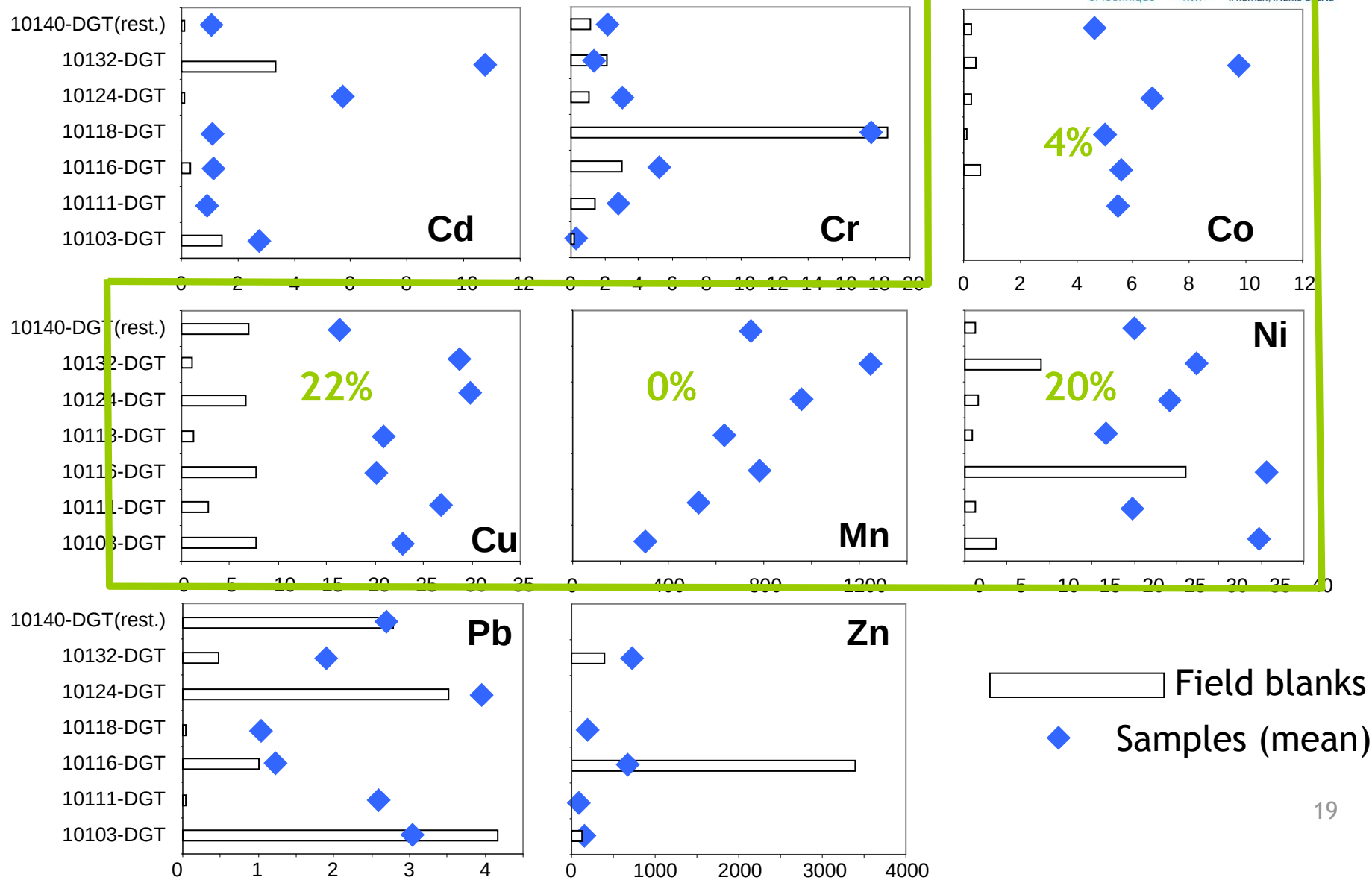
Field blanks for metals (ng/tool)

Ternay : 2 lab. substracted field blanks



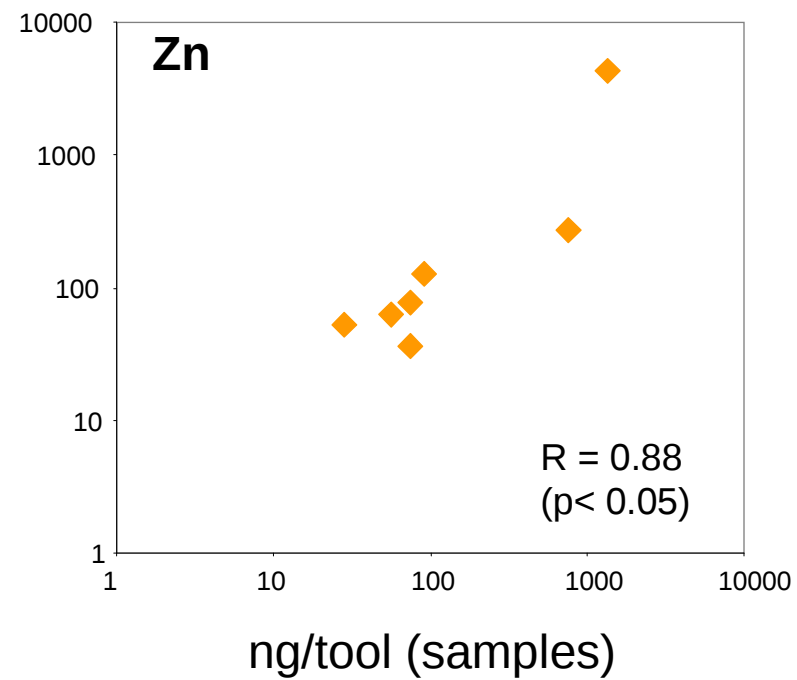
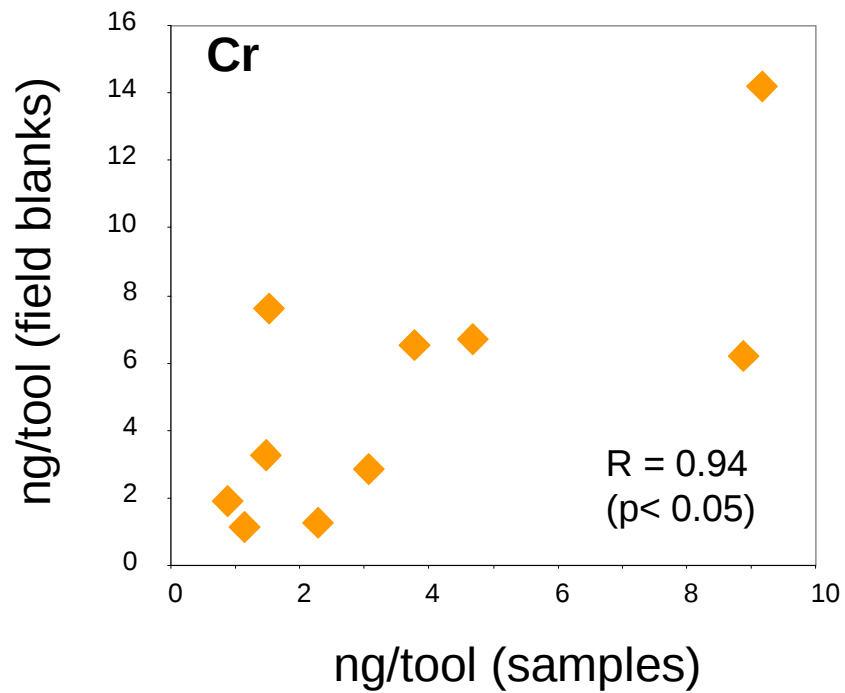
Field blanks for metals (ng/tool)

Thau : 1 lab. subtracted field blanks

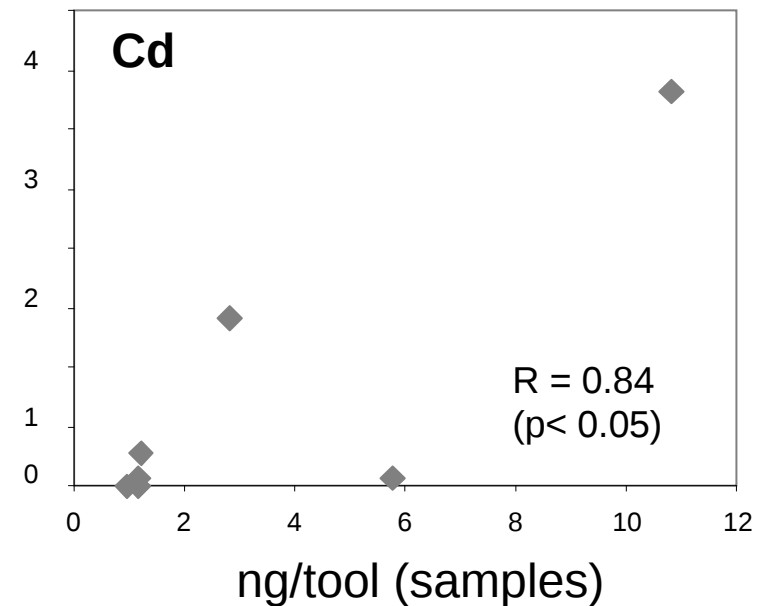
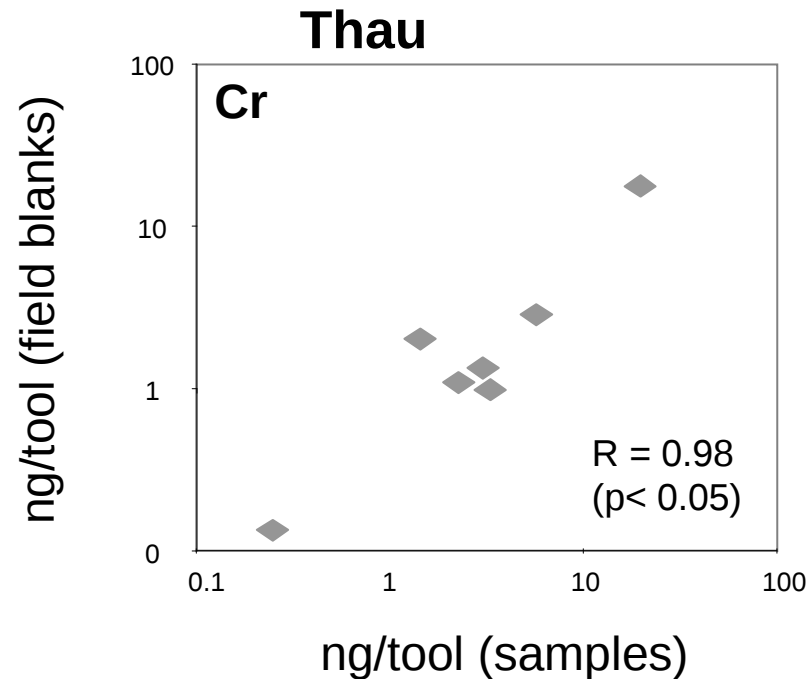


► Field blanks for metals

Ternay



► Field blanks for metals



- Field blanks are partly responsible for PSs TWAC variability in these exercises:

Cr, Zn : Ternay
Cr, Cd : Thau

- In other cases, field blanks are high but there is no relationship

- For all metals, there is a need to better determine contamination origin : by discriminating field blanks and lab-blanks

► Conclusions and perspectives



- Estimation of water concentrations by passive sampling for metals shows low and satisfying variability, considering various lab, strategies and tools.
- RSD are comparable to analytical interlab. Exercise (SWIFT)
- Since analytical interlab. variability was low in this exercise (from 8 to 44%), the variability was mainly due to PS step
- PSs allow to measure low concentrations
- PSs allow to facilitate the measurement of some metals in saline matrix
- After this exercise, difficult to conclude for use a better tool since only one chemcatcher and two DGT with restrictive pores were used
- For metals, PSs only see a part of total dissolved concentrations, and depends on the metal and the environment
- Contamination of field blanks (in particular for Cr, Cd, Zn, Pb) is partly responsible for DGT TWAC variability

► Conclusions and perspectives



- Need to discriminate sources of PS uncertainties for each lab (including steps of assembly, deployment, dismantling, elution, ...)
 - by obtaining lab-blanks for each laboratory and to compare with field blanks

- Need to compare more precisely Chemcatcher, DGT open and restrictive pores

- Considering WFD:
 - A need of detailed protocols for non expert lab. (to better control blanks)
 - A need to clarify the fraction which is sampled by these tools in contrasted environment and during contrasted conditions

Thanks to the participant lab



- **ALS Scandinavia AB (SW)**
- **AZTI-Foundation (ES)**
- **BRGM (FR)**
- **Cefas (UK)**
- **Cemagref (FR)**
- **EDF R&D/LNHE (FR)**
- **IFREMER (FR),**
- **NIVA (NO)**
- **Universita di Cagliari (IT)**

► Thanks to central lab for water analysis



- **IFREMER (metals and physico-chemical parameters in Thau site)**
- **Cemagref of Lyon (metals and physicochemical parameters at Ternay site)**
- **Ineris for data treatment**

Thank you for your attention