ANNEX I: List of ISO Standards for soil analysis

ISO 11465:1993	Soil quality - Determination of dry matter and water content on a mass basis - Gravimetric method
ISO 11466:1995	Soil quality - Extraction of trace elements soluble in aqua regia
ISO 11277:1998	Soil quality - Determination of particle size distribution in mineral soil material - Method by sieving and sedimentation
ISO 10694:1995	Soil quality - Determination of organic and total carbon after dry combustion (elementary analysis)
ISO 14869-1:2001	Soil quality - Dissolution for the determination of total element content - Part 1: Dissolution with hydrofluoric and perchloric acids
ISO 11047:1998	Soil quality - Determination of cadmium, chromium, cobalt, copper, lead, manganese, nickel and zinc - Flame and electrothermal atomic absorption spectrometric methods
ISO 14507:2003	Soil quality - Pretreatment of samples for determination of organic contaminants
ISO 14154:2005	Soil quality - Determination of some selected chlorophenols - Gas- chromatographic method with electron-capture detection
ISO 15009:2002	Soil quality - Gas chromatographic determination of the content of volatile aromatic hydrocarbons, naphthalene and volatile halogenated hydrocarbons - Purge-and-trap method with thermal desorption
ISO 16772:2004	Soil quality - Determination of mercury in aqua regia soil extracts with cold-vapour atomic spectrometry or cold-vapour atomic fluorescence spectrometry
ISO 22155:2005	Soil quality - Gas chromatographic quantitative determination of volatile aromatic and halogenated hydrocarbons and selected ethers - Static headspace method
ISO 11264:2005	Soil quality - Determination of herbicides - Method using HPLC with UV-detection
ISO 10382:2002	Soil quality - Determination of organochlorine pesticides and polychlorinated biphenyls - Gas-chromatographic method with electron capture detection
ISO 13877:1998	Soil quality - Determination of polynuclear aromatic hydrocarbons - Method using high-performance liquid chromatography
ISO 18287:2006	Soil quality - Determination of polycyclic aromatic hydrocarbons (PAH) - Gas chromatographic method with mass spectrometric detection (GC-MS)
ISO/DIS 22036	Soil quality - Determination of trace elements in extracts of soil by inductively coupled plasma atomic emission spectrometry (ICP/AES)
ISO 22892:2006	Soil quality - Guidelines for the identification of target compounds by gas chromatography and mass spectrometry
ISO/DIS 23161	Soil quality - Determination of selected organotin compounds - Gas- chromatographic method

ANNEX II: Substance Guidance Sheets

Compound Alachlor							
CAS N	umber 15972-60-8		Log K ~ 2.9		Wat	er Solubility [mg/L] ~ 240	
	AA-EQ	S [µg/L]			MAC-	EQS [µg/L]	
Inland	Surface Waters 0.3	Other Su	rface Waters 0.3	Inland Su	urface Waters 0.7	Other Surface Waters 0.7	
	ole Standard Meth	nod		Method I	Description		
EN ISO 10695:2000 Determination of selected organic nitrogen and phosphorus compounds - Gas chromatography methods [Note: Alachlor is not explicitly mentioned in this standard but the method may also be applied to the analysis of alachlor provided the method has been properly validated for this compound]. <u>Matrix</u> Drinking waters, ground waters, surface waters and waste waters containing up to 50 mg/L of suspended solids <u>Sampling</u> ISO 5667-2:1991			liquid/sol material c e.g. metha After con gas chron	d extraction (SP r other adsorben nol or acetone. centration, the sa	ith dichloromethane or E) on reversed-phase (RP)-C18 t. Elution of the cartridges with ample extracts are analysed by g a nitrogen-phosphorus or or.		
Pretrea	atment			Limit of	Quantification (LOQ): 0.04 μg/L	
Storage	2						
	I Validation: no d		e				
SPE-GO Separat coupled GC-MS SPE ext SPE ext		tography, id etric detectio z 160, 188, water; LOQ er; LOQ ~ 3	on (GC-MS) usi 161 and 146 [1-) ~ 12 ng/L [1] ng/L [2]	ng electron		e by gas chromatography de.	
<u>GC-NP</u> EPA me	<u>D</u> ethod 507 [4]						
	<u>-MS/MS</u> E of 50 mL water;	MRM 270	> 161.5; LOQ ~	0.1 μg/L [:	5]		
	On-line SPE-LC-MS-MS On-line SPE of 10 mL samples; MRM 270 > 238; LOQ ~ 47 ng/L [6]						
Referen	nces						
[1]							
[2]		ic Pesticide	s in Natural Wat			taneous Determination of er Liter Level. <i>Analytical</i>	

[3]	D. de Almeida Azevedo, S. Lacorte, T. Vinhas, P. Viana, D. Barceló, Monitoring of Priority Pesticides and Other Organic Pollutants in River Water From Portugal by Gas Chromatography– Mass Spectrometry and Liquid Chromatography–Atmospheric Pressure Chemical Ionization Mass Spectrometry. <i>Journal of Chromatography A</i> 879, 2000, 13-26.
[4]	D. Barceló, Environmental Protection Agency and Other Methods for the Determination of Priority Pesticides and Their Transformation Products in Water. <i>Journal of Chromatography A</i> , 643, 1993, 117-143.
[5]	R A Yokley L C Mayer S-B Huang L D Vargo Analytical Method for the Determination of

- [5] R. A. Yokley, L. C. Mayer, S.-B. Huang, J. D. Vargo, Analytical Method for the Determination of Metolachlor, Acetochlor, Alachlor, Dimethenamid, and Their Corresponding Ethanesulfonic and Oxanillic Acid Degradates in Water Using SPE and LC/ESI-MS/MS. *Analytical Chemistry* 74, 2002, 3754-3759.
- [6] M. Kuster, M. J. Lopez de Alda, C. Barata, D. Raldua, D. Barceló, Analysis of 17 polar to semi-polar pesticides in the Ebro river delta during the main growing season of rice by automated on-line solidphase extraction-liquid chromatography-tandem mass spectrometry. *Talanta* 75, 2008, 390-401.

Compound Anthr							
CAS Number		Log K	ow	Wat	er Solubi	lity [mg/L]	
120-12-7		4.55			0.043	34	
AA-EQ	S [µg/L]			MAC-	EQS [µg/	L]	
Inland Surface Waters 0.1	Other Su	rface Waters 0.1	Inland S	urface Waters 0.4	Other S	urface Waters 0.4	
Available Standard Meth EN ISO 17993:2003 Determination of 15 polyc hydrocarbons (PAH) in w fluorescence detection after Matrix Drinking, ground, Sampling Pretreatment Storage Method Validation	Method Description EN ISO 17993:2003 specifies a method using high performance liquid chromatography (HPLC) with fluorescence detection for the determination of 15 selected polycyclic aromatic hydrocarbons (PAH). Limit of Quantification (LOQ): 0.01 μ g/L l Number of laboratories n_{AP} percental rate of outliers s_R standard deviation between the laboratories CV_R reproducibility variation coefficient						
Matrix	l	n _{AP} in %	= x in µ	g/L s _R in f	ug/L	CV _R in %	
Drinking water	33	0	26.8	4 4.4	74	16.7	
Other Analytical Methods USA EPA 8270c, 1996 [1] Semivolatile organic compounds by GC/MS This method claims detection limits of 10 µg/L which is obviously too high. Other analytical methods based on this standard using modern GC/MS equipment however may attain the required low LOQs. Comments References [1] http://www.accustandard.com/asi/pdfs/epa_methods/8270c.pdf							

Compound Atrazine						
CAS Number 1912-24-9		Log K ~ 2.5		Wat		bility [mg/L] 33
AA-EQ	S [µg/L]			MAC-	EQS [µ	g/L]
Inland Surface Waters 0.6	Other Sur	face Waters 0.6	Inland Su	urface Waters 2.0	Other	Surface Waters 2.0
Available Standard Method EN ISO 10695:2000 Determination of selected organic nitrogen and phosphorus compounds - Gas chromatography methods Matrix Drinking waters, ground waters, surface waters and waste waters containing up to 50 mg/L of suspended solids Sampling ISO 5667-1 and 5667-2			Method DescriptionLiquid/liquid extraction with dichloromethane or liquid/solid extraction (SPE) on reversed-phase (RP)-C18 material or other adsorbent. Elution of the cartridges with e.g. methanol or acetone. After concentration, the sample extracts are analysed by gas chromatography, using a nitrogen-phosphorus or mass spectrometric detector.			
<u>Pretreatment</u> <u>Storage</u>			Limit of Quantification (LOQ): Liquid/liquid extraction method: 0.5 μg/L Liquid/solid extraction method: 0.015 μg/L			
Interlaboratory study 199	3 for liquid	/solid extraction	1	s _R star	ndard devia	<i>l</i> Number of laboratories n_{AP} percental rate of outliers n after elimination of outliers thion between the laboratories hubbli y variation coefficient
Matrix	l	n _{AP} in %	= x in µ	g/L s _R in J	ug/L	CV _R in %
Drinking water	13	0	0.133	3 0.01	04	35.6
Other Analytical Methods Gas Chromatography - Mass Spectrometry GC-MS determination of the ions 200 and 215; LOQ ~ 1 ng/L (after SPE) [1-3] (EPA method 525) <u>GC-NPD</u> EPA method 507 [4] <u>GC-ECD</u> EPA method 505; microextraction with hexane and GC-ECD analysis [4]] Liquid Chromatography - Mass Spectrometry Identification and quantification of atrazine (and other pesticides) by liquid chromatography coupled to (tandem) mass spectrometric detection (LC-MS-MS) using positive electrospray ionization (ESI) LC-MS fragment ions: m/z 216 and 174 [5] LC-MS-MS transitions: 216 > 174 and 132 [6] LOQ ~ 1 ng/L (depending on the SPE enrichment factor)						
<u>Comments</u>						

- [1] Z. Cai, V. M. S. Ramanujam, D. E. Giblin, M. L. Gross, and R. F. Spalding, Determination of Atrazine in Water at Low- and Sub-Parts-Per-Trillion Levels by Using Solid-Phase Extraction and Gas Chromatography/High-Resolution Mass Spectrometry. *Analytical Chemistry* 65, 1993, 21-26.
- [2] T. D. Bucheli, F. C. Grüebler, S. R. Müller, R. P. Schwarzenbach, Simultaneous Determination of Neutral and Acidic Pesticides in Natural Waters at the Low Nanogram per Liter Leve., *Analytical Chemistry* 69, 1997, 1569-1576.
- C. Planas, A. Puig, J. Rivera, J. Caixach, Analysis of pesticides and metabolites in Spanish surface waters by isotope dilution gas chromatography/mass spectrometry with previous automated solid-phase extraction; Estimation of the uncertainty of the analytical results. Journal of. Chromatography A 1131, 2006, 242–252
- [4] D. Barceló, Environmental Protection Agency and Other Methods for the Determination of Priority Pesticides and Their Transformation Products in Water. *Journal of Chromatography A* 643, 1993, 117-143.
- [5] A. Di Corcia, C. Crescenzi, E. Guerriero, R. Saperi, Ultratrace Determination of Atrazine and Its Six Major Degradation Products in Water by Solid-Phase Extraction and Liquid Chromatography-Electrospray/Mass Spectrometry. *Environmental Science & Technology* 31, 1997, 1658-1663.
- [6] R. J. C. A. Steen, A. C. Hogenboom, P. E. G. Leonards, R. A. L. Peerboom, W. P. Cofino, U. A. Th. Brinkman, Ultra-Trace-Level Determination of Polar Pesticides and Their Transformation Products in Surface and Estuarine Water Samples Using Column Liquid Chromatography– Electrospray Tandem Mass Spectrometry. *Journal of Chromatography A* 857, 1999, 157-166.

Compound Benzene						
CAS Number 71-43-2				Wat	er Solubili 1750	
AA-EQS	[µg/L]	2.15		MAC-	EQS [µg/L	
Inland Surface Waters 10		face Waters	Inland Su	urface Waters 50		rface Waters 50
Available Standard Method ISO 11423-1:1997 Determination of benzene and some derivatives – Head-space gas chromatographic method Matrix Water Sampling Pretreatment Storage Method Validation			Method DescriptionA defined volume of unfiltered water sample is heated in a gas-tight septum-covered vial. After establishment of equilibrium between the gaseous and the liquid phases, an aliquot of the gaseous phase is transferred to a gas chromatograph.Limit of Quantification (LOQ): $\sim 2 \mu g/L$ I Number of laboratories n_{AP} percental rate of outliers s_R standard deviation between the laboratories CV_R reproducibility variation coefficient			
Interlaboratory study 1991 Matrix	(Head Spo	ace GC – ISO 1 n _{AP} in %	=	s/L S _R in J	ua/I	CV _R in %
Surface water	8	14.3	x in μ _ξ 5.6	<u>3/L</u> 3, m p 0.8	_	15.7
Interlaboratory study 1991	GC-FID	– ISO 11423 - 2))	I		
Surface water	9	6.5	4.55	1.3	4	29.4
Other Analytical Methods Determination of benzene and some derivatives - Method using extraction and gas chromatography (ISO 11423-2) The unfiltered water sample is extracted with a non-polar solvent (e.g. pentane) and the extract is analysed by GC-MS. LOQ ~ 5 µg/L ISO 15680:2003 Gas-chromatographic determination of a number of monocyclic aromatic hydrocarbons, naphthalene and several chlorinated compounds using purge-and-trap and thermal desorption. (Selected ion: 78) LOQ ~ 2 ng/L for benzene Comments References						

<u>Compound</u> Pentabromodiphenylether					Br Br Br				
CAS Numb	9er 32534-81-9		Log I 6.5			Water Solubility 13.3	[µg/L]		
		S [µg/L]	0.5		MA	AC-EQS [µg/L]			
	face Waters		face Waters 0002		Surface Wate t applicable		ace Waters applicable		
Available S	Standard Met	hod		Method	Description				
 PBDE in water available ISO 22032:2006 Determination of selected polybrominated diphenylethers (PBDE) in sediment and sewage sludge - Method using extraction and gas chromatography/mass spectrometry Matrix Sediment Sampling ISO 5667-13 Pretreatment homogenizing, freezing and freeze-drying, grind and sieve it 					 ISO 22032:2006 specifies a method for the determination of selected PBDEs representative for penta-, octa-, and decaBDE technical formulations in sediment using gas chromatography/mass spectrometry in the electron impact or electron capture ionisation mode. Extraction of PBDEs from the dried sample by an organic solvent is followed by clean-up of the extract by e.g. multi-layer silica gel column chromatography. For quantification an internal standard calibration is applied. Limit of Quantification (LOQ): When applying GC-EI-MS, the method is applicable to samples containing 0.05 				
Storage at 4 bottles	4 °C in the dar	kness in wid	e necked	Approx	µg/kg to 25 µg/kg of tetra- to decabromo congeners. Approximately ten times lower concentrations can be quantified when using GC-ENCI-MS.				
Method Validation / Number of laboratories nAP percental rate of outliers = X Total mean after elimination of outliers sR standard deviation between the laboratories CVR reproducibility variation coefficient Interlaboratory Study 2004/2005									
Matrix	Substance	1	n _{AP} ii	n %	= x in μg/kg	s _R in µg/kg	CV_R in %		
Sediment	BDE 47 BDE 100 BDE 99 BDE 154 BDE 153	16 16 16 16 16	0 0 0		362 93.3 518 39.2 47.7	50.5 28.96 99.6 9.11 9.28	14.0 31.0 19.2 23.2 19.5		
Other Ana			0	I	7/./	9.20	17.3		

U.S. EPA Method 527 employs solid-phase extraction with analysis by gas chromatography/ mass spectrometry as described in PEPICH et al. 2005 [2], but MDL is fairly high (0.39 µg/l).

EPA Method 1614, 2007 [3] applies HRGC/HRMS for the analysis of PBDE in water, soil, sediment and tissue. MDL for BDE 99 is $0.00004 \mu g/l$.

Comments

There are a few reports on extremely low levels of PBDEs in surface water samples [4,5]. The authors enriched 100 and 2500 L of water, respectively, on XAD resin. SPME has been proposed to extract selected BDE congeners from water samples by POLO et al. 2004 [6].

Environmental studies conducted primarily in Europe, Japan and North America indicate that these chemicals are ubiquitous in sediment and biota [7].

- [1] A. Covaci, S. Voorspoels, J. de Boer, Determination of brominated flame retardants, with emphasis on polybrominated diphenyl ethers (PBDEs) in environmental and human samples—a review. *Environment International* 29, 2003, 735-756.
- [2] B. V. Pepich, B. Prakash, M. M. Domino, T. A. Dattilio, Development of U.S. EPA method 527 for the Analysis of Selected Pesticides and Flame Retardants in the UCMR Survey. *Environmental Science & Technology* 39, 2005, 4996-5004.
- [3] http://www.accustandard.com/asi/pdfs/epa_methods/1614.pdf.
- [4] F. Luckey, B. Fowler, S. Litten, Establishing Baseline Levels of Polybrominated Diphenyl Ethers in Lake Ontario Surface Waters. The Second International Workshop on Brominated Flame Retardants, Stockholm University, Sweden, May 14-16, 2001, 337-339.
- [5] D. R. Oros, D. Hoover, F. Rodigari, D. Crane, J. Sericano, Levels and Distribution of Polybrominated Diphenyl Ethers in Water, Surface Sediments, and Bivalves from the San Francisco Estuary. *Environmental Science & Technoogy* 39, 2005, 33-41.
- [6] M. Polo, G. Gómez-Noya, J.B. Quintana, M. Llompart, C. García-Jares, R. Cela, Development of a solid-phase microextraction gas chromatography/tandem mass spectrometry method for polybrominated diphenyl ethers and polybrominated biphenyls in water samples. *Analytical Chemistry* 76, 2004, 1054-1062.
- [7] R. J. Law, C. R. Allchin, J. de Boer, A. Covaci, D. Herzke, P. Lepom, S. Morris, J. Tronczynski, C. A. de Wit, Levels and trends of brominated flame retardants in the European environment. *Chemosphere* 64(2), 2006, 187-208.

CAS Number	Log	K _D [L/kg]	Water S	olubility [mg/L]		
7440-43-9	suspended matter sediment/water: 3	<i>r/water:</i> 4.7 (Cd) [1 3.6 (Cd) [1]] dependi	ng on compound		
Water hardness [mg CaCO ₃ /L]	AA-EQS (depending on wate	S [μg/L] er hardness classes)		EQS [µg/L] ater hardness classes)		
	Inland Surface Waters	Other Surface Waters	Inland Surface Waters	Other Surface Waters		
Class 1: < 40 Class 2: $40 \text{ to} < 50$ Class 3: $50 \text{ to} < 100$ Class 4: $100 \text{ to} < 200$	≤ 0.08 0.08 0.09 0.15	0.2	≤ 0.45 0.45 0.6 0.9	≤ 0.45 0.45 0.6 0.9		
Class 5: \geq 200 Available Standard Method EN ISO 17294-2:2004	0.25	1.5 1.5 Method Description 1.5				
Application of inductively cou spectrometry (ICP-MS) - Part 62 elements <u>Matrix</u> Drinking waters, groun waters and waste waters <u>Sampling</u> ISO 5667-1, 5667-2 <u>Pretreatment</u> For dissolved el aqueous sample through a 0.45 filter. Adjust the pH of the filt HNO ₃ . <u>Storage</u>	2: Determination of nd waters, surface <i>and 5667-3</i> lements, filter 5-µm pore membrane	1.51.5Method DescriptionEN ISO 17294-2:2004 specifies a method for the determination of the cadmium in water (for example drinking water, surface water, groundwater, wastewater and eluates). Taking into account the specific and additionally occurring interferences, these elements can also be determined in digests of water, sludges and sediments.The detection limits of most elements are affected by blank contamination and depend predominantly on the laboratory air-handling facilities available.The lower limit of application is higher in cases where the determination is likely to suffer from interferences or in case of memory effects.Limit of Quantification (LOQ): Drinking water and relatively unpolluted water: 0.1 µg/L - 1.0 µg/L				
<u>Method Validation</u>			=	<i>l</i> Number of laborator n_{AP} percental rate of outli after elimination of outli		

Matrix	l	n _{AP} in %	$=$ x in $\mu g/L$	s _R in µg/L	CV _R in %
Surface water	37	5.2	5.75	0.491	8.5

Other Analytical Methods

EN ISO 5961:1995 specifies two methods for the determination of cadmium using atomic absorption spectrometry (AAS) in an air-acetylene flame (aspiration of the acidified sample into the flame and measurement of the cadmium concentration at a wavelength of 228.8 nm) and by electrothermal atomization AAS (injection of the acidified sample into an electrically heated graphite tube of an electrothermal atomization atomic absorption spectrometer and measurement of the absorbance at a wavelength of 228.8 nm).

EN ISO 15586:2003 determination using atomic absorption spectrometry with electrothermal atomization in a graphite furnace. The detection limit of the method for each element depends on the sample matrix as well as of the instrument, the type of atomizer and the use of chemical modifiers. For water samples with a simple matrix (i.e. low concentration of dissolved solids and particles), the method detection limits will be close to instrument detection limits. The minimum acceptable LOQ for a 20 μ L sample volume are specified.

EN ISO 11885:1997 specifies a method by inductively coupled plasma atomic emission spectroscopy.

EPA 200.8 (1994): Determination of trace elements in waters by inductively coupled plasma - mass spectrometry (LOQ: 0.5 µg/L); http://www.accustandard.com/asi/pdfs/epa_methods/200_8.pdf

Standard Methods Online (http://standardmethods.org/) 3125: Metals in Water by ICP/MS (LOQ: 0.003 µg/L)

Comments

$\frac{\text{Compound}}{C_{10-13}\text{-}Chlc}$	$C_xH(_{2x-y+2})Cl_y$ where x = 10-13 AND y = 1-13					
CAS Number	Log K _{OW}	Water Solubility [mg/L]				
85535-84-8	4.39-8.69 (depending on chloring	e content)	0.15-0.4	7 (59% chlorine content)		
AA-EQ			MAC-	EQS [µg/L]		
Inland Surface Waters 0.4	Other Surface Waters 0.4	Inland St	urface Waters	Other Surface Waters 1.4		
Available Standard Meth	nod	Method 1	Description			
ISO CD 12010 Error! Reference source Matrix Sampling Pretreatment Storage	not found.	Method Description Determination of the sum of SCCP in technical mixtures with chlorine contents 49 % to 67 % independent of the chlorine content and independent of the C-number distribution pattern of the congeners. No recognition of the chlorine content is necessary. Extraction of the whole water sample by liquid-liquid-extraction by an organic solvent, alternatively by solid phase extraction. After concentration and clean up, capillary gas chromatography of the approximately 6300 congeners at a relatively short column within a short retention range. Detection of selected mass fragments by mass spectrometry in the selected ion monitoring mode using negative ion chemical ionisation (NCI). The selection of the mass fragments is specific for the variety of technical mixtures as well as for the chlorine content and C-number distribution patterns in environmental samples. Alternative selections of mass fragment combinations for quantification are given in this standard too. The chromatogram is to be integrated over the full retention range of the SCCP. The quantification of the sum of SCCP is performed after calibrating by a multiple linear regression model with solutions of different technical mixtures. This reflects the fact that the variety in respect of chlorine content and C-number distribution of technical SCCP-mixtures as well as of SCCP in environmental samples defined standard. The selection of the mass fragments for quantification and the special calibration allow a quantification of the sum of SCCP independent of the sum of SCCP independent of the sum of scCCP independent of the sum of scCCP independent of the sum of scCCP independent of the sum of scCP independ				
Method Validation						
	no da	ıta available	;			
Other Analytical Method	<u>ls</u>					

GC-ECNI-HRMS in the SIM mode at an ion source temperature of 120°C [1]. The molecular compositions of commercial SCCPs and of SCCP-containing extracts were determined by monitoring the two most intensive ions in the [M-Cl]- cluster, one for quantification and the other for confirmation for the following formula groups:

C10 (Cl5 to Cl10), C11 (Cl5 to Cl10), C12 (Cl6 to Cl10), and C13 (Cl7 to Cl9), and assuming that integrated signals are proportional to molar concentrations weighted by the number of chlorine atoms in the formula group. Quantification was achieved by selecting the biggest peak corresponding to [M-Cl]- ion in the most abundant formula group present in the sample and correcting for variations in the formula group abundances between standard and sample. The analytical detection limit was 60 pg of injected SCCP at a signal-to-noise ratio of 4:1, while LOQ was 23 ng/g.

Short-column (62 cm) GC-ECNI- LRMS at an ion source temperature of 100°C using methane as reagent gas [2]. Detection limits in the full- scan mode ranged from 10 to100 pg depending on carbon chain length of the n-alkane and on the degree of chlorination. The method was applied to the analysis of SCCP in fish samples.

Metastable atom bombardment ionisation (MAB) and high resolution mass spectrometry [3]. The detection limits were estimated to be between 10 and 100 pg/L. The MAB method has been applied to the analysis of high-volume water samples.

GC-MS/MS electron ionisation (EI) for fast determination of the sum of short medium chain chlorinated paraffins [4]. Collision-induced reactions of m/z 91 \rightarrow 53 (LOQ = 0.15 ng/µL), 102 \rightarrow 65 (LOQ = 0.2 ng/µL), and 102 \rightarrow 67 (LOQ = 0.1 ng/µL) were used to quantify the total short- and medium-chain PCA content of pooled fish liver samples.

Quantification procedure using GC-ECNI-MS, which is independent of the chlorine content of the reference standard used for calibration [5]. The authors calculated the total response factors for seven standard CP mixtures of various chlorine contents (51-70%) from the relative total CP areas and found a linear correlation between the total response factors of CP mixtures and their chlorine contents (R2=0.9494). Using this correlation, total response factors according to the chlorine content of the SCCPs present in the sample can be calculated and used for quantification.

SPE and carbon skeleton analysis after simultaneous catalytic dechlorination and hydrogenation by gas chromatography with mass spectrometric detection seems to be promising option for routine analysis of of SCCPs in water even though the method has not yet been fully validated [6,7]

Comments

Although some work has been conducted on development of selective and sensitive methods for SCCP analysis in recent years, for the time being, no fully validated procedure is available that could be recommended for routine monitoring of SCCPs in environmental samples.

SCCP concentrations in environmental samples analysed by GC-ECNI-MS can vary widely (by a factor of ten) depending on chlorine content of the standard used for quantification [8].

- [1] G. T. Tomy, G. A. Stern, D. C. G. Muir, A. T. Fisk, C. D. Cymbalisty, J. B. Westmore, Quantifying C10-C13 polychloroalkanes in environmental samples by high-resolution gas chromatography/electron capture negative ion high-resolution mass spectrometry. *Analytical Chemistry* 69, 1997, 2762-2771.
- [2] M. Coelhan, Determination of short-chain polychlorinated paraffins in Fish Samples by short-column GC/ECNI-MS. *Analytical Chemistry* 71, 1999, 4498-4505.
- [3] S. Moore, L. Vromet, B. Rondeau, Comparison of metastable atom bombardment and electron capture negative ionization for the analysis of polychloroalkanes. *Chemosphere* 54, 2004, 453-459.
- [4] Z. Zencak, M. Reth, M. Oehme, Determination of Total Polychlorinated n-Alkane Concentration in Biota by Electron Ionization-MS/MS. *Analytical Chemistry* 76, 2004, 1957-1962.
- [5] M. Reth, Z. Zencak, M. Oehme, New quantification procedure for the analysis of chlorinated paraffins using electron capture negative ionization mass spectrometry. *Journal of Chromatography A* 1081, 2005, 225–231.
- [6] F. Pellizzato, M. Ricci, A. Held, H. Emons. Analysis of short-chain chlorinated paraffins: a discussion paper. *International Journal of Environmental Analytical Chemistry* 9, 2007, 924-930.
- [7] F. Pellizzato, M. Ricci, A. Held, H. Emons. Determination of short-chain chlorinated paraffins by carbon skeleton gas chromatography. Organohalogen Componds 70, 2008, 776-778.

[8] M. Coelhan, M. Saraci, H. Parlar, A comparative study of polychlorinated alkanes as standards for the determination of C10-C13 polychlorinated paraffins in fish samples. *Chemosphere* 40, 2000, 685-689.

Compound Chlorfenvinphos			H ₃ CH ₂ CO H ₃ CH ₂ CO CI			
CAS Number 470-90-6	Log K ~ 3.8		Wat	ter Solubility [mg/L] ~ 145 [1]		
AA-EQS [µg/L]			MAC-	EQS [µg/L]		
Inland Surface Waters Other Su	rface Waters 0.1	Inland Su	urface Waters 0.3	Other Surface Waters 0.3		
Available Standard Method EN ISO 10695:2000 Determination of selected organic nitr phosphorus compounds - Gas chroma methods Note: Chlorfenvinphos is not explicit this standard but the method may also the analysis of chlorfenvinphos provid has been properly validated for this c <u>Matrix</u> Drinking waters, ground wate waters and waste waters containing up of suspended solids <u>Sampling</u> <u>Pretreatment</u> <u>Storage</u>	tography y mentioned in b be applied to ded the method ompound. ers, surface	0.3 0.3 Method Description Liquid/liquid extraction with dichloromethane or liquid/solid extraction (SPE) on reversed-phase (RP)-C18 material or other adsorbent. Elution of the cartridges with e.g. methanol or acetone. After concentration, the sample extracts are analysed by gas chromatography, using a nitrogen-phosphorus or mass spectrometric detector.				
Method Validation	no da	ta available	:			
no data available Other Analytical Methods Solid-phase microextraction (SPME) SPME in 4 mL glass vials with a 60 µm PDMS-DVB coated fibre at 60°C [2]. Gas Chromatography - Mass Spectrometry GC-MS determination of the ions 267, 323, 295 [2, 4] LOQ ~ 25 ng/L [2] LOQ ~ 25 ng/L [2] GC tandem MS-MS Parent ion m/z 267; product ions m/z 159 and 203 [2] LOQ ~ 25 ng/L [2] Solid-phase extraction (SPE) – HPLC/UV LOQ ~ 25 ng/L [5] GC - Flame Photometric Detection (FPD) EPA method 1657; LOQ ~ 2 ng/L (solvent extraction) [1, 6]						
Comments						

Existence of E and Z double bond isomers; the Z-isomer has a water solubility of 121 mg/L and the E-isomer of 7.3 mg/L (at 20°C); the mixture 145 mg/L at 23°C; log $K_{OW} \sim 3.85$ (Z-isomer) and 4.22 (E-isomer).

- [1] US Department of Health and Human Services, Agency for Toxic Substances and Disease Registry (ATSDR), Toxicological Profile Information Sheet, URL: http://www.atsdr.cdc.gov/toxprofiles/tp83.html.
- [2] C. Gonçalves, M. F. Alpendurada, Solid-phase micro-extraction–gas chromatography–(tandem) mass spectrometry as a tool for pesticide residue analysis in water samples at high sensitivity and selectivity with confirmation capabilities. *Journal of Chromatography A* 1026, 2004, 239-250.
- [3] C. Goncalves, M. F. Alpendurada, Assessment of pesticide contamination in soil samples from an intensive horticulture area, using ultrasonic extraction and gas chromatography–mass spectrometry. Talanta 65, 2005, 1179-1189.
- [4] C. Planas, A. Puig, J. Rivera, J. Caixach, Analysis of pesticides and metabolites in Spanish surface waters by isotope dilution gas chromatography/mass spectrometry with previous automated solid-phase extraction; Estimation of the uncertainty of the analytical results. *Journal of Chromatography A* 1131, 2006, 242–252.
- [5]] C. Schlett, Multi-residue analysis of pesticides by HPLC after solid-phase extraction. *Fresenius' Journal of Analytical Chemistry*. 339, 1991, 344-347.
- [[6] F.J. Lopez, J. Beltran, M. Forcada, F. Hernandez, Comparison of simplified methods for pesticide residue analysis; Use of large-volume injection in capillary gas chromatography. *Journal of Chromatography A* 823, 1998, 25-33.

Compound Chlorpyrifos			$H_{3}CH_{2}CO \xrightarrow{S} \begin{array}{c} CHCI \\ H_{3}CH_{2}CO \end{array} \xrightarrow{P - O - C} \begin{array}{c} CI \\ CI \\ CI \end{array} \xrightarrow{CI} CI$				
CAS Number 2921-88-2		Log K ~ 4.9		Wat	ter Solubility [mg/L] ~ 0.762		
	S [µg/L]	~ 4.9	6	MAC-	~ 0.762 EQS [μg/L]		
Inland Surface Waters		face Waters	Inland St	Inface Waters	Other Surface Waters		
0.03		0.03	inianu St	0.1	0.1		
Available Standard Meth EN ISO 10695:2000 Determination of selected phosphorus compounds - of methods Note: Chlorpyriphos is not this standard but the meth the analysis of chlorpyriph has been properly validated Matrix Drinking waters, g waters and waste waters c of suspended solids Sampling Pretreatment Storage	tography nentioned in be applied to l the method ompound. rs, surface	0.1 0.1 Method Description Liquid/liquid extraction with dichloromethane or liquid/solid extraction (SPE) on reversed-phase (RP)-C18 material or other adsorbent. Elution of the cartridges with e.g. methanol or acetone. After concentration, the sample extracts are analysed by gas chromatography, using a nitrogen-phosphorus or mass spectrometric detector. Limit of Quantification (LOQ): 0.01 μg/L					
Method Validation		no da	ta available				
Other Analytical Metho	ds						
SPME in 4 mL glass vials Solid-phase extraction (SI	Solid-phase microextraction (SPME) SPME in 4 mL glass vials with a 60 μm PDMS-DVB coated fibre at 60°C [2] Solid-phase extraction (SPE) SPE with C18 cartridges; elution with ethylacetate [3]						
GC-MS determination of LOQ \sim 1-2 ng/L [2, 5] GC tandem MS-MS	the ions 199	, 197, 314, 316	[1-5]				
Parent ion m/z 314; product ions m/z 286 and 258 [2] LOQ ~ 1 ng/L [2] <u>GC-NPD</u> :							
LOQ ~ 20 ng/L [3] GC - Flame Photometric I	Detection (F	<u>PD)</u> [7]					
<u>Comments</u> Chlorpyrifos is a non-pola	r insecticide	e. If released to	water, chlor	pyrifos partition	s significantly from the water		

column to sediments.

- [1] M. Kirchner, E. Matisová, S. Hrouzková, J. de Zeeuw, Possibilities and limitations of quadrupole mass spectrometric detector in fast gas chromatography. *Journal of Chromatography A* 1090, 2005, 126-132.
- [2] C. Gonçalves, M.F. Alpendurada, Solid-phase micro-extraction–gas chromatography–(tandem) mass spectrometry as a tool for pesticide residue analysis in water samples at high sensitivity and selectivity with confirmation capabilities. *Journal of Chromatography A* 1026, 2004, 239-250.
- [3] S. Lacorte, C. Molina, D. Barceló, Screening of organophosphorus pesticides in environmental matrices by various gas chromatographic techniques. *Analytica Chimica Acta* 281, 1993, 71-84.
- [4] K. Kawata, T. Asada, K. Oikawa, Determination of pesticides in compost by pressurized liquid extraction and gas chromatography–mass spectrometry. *Journal of Chromatography A* 1090, 2005, 10-
- 15.
- C. Planas, A. Puig, J. Rivera, J. Caixach, Analysis of pesticides and metabolites in Spanish surface waters by isotope dilution gas chromatography/mass spectrometry with previous automated solid-phase extraction; Estimation of the uncertainty of the analytical results. Journal of Chromatography A, 1131, 2006, 242–252
- [6] US Department of Health and Human Services, Agency for Toxic Substances and Disease Registry (ATSDR), Toxicological Profile Information Sheet, URL: <u>http://www.atsdr.cdc.gov/toxprofiles/tp84.html</u>
- [7] F. J. Lopez, J. Beltran, M. Forcada, F. Hernandez, Comparison of simplified methods for pesticide residue analysis; Use of large-volume injection in capillary gas chromatography. *Journal of Chromatogaphy. A* 823, 1998, 25-33.

Compound										
		ci				CI				
Aldrin		Di	ieldrin		Endrin			Isodrin		
		CAS Num	ber		Log K _{OW}		Water S	Solubility [mg/L]		
Aldrin		309-00-	2		~ 6.50			~ 0.011		
Dieldrin		60-57-1			~ 6.2			~ 0.110		
Endrin		72-20-8			~ 5.6			~ 0.20		
Isodrin		465-73-			~ 6.75			~ 0.014		
15041111		102 75	Ŭ		0.70			0.011		
	-	S [µg/L]					EQS [µ	-		
Inland Surface Wa	ters	Other Sur	rface Waters	5	Inland Surface		Other	Surface Waters		
$\Sigma = 0.01$		$\Sigma =$	= 0.005		not applicat	ble		not applicable		
Available Standard	l Metl	hod			Method Descrip	tion				
<i>EN ISO 6468:1996</i> Determination of certain organochlorine insecticides, polychlorinated biphenyls and chlorobenzenes – Gas chromatographic method after liquid-liquid extraction <u>Matrix</u> Drinking, ground, surface and waste waters				8	Liquid-liquid extraction of organochlorine insecticides, chlorobenzenes and PCBs by an extraction solvent. After concentration and clean-up the sample extracts are analysed by gas chromatography, using an electron- capture detector (GC-ECD).					
<u>Sampling</u> <u>Pretreatment</u>					The method is applicable to samples containing up to 50 mg/L of suspended solids.					
								0.004 0.01 5		
<u>Storage</u>					Limit of Quanti	fication (<u>(LOQ):</u>	~ 0.001 – 0.01 µg/L		
Method Validation								<i>l</i> Number of laboratories n_{AP} percental rate of outliers		
						s _R star	ndard devia	n after elimination of outliers ation between the laboratories lucibility variation coefficient		
Interlaboratory stud	ły (Ex	traction of s	surface water	r w	ith Hexane)					
Substance		l	n _{AP} in %		= x in ng/L	s _R in J	ug/L	CV_R in %		
Dieldrin		14	0		33.3	17	.2	51.7		
Endrin		14	9.8		50.0	11	.1	22.3		

Other Analytical Methods

<u>Solid-phase extraction gas chromatography - mass spectrometry</u> SPE with Oasis HLB cartridges; elution with dichloromethane. GC-MS determination of the ions 66 for aldrin, 79 for dieldrin, 281 for endrin, and 193 for isodrine [1,2].

LOQ ~ 20 ng/L for aldrin, 10 ng/L for dieldrin, 15 ng/L for endrin, and 12 ng/L for isodrin

(SPE extraction of 200 mL water) [1]

SPME GC-MS

SPME in 4 mL glass vials with a 60 μ m PDMS-DVB coated fibre at 60°C; LOQ ~ 12 ng/L for aldrin, 9 ng/L for dieldrin, 60 ng/L for endrin, and 10 ng/L for isodrin [2]

<u>SPE-GC- triple quadrupole-MS-MS</u> C18-SPE, 100 mL, SRM 263 > 193 (dieldrin), 261 > 191 (aldrin), 193 > 157 (isodrin); LOQ ~ 25 ng/L [3]

<u>SPE-GC-NCI-MS</u> C18-SPE, 100 mL, LOQ ~ 25 ng/L [3]

Comments

- S. Lacorte, I. Guiffard, D. Fraisse, D. Barceló, Broad Spectrum Analysis of 109 Priority Compounds Listed in the 76/464/CEE Council Directive Using Solid-Phase Extraction and GC/EI/MS. *Analytical Chemistry* 72, 2000, 1430-1440.
- [2] C. Gonçalves, M.F. Alpendurada, Solid-phase micro-extraction–gas chromatography–(tandem) mass spectrometry as a tool for pesticide residue analysis in water samples at high sensitivity and selectivity with confirmation capabilities. *Journal of Chromatography A* 1026, 2004, 239-250.
- [3] E. Pitarch, C. Medina, T. Portolés, F.J. López, F. Hernández, Determination of priority organic micropollutants in water by gas chromatography coupled to triple quadrupole mass spectrometry. Analytica Chimica Acta 583, 2007, 246–258.

Compound	1						
DDT para-pa							
CAS Number	Log K	ow	Wat	er Solub	ility [mg/L]		
para-para- <i>D</i> DT 50-29-3 <i>o,p p,p</i>		<i>o,p</i> '-DDT <i>p,p</i> '-DDE	<i>p,p</i> '-DDT ~ 6.91 <i>o,p</i> '-DDT ~ 6.79 <i>p,p</i> '-DDE ~ 6.51 <i>p,p</i> '-DDD ~ 6.02		p,p'-DDT ~ 0.025 o,p'-DDT ~ 0.085 p,p'-DDE ~ 0.12 p,p'-DDD ~ 0.090		
AA-EQS [µg/L]				MAC-	EQS [µg	/L]	
Inland Surface Waters	Other Sur	face Waters	Inland St	rface Waters	Other S	Surface Waters	
DDT total 0.025 para-para-DDT 0.01	DDT total para-para-		not d	pplicable		not applicable	
Available Standard Meth	Method I	Description					
<i>EN ISO 6468:1996</i> Determination of certain o insecticides, polychlorinatic chlorobenzenes – Gas chro after liquid-liquid extraction <u>Matrix</u> Drinking, ground, <u>Sampling</u> <u>Pretreatment</u>	s and c method	Liquid-liquid extraction of organochlorine insecticides, chlorobenzenes and PCBs by an extraction solvent. After concentration and clean-up the sample extracts are analysed by gas chromatography, using an electron- capture detector (GC-ECD). The method is applicable to samples containing up to 50 mg/L of suspended solids.					
<u>Storage</u>			Limit of (Quantification (LOQ): ~	ν 0.001 – 0.01 μg/L	
<u>Method Validation</u> Interlaboratory study (Ex.	traction of f	p,p'-DDT with I	Hexane)	s _R stan	Total mear dard deviat	<i>l</i> Number of laboratories n_{AP} percental rate of outliers a after elimination of outliers ion between the laboratories icibility variation coefficient	
Matrix	l	n _{AP} in %	= x in n;	g/L s _R in J	ug/L	CV _R in %	
Drinking water	11	10.3	35.7	·	.9	64.1	
Other Analytical Method	ls						
Solid-phase extraction - gas SPE with Oasis HLB cartr GC-MS determination of t The second qualitative ion $LOQ \sim 4 \text{ ng/L for } p,p'-D$ 11 ng/L for p,p'-D 12 ng/L for p,p'-I 12 ng/L for p,p'-I	idges; elution he ions 235 s are 165 fo DT, DDT, DE, and	n with dichloro for DDT and D r DDT and DDI	methane. DD, and 24 D, and 176	6 for DDE.	,2]		
SPME in 4 mL glass vials DDD and 1 ng/L for DDE		m PDMS-DVB	coated fibro	e at 60°C; LOQ ~	~ 12 ng/L	for DDT, 2 ng/L for	

GC-ECD

Bettinetti et al. detected 0.05 and 0.16 ng/L of dissolved pp'DDT and pp'DDE in the liquid water fraction of Lake Maggiore, Italy [4].

EPA methods 508 (GC-ECD) and 625 (GC-MS): Liquid-liquid extraction of 1 L water with dichloromethane.

Comments

Technical grade DDT consists of 65-80 % of *p*,*p*'-DDT, 15-21 % of *o*,*p*'-DDT, up to 4 % of *p*,*p*'-DDD. *p*,*p*'-DDE is a metabolite of DDT.

DDT is very persistent in the environment, with a reported half-life between 2-25 years; it has a low solubility in water.

- [1] S. Lacorte, I. Guiffard, D. Fraisse, D. Barceló, Broad Spectrum Analysis of 109 Priority Compounds Listed in the 76/464/CEE Council Directive Using Solid-Phase Extraction and GC/EI/MS. *Analytical Chemistry*, 2000, 1430-1440.
- [2] D. de Almeida Azevedo, S. Lacorte, T. Vinhas, P. Viana, D. Barceló, Monitoring of Priority Pesticides and Other Organic Pollutants in River Water From Portugal by Gas Chromatography Mass Spectrometry and Liquid Chromatography–Atmospheric Pressure Chemical Ionization Mass Spectrometry. *Journal of Chromatography A* 879, 2000, 13-26.
- [3] C. Gonçalves, M.F. Alpendurada, Solid-phase micro-extraction–gas chromatography–(tandem) mass spectrometry as a tool for pesticide residue analysis in water samples at high sensitivity and selectivity with confirmation capabilities. *Journal of Chromatography A* 1026, 2004, 239-250.
- [4] R. Bettinetti, V. Croce, S. Galassi, P. Volta, Pp'DDT and pp'DDE accumulation in a food chain of Lake Maggiore (Northern Italy): testing steady-state condition, *Environmental. Scence and Pollution Research International* 13, 2006, 59-66.

<u>Compound</u> 1,2-Dichlor	oethane			cı—CI				
CAS Number 107-06-2		Log K 1.45			ter Solubility [g/L] 8.5-9.0			
AA-EQS	[µg/L]			MAC-	EQS [µg/L]			
Inland Surface Waters 10	Other Sur	face Waters 10		urface Waters applicable	Other Surface Waters not applicable			
Available Standard Metho EN ISO 15680: 2003 Gas-chromatographic deterr monocyclic aromatic hydrod and several chlorinated com and-trap and thermal desorp <u>Matrix</u> drinking water, grou seawater and (diluted) waste <u>Sampling</u> ISO 5667-1, 5667 <u>Pretreatment</u> <u>Storage</u> hermetically sealed sunlight, analysis within 5 d <u>Method Validation</u>	nination of carbons, na pounds usi tion und water, e water 7-2 and ISC at 4 °C, av	nphthalene ing purge- surface water, 2 <i>5667-3</i> void direct	EN ISO 1 determina water by chromato Detection in the ele be applied	ation of volatile of purge-and-trap is graphy (GC). It is preferably can otron impact moo d as well. Quantification (fies a general method for the organic compounds (VOCs) in solation and gas rried out by mass spectrometry le (EI), but other detectors may LOQ): 0.01 µg/L			
	es two met aphy with	e.g. electron caj			le halogenated hydrocarbons in xtraction by an organic solvent			
The EPA Method 1624 is de trap gaschromatography-ma Hubrechts et al. 2003 give a compounds in estuarine wat <u>Purge and trap GC-MS</u> SIM-GC-MS detection of th (Modification of EPA metho http://www.accustandard.co	ss spectron review of ers with sp the ions 62, od 524.2:V	netry. gas chromatog ecial emphasis 98, 64; LOQ ~ 'OCs in Water I	raphy-based on monitor 2 ng/L [2] Using GC-1	d methods for an ring [1]. MS,	water amenable to purge and alysis of volatile organic			
<u>Comments</u>								
	opogenic	volatile organic	compound	s in estuarine wa	aromatography-based methods ters, illustrated with the river 197.			
	er by auton	nated purge and	l trap coupl		t analysis of volatile organic tography–mass spectrometry.			

Compound Dichloror		ci ci—					
CAS Number 75-09-2		Log K ~ 1.3					
AA-EQS [µg/L]				MAC	-EQS [µ	g/L]	
Inland Surface Waters 20	Other Su	r face Waters 20		rface Waters	Other	Surface Waters not applicable	
Available Standard Mether EN ISO 15680:2003 Gas-chromatographic deter monocyclic aromatic hydro and several chlorinated com and-trap and thermal desorp Matrix drinking water, gro seawater and (diluted) wast Sampling ISO 5667-1, 566 Pretreatment Storage at 4 °C air tight an analysis within 5 days	aphthalene ing purge- surface water, <i>O 5667-3</i>	Method Description EN ISO 15680:2003 specifies a general method for the determination of volatile organic compounds (VOCs) in water by purge-and-trap isolation and gas chromatography (GC). Detection is preferably carried out by mass spectrometry in the electron impact mode (EI), but other detectors may be applied as well. Limit of Quantification (LOQ): 0.01 µg/L					
Method Validation				s _R sta	ndard devia	<i>l</i> Number of laboratories n_{AP} percental rate of outliers an after elimination of outliers ation between the laboratories ducibility variation coefficient	
Interlaboratory study						n.a. = not available	
Matrix	l	n _{AP} in %	= x in µş	g/L s _R in	µg/L	CV_R in %	
Drinking water (0.2 µg/L)	11	n.a.	n.a.	n.	a.	55	
Surface water (0.2 μ g/L)	10	n.a.	n.a. n.a. 68				
Other Analytical Methods	<u> </u>	1	<u> </u>	I		I	

EN ISO 10301:1997 specifies two methods for the determination of highly volatile halogenated hydrocarbons in water using gas chromatography with e.g. electron capture detector after: a) the extraction by an organic solvent or using, b) a head-space method (LOQ: $100 \mu g/L$).

The EPA Method 1624 is designed to determine the volatile organic pollutants in water amenable to purge and trap gaschromatography-mass spectrometry.

HUYBRECHTS et al. 2003 give a review of gas chromatography-based methods for analysis of volatile organic compounds in estuarine waters with special emphasis on monitoring [1].

<u>Purge and trap GC-MS</u> SIM-GC-MS detection of the ions 84, 86, 49; LOQ ~ 62 ng/L [2] (Modification of EPA method 524.2 "VOCs in Water Using GCMS")

Comments

- [1] T. Huybrechts, J. Dewulf, H. Van Langenhove, State-of-the-art of gas chromatography-based methods or analysis of anthropogenic volatile organic compounds in estuarine waters, illustrated with the river Scheldt as an example. *Journal of Chromatography A* 1000, 2003, 283-297.
- [2] E. Martínez, S. Lacorte, I. Llobet, P. Viana, D. Barceló, Multicomponent analysis of volatile organic compounds in water by automated purge and trap coupled to gas chromatography–mass spectrometry. *Journal Chromatography A* 959, 2002, 181-190.

Compound Di(2-ethylhexyl)phthalate (DEHP) CAS Number Log Kategorian							
CAS Number 117-81-7			ow	Wat	ter Solul	bility [μg/L] ³	
AA-EQS	S [µg/L]	I		MAC-	EQS [µį	g/L]	
Inland Surface Waters 1.3	Other Su	rface Waters 1.3		urface Waters applicable	Other	Surface Waters not applicable	
Available Standard Method EN ISO 18856: 2005 Determination of selected phthalates using gas chromatography/mass spectrometry Matrix ground water, surface water, wastewater and drinking water Sampling ISO 5667-1, 5667-2 and 5667-3 Pretreatment			Method Description <i>EN ISO 18856:2005</i> specifies a method for the determination of phthalates in water after solid phase extraction and gas chromatography/mass spectrometry.				
Storage at 4 °C in the darkness, analysis within 4 days			Limit of Quantification (LOQ): 0.02 μg/L - 0.150 μg/L depending on the blank				
$=$ <i>X</i> Total mean s_{R} standard deviati					<i>l</i> Number of laboratories n_{AP} percental rate of outliers n after elimination of outliers ation between the laboratories ucibility variation coefficient		
Matrix	l	n _{AP} in %	x in n	g/L s _R in 1	ng/L	CV_R in %	
Surface water	7	0	373	25	7	69	
Other Analytical Methods SPME-GC-MS GC-MS determination of the ion 149; LOQ ~ 15 to 30 ng/L [1,2] Comments The analysis is difficult due to the omnipresence of phthalates. Specific care shall be taken to minimise blanks. The use of plastic material shall be avoided unconditionally.							
384.	ction of pht Giroud, G. quantificat	halate esters from Dessalces, M. F ion of 33 priorit	m water. <i>Jo</i> . Grenier-L y substance	urnal of Chroma oustalot, C. Crer	<i>utograph</i> n-Olivé. I	y A 922, 2001, 377- Multiresidue analytical	

Compound Diuron			СН ₃					
CAS Number 330-54-1		Log K ~ 2.7						
	AA-EQS [µg/L]			MAC-	~ 4 EQS [µg			
Inland Surface Waters		face Waters	Inland St	Irface Waters		Surface Waters		
0.2		0.2		1.8		1.8		
Available Standard Meth EN ISO 11369:1997	od		Method I	Description				
Determination of selected performance water by high performance with UV detection after so <u>Matrix</u> Drinking and grou	The plant treatment substances in the water sample are extracted by solid-liquid extraction (SPE) on reversed- phase (RP)-C18 material, eluted with a solvent, and then separated, identified and quantified by high performance liquid chromatography (HPLC) using UV detection.							
Sampling								
Pretreatment								
<u>Storage</u>			Limit of	Quantification (LOQ): 0	.1 μg/L		
Method Validation								
$\begin{bmatrix} l & \text{Number of laboratories} \\ n_{AP} & \text{percental rate of outliers} \\ = \\ & \mathcal{X} & \text{Total mean after elimination of outliers} \\ s_{R} & \text{standard deviation between the laboratories} \\ & \text{CV}_{R} & \text{reproducibility variation coefficient} \\ \end{bmatrix}$ Interlaboratory trial								
Matrix	l	n _{AP} in %	= x in µ	g/L s _R in p	ug/L	CV_R in %		
Drinking water	33	0	0.102	6 0.02	.99	29.1		
Ground water	32	5.1	0.281	5 0.05	70	20.2		
Other Analytical MethodsLiquid Chromatography - Mass SpectrometryIdentification and quantification of diuron (and other pesticides) by liquid chromatography coupled to (tandem)mass spectrometric detection (LC-MS-MS) using positive electrospray ionization (ESI)LC-MS fragment ions: m/z 233 [1,2]LC-MS-MS transitions: 233 > 72 and 46 [3,4,5]LOQ ~ 1 ng/L (depending on the SPE enrichment factor)Liquid Chromatography – Diode Array DetectorSea water analysis: Off-line SPE – LC-DAD; LOQ ~ 0.01 µg/L [6]Gas Chromatography - Mass Spectrometry (after derivatisation)Phenylurea herbicides require a derivatisation step to prevent the degradation of these thermolabile compoundsin the GC injector; LOQ ~ 1 ng/L [7]SPE-derivatisation-GC-MS; LOQ ~ 50 ng/L [8]								

Comments

The EN ISO 11369 HPLC-UV method is only applicable for drinking and ground water, not to the analysis of contaminated surface water. GC-MS determination of phenylurea herbicides is difficult due to the necessary derivatisation step. LC-MS-MS seems to be the method of choice.

- [1] C. Crescenzi, A. Di Corcia, R. Samperi, N. L. Dietz, E. Guerriero, Development of a Multiresidue Method for Analyzing Pesticide Traces in Water Based on Solid-Phase Extraction and Electrospray Liquid Chromatography Mass Spectrometry. Environmental Science & Technology 31, 1997, 479-488.
- [2] R. Loos, G. Hanke, S. J. Eisenreich: Multi-Component Analysis of Polar Water Pollutants Using Sequential Solid-Phase Extraction Followed by LC-ESI-MS. Journal of Environmental Monitoring 5, 2003, 384-394.
- [3] M. H. Lamoree, C. P. Swart, A. van der Horst, B. van Hattum, Determination of Diuron and the Antifouling Paint Biocide Irgarol 1051 in Dutch Marinas and Coastal Waters. Journal of Chromatography A 970, 2002, 183-190.
- [4] R. J. C. A. Steen, A. C. Hogenboom, P. E. G. Leonards, R. A. L. Peerboom, W. P. Cofino, U. A. Th. Brinkman, Ultra-Trace-Level Determination of Polar Pesticides and Their Transformation Products in Surface and Estuarine Water Samples Using Column Liquid Chromatography–Electrospray Tandem Mass Spectrometry. Journal of Chromatography A 857, 1999, 157-166.
- [5] M. Kuster, M. J. Lopez de Alda, C. Barata, D. Raldua, D. Barceló, Analysis of 17 polar to semi-polar pesticides in the Ebro river delta during the main growing season of rice by automated on-line solidphase extraction-liquid chromatography-tandem mass spectrometry. Talanta 75, 2008, 390-401.
- [6] G. Gatidou, A. Kotrikla, N. S. Thomaidis, T. D. Lekkas, Determination of the Antifouling Booster Biocides Irgarol 1051 and Diuron and their Metabolites in Seawater by High Performance Liquid Chromatography–Diode Array Detector. Analytica Chimica Acta 528, 2005, 89-99.
- [7] A. C. Gerecke, C. Tixier, T. Bartels, R. P. Schwarzenbach, S. R. Müller, Determination of Phenylurea Herbicides in Natural Waters at Concentrations Below 1 ng l⁻¹ Using Solid-Phase Extraction, Derivatisation, and Solid-Phase Microextraction–Gas Chromatography–Mass Spectrometry. Journal of Chromatography A 930, 2001, 9-19.
- [8] P. Frank, M. Karg, Determination of phenylurea pesticides in water by derivatisation with heptafluorbutyric anhydride and gas chromatography – mass spectrometry. Journal of Chromatography A 634, 1993, 87-100.

-							
Compound Endos	sulfan						
			CI				
(alfa and b	eta isomer)						
						s==0	
			C1				
CAS Number		Log K		Wat		oility [mg/L]	
115-29-7 AA-EQS	S [3.83		МАС	0.3		
		0 117 4			EQS [µg		
Inland Surface Waters 0.005		face Waters .0005	Inland St	urface Waters 0.01	Other	Surface Waters 0.004	
Available Standard Meth	nod		Method I	Description	•		
<i>EN ISO 6468:1996</i> Determination of certain o	rganochlori	ne	EN ISO 6	468·1996 descril	oes a met	hod for determination	
	insecticides, polychlorinated biphenyls and					les, polychlorinated	
chlorobenzenes - Gas chromatographic method after liquid-liquid extraction						nes (except the mono-	
inquid-inquid extraction				aters and waste v		water, ground water, he method is	
Matrix Drinking, ground, waste and surface water				e to samples cont	aining u	p to 50 mg/L of	
Sampling			suspended	1 Solids.			
Pretreatment							
64	Limit of	Quantification (LOQ): (0.001 up to 0.01 µg/L			
<u>Storage</u>							
Method Validation							
						l Number of laboratories n_{AP} percental rate of outliers	
				= x	Total mean	n after elimination of outliers	
						tion between the laboratories ucibility variation coefficient	
Interlaboratory study (Ex	traction of <i>j</i>	8-Endosulfan w	ith Hexano		~ 1	5	
Matrix	l	n _{AP} in %	= x in n	g/L s _R in p	ug/L	CV_R in %	
Surface water	14	6.6	21.2	14.	.4	67.9	
Other Analytical Method	ls						
SPE extraction of 500 ml v		00 mg SDB, elu	ution with e	thyl acetate; GC	-MS dete	ermination	
<u>LOQ for α-endosulfan ~ 11</u>	<u>ng/L</u> [1]						
<u>SPE-GC-NCI-MS</u> C18-SPE, 100 mL, LOQ f	or <u>α- or β-en</u>	udosulfan ~ 25 n	g/L [2]				
The LOQ of these alternat	ive SPE-GC	-MS methods n	nav be low	enough if good e	auinmen	t and well trained	
personnel are available. The feasible).							
Comments	10				10		
The technical grade of ende environment in particular in					osulfan (i	n a ratio of 7:3). In the	
By the SPE extraction of high	gher water v	olumes lower LO	DQs could b	e achieved.			
New research results show	that sufficie	ent LOQs in the	low ng/L c	r even pg/L rang	e can be	achieved with negative	
chemical ionization (NCI) (JC-IVIS, USIN	g SPE OF 1 OF 10	L water [3]				

- [1] J. Quintana, I. Martí, F. Ventura, Monitoring of Pesticides in Drinking and Related Waters in NE Spain with a Multiresidue SPE-GC–MS Method Including an Estimation of the Uncertainty of the Analytical Results. *Journal of Chromatography A* 938, 2001, 3-13.
- [2] E. Pitarch, C. Medina, T. Portolés, F.J. López, F. Hernández, Determination of priority organic micropollutants in water by gas chromatography coupled to triple quadrupole mass spectrometry. Analytica Chimica Acta 583, 2007, 246–258.

[3] M. Oehme, N. Theobald, A.-C. Baaß, J. Hüttig, M. Reth, S. Weigelt-Krenz, Z. Zencak, M. Haarich, Identification of organic compounds in the North and Baltic Sea, Federal Environment Agency, Research Report 20025224, 2008, ISSN 1862-4804, http://www.umweltdaten.de/publikationen/fpdf-l/3509.pdf

Compound Fluora									
CAS Number		Log K	ow	Wat	ter Solubilit	ty [mg/L]			
206-44-0 5.16					0.265				
AA-EQS [µg/L]				MAC-	EQS [µg/L]]			
Inland Surface Waters 0.1	Other Su	rface Waters 0.1	Inland S	urface Waters	Other Su	rface Waters 1			
Available Standard Metl EN ISO 17993:2003 Determination of 15 polyc hydrocarbons (PAH) in wa fluorescence detection afte Matrix Drinking, ground, Sampling Pretreatment	Method Description EN ISO 17993:2003 specifies a method using high performance liquid chromatography (HPLC) with fluorescence detection for the determination of 15 selected polycyclic aromatic hydrocarbons (PAH).								
<u>Storage</u>					Limit of Quantification (LOQ): Drinking and ground water: > 0.005 μg/L Surface water: > 0.01 μg/L				
Method Validation l Number of laboratories n_{AP} percental rate of outliers = X Total mean after elimination of outliers s_R standard deviation between the laboratories CV_R reproducibility variation coefficient Interlaboratory study 1996 in Germany									
Matrix	l	n _{AP} in %	= x in µį	g/L s _R in J	ug/L	CV _R in %			
Spiked drinking water	30	10	46.48	3 4.2	25	9.1			
Other Analytical Methods USA EPA 8270c, 1996 [1]: Semivolatile organic compounds by GC/MS. This method_claims detection limits of 10 µg/L which is obviously too high. Other analytical methods based on this standard using modern GC/MS equipment however may attain the required low LOQs. Comments References [1] http://www.accustandard.com/asi/pdfs/epa_methods/8270c.pdf									

<u>Compound</u> Hexachlor	Hexachlorobenzene						
CAS Number 118-74-1		Log K 5.73					
AA-EQS [µg/L]				MAC-	EQS [µg	ŗ/L]	
Inland Surface Waters 0.01		face Waters		urface Waters 0.05	Other	Surface Waters 0.05	
Available Standard Method EN ISO 6468:1996 Determination of certain organochlorine insecticides, polychlorinated biphenyls and chlorobenzenes – Gas chromatographic method after liquid-liquid extraction Matrix Drinking, ground, surface and waste waters Sampling Pretreatment Storage Method Validation			Method DescriptionLiquid-liquid extraction of organochlorine insecticides, chlorobenzenes and PCBs by an extraction solvent. After concentration and clean-up the sample extracts are analysed by gas chromatography, using an electron- capture detector (GC-ECD).The method is applicable to samples containing up to 50 mg/L of suspended solids.Limit of Quantification (LOQ): $\sim 0.001 - 0.01 \ \mu g/L$ I Number of laboratories n_{AP} percental rate of outliers s_R standard deviation between the laboratories CV_R reproducibility variation coefficient				
Matrix	l	n mexane) n _{AP} in %	= x in ng	$r_{Z/L}$ $s_R in$	ug/L	CV _R in %	
Surface water	15	0	48.8	16	.6	34.1	
	s in the aqu	i (GC-MS): Liquatic environments, D. van Wijk, K	nt is given b	y Barber et al. (Hexachlorobenz	2005) [1]		

Compound Hexachlorobutadiene							
CAS Number 87-68-3		Log K _{OW} 4.9		Water Solubility [mg/L] 2.55 at 20 °C			
AA-EQS	S [µg/L]			МАС-ЕС	QS [µg/L]		
Inland Surface Waters 0.1	Other S	Surface Waters 0.1	Inla	nd Surface Waters 0.6	Other Surface Waters 0.6		
Available Standard Method EN ISO 10301:1997 Determination of highly volatile halogenated hydrocarbons - Gas-chromatographic methods Matrix Sampling ISO 5667-1, 5667-2 and ISO 5667-3 Pretreatment Storage Method Validation			Method DescriptionEN ISO 10301:1997 specifies two methods for the determination of highly volatile halogenated hydrocarbons in water using gas chromatography with e.g. electron capture detector after: a) the extraction by an organic solvent or using, b) a head-space method. The static headspace method may not offer sufficient sensitivity dependent on the instrumentation available.Limit of Quantification (LOQ): 100 µg/L				
		no data a	availał	le			
Other Analytical Methods EPA method 8260B [1]. Volatile Organic Compounds by GC/MS. This method is suitable for a variety of matrices.							
<u>Comments</u>							
References[1]http://www.accusta	andard.co	m/asi/pdfs/epa_met	hods/8	260b.pdf			

Compound Hexachloroo	cyclohexane	2					
(<i>HC</i> α-, β-, γ- and		ers					
			CIMM				
			(y-HCH; lindane)				
CAS Number		Log K		W		bility [mg/L]	
608-73-1		α-HCH ~ β-HCH ~				H ~ 10 2H ~ 5	
		γ-HCH ~	3.72		γ-HCI	H ~ 7.3	
δ-HCH ~ AA-EQS [μg/L]			4.14	MA	<u>д-нсі с-EQS [µ</u>	H ~ 10 g/L]	
Inland Surface Waters	Other Su	rface Waters	Inland St	urface Waters	Other	Surface Waters	
0.02	(0.002		0.04		0.02	
Available Standard Meth	od		Method I	Description			
	EN ISO 6468:1996				. f	alalanina inggatiai dag	
Determination of certain organochlorine insecticides, polychlorinated biphenyls and						chlorine insecticides, xtraction solvent. After	
chlorobenzenes – Gas chromatographic method						nple extracts are	
after liquid-liquid extraction				by gas chroma etector (GC-EC		using an electron-	
Matrix Drinking, ground, surface and waste waters			-		,	es containing up to 50	
Sampling			mg/L of suspended solids.				
<u>Pretreatment</u>			Limit of	Quantification	(LOQ):	~ 0.001 – 0.01 µg/L	
<u>Storage</u>	<u>Storage</u>						
Method Validation			I			<i>l</i> Number of laboratories	
						n_{AP} percental rate of outliers =	
				S _R S		X Total mean ation between the laboratories lucibility variation coefficient	
Interlaboratory study (Ex	traction of	y-HCH with He	xane)				
Matrix	1	n _{AP} in %	= x in ng	g/L S _R ii	n µg/L	CV_R in %	
Surface water	15	14.3	38.6	1	4.3	38.4	
Other Analytical Method	ls						
Solid-phase extraction gas	chromatog	raphy - mass spe	ectrometrv				
GC-MS determination of t				s [1-4]			
LOQ ~ 10 ng/L for α -HC	н						
5 ng/L for β-HCH							
5 ng/L for γ -HCH and							
10 ng/L for δ- HC	^C H		(SPE extraction of 200 mL water) [1,2]				
LOQ for γ -HCH (lindane) LOQ for γ -HCH (lindane)							
SPE-GC- triple quadrupole	e-MS-MS						
C18-SPE, 100 mL, SRM 2		$LOQ \sim 25 \text{ ng/L}$ (for lindane) [5]				

SPE-GC-NCI-MS

C18-SPE, 100 mL, LOQ ~ 25 ng/L (for lindane) [5]

EPA methods 508, 608 (GC-ECD). EPA 625 (GC-MS) may also be used but losses of HCH can occur due to the alkaline extraction procedure.

Comments

HCH exists in eight isomer forms. Technical-grade HCH was used as an insecticide and typically contained 10-15% γ -HCH (lindane) as well as the alpha (α), beta (β), and delta (δ) forms of HCH.

- [1] S. Lacorte, I. Guiffard, D. Fraisse, D. Barceló, Broad Spectrum Analysis of 109 Priority Compounds Listed in the 76/464/CEE Council Directive Using Solid-Phase Extraction and GC/EI/MS. *Analytical Chemistry* 72, 2000, 1430-1440.
- [2] D. de Almeida Azevedo, S. Lacorte, T. Vinhas, P. Viana, D. Barceló, Monitoring of Priority Pesticides and Other Organic Pollutants in River Water From Portugal by Gas Chromatography–Mass Spectrometry and Liquid Chromatography–Atmospheric Pressure Chemical Ionization Mass Spectrometry. *Journal of Chromatography A* 879, 2000, 13-26.
- [3] J. Quintana, I. Martí, F. Ventura, Monitoring of Pesticides in Drinking and Related Waters in NE Spain with a Multiresidue SPE-GC–MS Method Including an Estimation of the Uncertainty of the Analytical Results. *Journal of Chromatography A* 938, 2001, 3-13.
- [4] C. Planas, A. Puig, J. Rivera, J. Caixach, Analysis of pesticides and metabolites in Spanish surface waters by isotope dilution gas chromatography/mass spectrometry with previous automated solid-phase extraction; Estimation of the uncertainty of the analytical results. Journal of Chromatography A, 1131 (2006) 242– 252.
- [5] E. Pitarch, C. Medina, T. Portolés, F.J. López, F. Hernández, Determination of priority organic micropollutants in water by gas chromatography coupled to triple quadrupole mass spectrometry. Analytica Chimica Acta 583, 2007, 246–258

<u>Compound</u> Isoproturon			H ₃ C H ₃ C CH ₃					
			сн _з					
CAS Number 34123-59-6	8			Wat	er Solut ~ ^	bility [mg/L] 70		
AA-EQ	S [µg/L]	I		MAC-	EQS [µĮ	g/L]		
Inland Surface Waters 0.3	Other Su	face Waters 0.3	Inland Su	urface Waters 1.0	Other	Surface Waters 1.0		
Available Standard Meth EN ISO 11369:1997 Water quality - Determina treatment agents in water I liquid chromatography wit solid-liquid extraction. Matrix Drinking and grout Sampling Pretreatment Storage	Method Description The plant treatment substances in the water sample are extracted by solid-liquid extraction (SPE) on reversed-phase (RP)-C18 material, eluted with a solvent, and then separated, identified and quantified by high performance liquid chromatography (HPLC) using UV detection. Limit of Quantification (LOQ): 0.1 µg/L							
Interlaboratory study	$= \frac{l \text{ Number of laboratories}}{X \text{ Total mean after elimination of outliers}}$ $= X \text{ Total mean after elimination of outliers} s_{R} \text{ standard deviation between the laboratories} CV_{R} reproducibility variation coefficient}$ Interlaboratory study							
Matrix	l	n_{AP} in %	x in µ	g/L s _R in J	ug/L	CV_R in %		
Drinking water	32	0	0.172	7 0.03	94	22.8		
Ground water	32	6	0.111	0 0.02	249	22.5		
Other Analytical Methods Liquid Chromatography - Mass Spectrometry Identification and quantification of isoproturon (and other pesticides) by liquid chromatography coupled to (tandem) mass spectrometric detection (LC-MS-MS) using positive electrospray ionization (ESI). LC-MS fragment ions: m/z 207 [1,2] LC-MS-MS transitions: 207 > 72 [3,4] LOQ ~ 1 ng/L (depending on the SPE enrichment factor) Gas Chromatography - Mass Spectrometry (after derivatisation) Phenylurea herbicides require a derivatisation step to prevent the degradation of these thermolabile compounds in the GC injector; LOQ ~ 1 ng/L [5]. SPE-derivatisation-GC-MS: LOQ ~ 40 ng/L [6]								

The EN ISO 11369 HPLC-UV method is only applicable for drinking and ground water, not to the analysis of contaminated surface water. GC-MS determination of phenylurea herbicides is difficult due to the necessary derivatisation step. LC-MS-MS seems to be the method of choice.

- [1] C. Crescenzi, A. Di Corcia, R. Samperi, N. L. Dietz, E. Guerriero, Development of a Multiresidue Method for Analyzing Pesticide Traces in Water Based on Solid-Phase Extraction and Electrospray Liquid Chromatography Mass Spectrometry. *Environmental Science & Technology* 31, 1997, 479-488.
- [2] R. Loos, G. Hanke, S. J. Eisenreich: Multi-Component Analysis of Polar Water Pollutants Using Sequential Solid-Phase Extraction Followed by LC-ESI-MS. *Journal of Environmental Monitoring* 5, 2003, 384-394.
- [3] R. Bossi, K. V. Vejrup, B. B. Mogensen, W. A. H. Asman, Analysis of Polar Pesticides in Rainwater in Denmark by Liquid Chromatography–Tandem Mass Spectrometry. *Journal of Chromatography A* 957, 2002, 27-36.
- [4] M. Kuster, M. J. Lopez de Alda, C. Barata, D. Raldua, D. Barceló, Analysis of 17 polar to semi-polar pesticides in the Ebro river delta during the main growing season of rice by automated on-line solidphase extraction-liquid chromatography-tandem mass spectrometry. *Talanta* 75, 2008, 390-401.
- [5] A. C. Gerecke, C. Tixier, T. Bartels, R. P. Schwarzenbach, S. R. Müller, Determination of Phenylurea Herbicides in Natural Waters at Concentrations Below 1 ng l⁻¹ Using Solid-Phase Extraction, Derivatisation, and Solid-Phase Microextraction–Gas Chromatography–Mass Spectrometry. *Journal of Chromatography A* 930, 2001, 9-19.
- [65] P. Frank, M. Karg, Determination of phenylurea pesticides in water by derivatisation with heptafluorbutyric anhydride and gas chromatography mass spectrometry. *Journal of Chromatography* A 634, 1993, 87-100.

Compound Lead and its	сотро	unds					
CAS Number		Log F	L/kg]		Water	Solubility [mg/L]	
7439-92-1		_	/water: 5.6 (Pb) [1]		depending on compound		
AA-EQ	AA-EQS [µg/L]			MAC-I	EQS [µĮ	g/L]	
Inland Surface Waters 7.2	Othe	r Surface Waters 7.2	Inland Surface V not applicable		Oth	er Surface Waters not applicable	
Available Standard Meth EN ISO 17294-2:2004 Application of inductively spectrometry (ICP-MS) - 1 62 elements <u>Matrix</u> Drinking waters, g waters and waste waters <u>Sampling</u> ISO 5667-1, 566 <u>Pretreatment</u> For dissolve aqueous sample through a filter. Adjust the pH of the HNO ₃ . <u>Storage</u>	couple Part 2: I ground 67-2 an ed elem 0.45-µ:	Determination of waters, surface <i>ad 5667-3</i> nents, filter m pore membrane	Method Description EN ISO 17294-2:2004 specifies a method for the determination of the lead in water (for example drinkin, water, surface water, groundwater, wastewater and eluates). Taking into account the specific and additiona occurring interferences, these elements can also be determined in digests of water, sludges and sediments. The working range depends on the matrix and the interferences encountered. Limit of Quantification (LOQ): Drinking water and relatively unpolluted water:			(for example drinking wastewater and becific and additionally tents can also be dges and sediments. matrix and the	
<u>Method Validation</u> Interlaboratory study 199		$s_{\rm R}$ stand	lard devia	<i>l</i> Number of laboratories n_{AP} percental rate of outliers n after elimination of outliers tion between the laboratories ucibility variation coefficient			
Matrix	l	n _{AP} in %	$= x in \mu g/L$	s _R in µ	g/L	CV _R in %	
Surface water	39	2.5	13.6	1.13	3	8.3	
Other Analytical Method EN ISO 15586:2003 deter		n using atomic abso	prption spectrometry	with elec	ctrother	mal atomization in a	

graphite furnace. The detection limit of the method for each element depends on the sample matrix as well as of the instrument, the type of atomizer and the use of chemical modifiers. For water samples with a simple matrix (i.e. low concentration of dissolved solids and particles), the method detection limits will be close to instrument detection limits. The minimum acceptable detection limit values for a 20 μ L sample volume are specified.

EN ISO 11885:1997 specifies a method by inductively coupled plasma atomic emission spectroscopy.

DIN 38406-6:1998-07 determination using atomic absorption spectrometry

DIN 38406-16 (1990-03) specifies a voltammetric determination.

EPA 200.8 (1994) determination of trace elements in waters by inductively coupled plasma - mass spectrometry (LOQ: $0.6 \mu g/L$)

Standard Methods Online (http://standardmethods.org/) 3125: Metals in Water by ICP/MS (LOQ: 0.005 µg/L)

Compound Mercury and t	Mercury and its compounds						
CAS Number		Log K	L/kg]	Water Solubility [mg/L]			
7439-97-6		suspended matter/ sediment/water: 4.	<i>water:</i> 5.3 (Hg) [1] 9 (Hg) [1]		Depending on compound		
AA-EQ	S [µg/L		MAC-EQS [µg/L]				
Inland Surface Waters 0.05	Other	Surface Waters 0.05	Inland Surface V 0.07	d Surface Waters 0.07 Other Surface Waters 0.07			
Available Standard Meth EN ISO 17852:2008 Determination of mercury spectrometry <u>Matrix</u> Drinking waters, g waters <u>Sampling</u> ISO 5667-1, 560 <u>Pretreatment</u> stabilise wi and acidification to pH< 2 Acid	by ator ground - 67-2 ard th Pota:	waters and surface ad 5667-3 ssium dichromate	Method Description EN ISO 17852:2008 specifies a method for the determination of mercury in water using atomic fluorescence spectrometry. Limit of Quantification (LOQ): appr. 0.001 μg/L				
<u>Storage</u>			(largely depends on the operational parameters)				
<u>Method Validation</u> Interlaboratory study 199	9 in Gr	eat Britain		s _R star	Total mean dard devia	<i>l</i> Number of laboratories n_{AP} percental rate of outliers n after elimination of outliers tion between the laboratories ucibility variation coefficient	
Matrix	l	n _{AP} in %	$= x in \mu g/L$	s _R in J	ug/L	CV _R in %	
Surface water	18	9.4	44.2	11.	57	25.8	
Other Analytical Method EN 12338:1998 specifies to ISO 16590:2000 specifies Comments							
<u>References</u>							

Compound Naphta	halene						
CAS Number 91-20-3		Log K 3.3	ow	Water	Solubil 31	ity [mg/L]	
AA-EQ	S [µg/L]			МАС-ЕС	QS [µg/I	2]	
Inland Surface Waters 2.4	Other Su	rface Waters 1.2	Inland Surface Waters not applicableOther Surface Waters not applicable				
Available Standard Meth	nod		Method I	Description	I		
<i>EN ISO 17993:2003</i> Determination of 15 polyc hydrocarbons (PAH) in wa fluorescence detection after <i>EN ISO 15680: 2002</i>	<i>EN ISO 17993:2003</i> specifies a method using high performance liquid chromatography (HPLC) with fluorescence detection for the determination of 15 selected polycyclic aromatic hydrocarbons (PAH).						
 EN ISO 15680: 2003 Gas-chromatographic determination of a number of monocyclic aromatic hydrocarbons, naphthalene and several chlorinated compounds using purgeand-trap and thermal desorption Matrix Drinking, ground, waste and surface water Sampling 			<i>EN ISO 15680:2003</i> specifies a general method for the determination of volatile organic compounds (VOCs) in water by purge-and-trap isolation and gas chromatography (GC). Annexes A, B and C provide examples of analytes that can be determined. Detection is carried out by mass spectrometry in the electron impact mode (EI).				
				Quantification (
<u>Pretreatment</u> <u>Storage</u>			HPLC - Drinking and ground water: > 0.005 μg/L HPLC - Surface water: > 0.01 μg/L Purge-and-Trap/Thermal Desorption GC-MS: > 0.01 μg/L				
Interlaboratory study (ISO	0 17993)			s _R standard d	n_{AP} p mean after leviation b	Number of laboratories ercental rate of outliers r elimination of outliers etween the laboratories ity variation coefficient n.a. = not available	
Matrix	l	n _{AP} in %	= x in µį	g/L s _R in	ug/L	CV _R in %	
Spiked drinking water	33	3	52.85	5 15	.5	29.3	
Interlaboratory study (IS	D 15680)		•	·			
Matrix	1	n _{AP} in %	= x in µį	g/L $s_R in$	ug/L	CV _R in %	
Surface water (0.2 μ g/L)	17	n.a.	n.a.	n.a	a.	32	
Other Analytical Method	<u>ls</u>						
USA EPA 8270c,1996, [1] methods based on this star LOQs.							
Comments							
References[1]http://www.accus	tandard.cor	n/asi/pdfs/epa_r	nethods/827	70c.pdf			

Nickel and it: CAS Number		-				
CAS Number						
		Log K	K _D [L/kg]	Water Solu	ıbility [mg/L]	
7440-02-0		suspended matter/ sediment/wa	<i>water:</i> 4.6 (Ni) [1] <i>ter:</i> 4.0 (Ni) [1]	depen	ding on compound	
AA-EQS	5 [µg/L			MAC-EQS [µ	g/L]	
Inland Surface Waters 20	Othe	r Surface Waters 20	Inland Surface Waters not applicableOther Surface Water not applicable			
Available Standard Meth	nod		Method Description	<u>on</u>		
<i>EN ISO 17294-2:2004</i> Application of inductively spectrometry (ICP-MS) - F 62 elements <u>Matrix</u> Drinking waters, g waters and waste waters <u>Sampling</u> <i>ISO 5667-1, 566</i> For dissolved elements, fil through a 0.45 μm pore mo	Determination of waters, surface <i>ad 5667-3</i> eous sample	<i>EN ISO 17294-2:2004</i> specifies a method for the determination of the nickel in water (for example drinking water, surface water, groundwater, wastewater and eluates). Taking into account the specific and additionally occurring interferences, these elements can also be determined in digests of water, sludges and sediments. The working range depends on the matrix and the interferences encountered.				
pH of the filtrate to < 2 with HNO ₃ . Storage			Limit of Quantification (LOQ): Drinking water and relatively unpolluted water: 0.1 - 1.0 μg/l			
Interlaboratory study 199	7 in Ge	rmany		s _R standard devi	n_{AP} percental rate of outliers an after elimination of outliers ation between the laboratories ducibility variation coefficient	
Matrix	l	n _{AP} in %	$= x in \mu g/L$	s _R in µg/L	CV_R in %	
Surface water	35	11	5.44	0.786	14.5	
Other Analytical Method	<u>ls</u>					
EN ISO 15586:2003 detern graphite furnace. The detect the instrument, the type of (i.e. low concentration of c detection limits. The minin	ction li atomiz lissolve	mit of the method fo er and the use of che ed solids and particle	r each element depen emical modifiers. For es), the method detect	ds on the samp water samples ion limits will	ble matrix as well as of s with a simple matrix be close to instrument	
EN ISO 11885:1997 specie	fies a n	nethod by inductivel	y coupled plasma ato	mic emission s	pectroscopy.	
EPA 200.8 1994 [1]: Deter spectrometry (LOQ: 0.5 µg		on of trace elements	in waters by inductiv	vely coupled pl	asma - mass	
Standard Methods Online	(http://s	standardmethods.org	/) 3125: Metals in W	ater by ICP/M	S (LOQ: 0.02 μg/L)	
Comments						
References [1] http://www.accus	tandarc	l.com/asi/pdfs/epa_r	nethods/200_8.pdf			

Compound Nonylphenol							
		(One Isome	·) HO				
			Wat	er Solubility [mg/L] ~ 6			
S [µg/L]		MAC-EQS [µg/L]					
	face Waters	Inland St			Surface Waters		
			2.0				
od		Method I	Description				
<i>EN ISO 18857-1:2006</i> Determination of selected alkylphenols - Part 1: Method for non-filtered samples using liquid extraction and gas chromatography with mass selective detection <u>Matrix</u> Drinking, ground and surface water				Method for the determination of 4-nonylphenols (mixture of isomers) in non-filtered samples of drinking water, ground water and surface water. Extraction of nonylphenol from the acidified water			
<u>Matrix</u> Drinking, ground and surface water <u>Sampling</u> <u>Pretreatment</u>				sample with toluene. Cleaning of the extract, if necessary with silica. Gas chromatographic separation and identification of the alkylphenol by mass spectrometry without derivatisation (mass fragments m/z 135 and 107). Quantification with an internal standard (¹³ C p-n-NP; m/z 113).			
<u>Storage</u>							
		<u>Limit of Quantification (LOQ)</u> : 0.02 to 0.2 µg/L					
2			s _R star	Total mea dard devia	n_{AP} percental rate of outliers n after elimination of outliers tion between the laboratories lucibility variation coefficient		
l	n _{AP} in %	= x in µ	g/L $s_R in$	ug/L	CV_R in %		
11	26.7	0.082	8 0.0	16	18.8		
<u>ls</u>							
Surface water1126.7 0.0828 0.016 18.8Other Analytical MethodsSolid-phase extractionExtraction of alkylphenols from water with solid-phase extraction (SPE) using C18 or polymeric adsorbents.Elution with methanol, acetone, ethylacetate, or dichloromethane [2-6].Liquid Chromatography - Mass SpectrometryIdentification and quantification of the analytes by liquid chromatography coupled to (tandem) massspectrometric detection (LC-MS-MS) using negative electrospray ionization (ESI).LC-MS fragment ion: m/z 219 [2]LC-MS-MS transitions: 219 > 133 and 219 > 147 [3,4]; LOQ ~ 5 ng/LInternal standard: 4n-NP; transition 219 > 106GC-MS after derivatisationGe-MS determination have been reported.E.g., the phenol group can be converted to a pentafluorobenzoylate ester (LOQ ~ 0.05 ng/L) [5], or silylatedusing bis(trimethylsilyl)trifluoroacetamide (BSTFA) (LOQ ~ 1 ng/L) [6], or methyl-N-(trimethylsilyl)trifluoroacetamide (MSTFA) [7].							
	S [µg/L] Other Sur alkylphenol mples using tography wi and surface 2 2 1 11 s from water ctone, ethyla <u>Mass Spectr</u> cation of the C-MS-MS)	Log K ~ 4.4 S [µg/L] Other Surface Waters 0.3 odd alkylphenols - Part 1: mples using liquid tography with mass and surface water and surface water 2 1 n_{AP} in % 11 26.7 Is from water with solid-phase etion of the analytes by lique C-MS-MS) using negative	Log Kow ~ 4.48S [µg/L]Inland SuOther Surface Waters 0.3Inland SuodMethod Ialkylphenols - Part 1: mples using liquid tography with massMethod fo of isomers ground wayand surface waterExtraction sample wi with silica identificat without day Quantification 113).IInimit of Quantification sample wi with silication identification sample wi with silication sample wi with silication sample wi with silication identification sample wi with silication sample wi with silication of the analytes by liquid chroma C-MS-MS) using negative electrospray	Log Kow ~ 4.48Wat ~ 4.48S [µg/L]MAC- MAC-Other Surface Waters 0.3Inland Surface Waters 2.0nodMethod Descriptionalkylphenols - Part 1: mples using liquid tography with massMethod for the determinat of isomers) in non-filtered ground water and surface waterand surface waterMethod for the determinat of isomers) in non-filtered ground water and surface is with silica. Gas chromatog identification of the alkylp without derivatisation (ma Quantification with an into 113).Limit of Ouantification (ma Quantification with an into 113). \bar{x} in µg/L \bar{x} s star C \bar{x} in µg/L \bar{x} in µg/L \bar{x} s star C \bar{x} s s	(One Isomer)(One Isomer)(One Isomer)(One Isomer)Value (Concentry)Inlog Kow ~ 4.48 MAC-EQS [µµ]Other Surface Waters 0.3 Other 2.0 NodMethod DescriptionMethod for the determination of 4- of isomers) in non-filtered samples ground water and surface water.Extraction of nonylphenol from the sample with toluene. Cleaning of t with silica. Gas chromatographic s identification of the alkylphenol by without derivatisation (mass fragm Quantification with an internal stat 113).Imit of Ouantification (LOO): of Total mea s_R standard devia CV_R reprodCInIn name with solid-phase extraction (SPE) using C18 or pol tone, ethylacetate, or dichloromethane [2-6].Mass Spectrometry axion of the analytes by liquid chromatography coupled to (tan- C-MS-MS) using negative electrospray ionization (ESI).		

"Nonylphenol" consists of a number of 211 possible differently branched nonyl chains isomers; it consists mainly (~90%) of para substituted 4-nonylphenol. Only recently it was recognized that for a correct risk assessment, isomer-specific toxicological studies and analysis is important because the estrogenic effects of the individual nonylphenol isomers is heavily dependent on the structure of the alkyl side chain. It is estimated that in biological and environmental relevant matrices approximately 50-80 isomers are present [8,9]. Chromatographic separation of all isomers is not possible (at the time being); it might become feasible in the future with two-dimensional GC [8]. Therefore (for now), the sum of the different nonylphenol isomers should be analysed.

When using liquid chromatography (LC), the nonylphenols have to get chromatographically separated from the ethoxy carboxylate metabolites (NPECs) because they produce the same MS ions.

Nonylphenols are relatively polar compounds, and therefore GC-MS without derivatisation can give rise to poor chromatographic peaks. Thus, nonylphenols are often derivatized prior to GC-MS.

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Compound Octylp (4-tert-Oct		но	H ₃ C	н ₃ - сн ₃ - сн ₃		
CAS Number 140-66-9		Log K 5.28		Wat	er Solubi	lity [mg/L]
AA-EQS	5 [µg/L]	5.20		MAC-	EQS [µg/	L]
Inland Surface Waters 0.1		rface Waters 0.01	Inland Su	urface Waters not applicable		
$= \frac{n_{AP} p}{x}$ X Total mean after s_{R} standard deviation b					acidified water sample t, if necessary with ton and identification netry without 135 and 107). dard (¹³ C p-n-NP; m/z	
Matrix	l	n _{AP} in %	= x in µį	g/L s _R in J	ug/L	CV _R in %
Surface water	13	13.3	0.066	8 0.01	789	26.8
Other Analytical Method Solid-phase extraction Extraction of alkylphenols Elution with methanol, ace Liquid Chromatography - I Identification and quantific spectrometric detection (LC LC-MS fragment ion: m/z LC-MS-MS transitions: 20 Internal standard: 4n-NP; t <u>GC-MS after derivatisation</u> Several derivatisation tech E.g., the phenol group can (LOQ ~ 2.6 ng/L) [6].	from water tone, ethyla Mass Spect cation of the C-MS-MS) 205 [2] 5 > 133 [3- ransition 2] <u>1</u> niques for a	acetate, or dichle <u>rometry</u> e analytes by liq using negative 4] 19 > 106 .lkylphenols prio	oromethane uid chroma electrospray or to GC-M	[2-6]. tography coupled ionization (ESI S determination	d to (tando). have been	em) mass

The term "octylphenol" represents a large number of isomeric compounds of the general formula $C_8H_{17}C_6H_4(OH)$. The octyl group may be branched in a variety of ways or be a straight chain. Of these potential isomers, 4-*tert*-octylphenol (CAS No. 140-66-9) is the most commercially (and toxicologically) important [7]. It has the MS-MS transistion 205 >133.

Another analytical standard is available: 4-octylphenol (CAS No. 1806-26-4). This standard contains linear octylphenol; it shows the characteristic MS-MS transistion 205 > 106.

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<u>Compound</u> Pentachlorobenzene							
CAS Number 60-93-5		Log K 5.17		Water Solubility [mg/L] 0.831			
60-93-3 AA-EQS	5 [µg/L]	5.17		MAC-			
Inland Surface Waters 0.007		face Waters .0007	Inland Surface Waters Other Surf		Other Surface Waters not applicable		
Available Standard Method EN ISO 6468:1996Method DescriptionDetermination of certain organochlorine insecticides, polychlorinated biphenyls and chlorobenzenes – Gas chromatographic method after liquid-liquid extractionLiquid-liquid extraction of organochlorine insectici chlorobenzenes and PCBs by an extraction solvent. concentration and clean-up the sample extracts are analysed by gas chromatography, using an electron capture detector (GC-ECD).Matrix SamplingDrinking, ground, surface and waste watersPretreatmentImit of Quantification (LOQ): c 0.001 – 0.01 µg					by an extraction solvent. After p the sample extracts are graphy, using an electron-)). to samples containing up to 50		
Method Validation		no da	ta available	;			
subject to evaporation with	as spectrome he ions 250, CD) [1] and B] water (elution), product 14 on of bigger chlorobenzer n a half-life of 3 m/sec.	, 215, 180 625 (GC-MS) [2 on with ethyl ac 42; LOQ = 25 n water volumes, ne will adsorb s of 6.5 hours esti The volatilizati	cetate / DCM ng/L. , a lower LC strongly to s imated for a on half-life	M) followed by C DQ might be ach sediments and big evaporation from from a model po	GC- triple-quad MS-MS. ieved. oconcentrate in fish. It will be a river 1 m deep, flowing at 1 ond, which considers the effects		
hydrolyze. References							
[1] <u>http://www.accust</u>	tandard.con	n/asi/pdfs/epa_n	nethods/50;	5.pdf			
[2] <u>http://www.accust</u>	tandard.con	n/asi/pdfs/epa_n	nethods/62;	5.pdf			

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CAS Number 87-86-5		Log K ~ 5.(
AA-EQ	S [µg/L]		, 	MAC-EQS [µg/L]			
Inland Surface Waters 0.4		face Waters 0.4	Inland Surface WatersOther Surface Water11				
Available Standard Meth EN 12673:1998 Gas chromatographic dete selected chlorophenols in v Matrix Drinking, ground, surface water Sampling Pretreatment Storage Method Validation	rmination of water		Method DescriptionThis European Standard describes the gas chromatographic determination of 19 chlorophenols in water.The methods consists of acetylaction of the chlorophenols with acetic anhydride followed by liquid/liquid extraction with hexane and determination by gas chromatography (GC) and electrone capture detection (ECD) or mass selective detection (MSD).Limit of Quantification (LOQ): volume:50 mL)				
<i>l</i> Number of labora					<i>AP</i> percental rate of outliers after elimination of outliers on between the laboratories		
Interlaboratory study Nov							
Interlaboratory study Nov Matrix	l	n _{AP} in %	= x in µ	g/L s _R in p	ug/L	CV _R in %	
	<i>l</i> 12	<i>n_{AP} in %</i> 22.2	= <i>x</i> in μ ₂ 0.11	g/L s _R in µ 0.02		<i>CV_R</i> in % 24	
Matrix	12 13			0.02	28		

SPE-LC-MS

LC-MS of Pentachlorophenol is difficult due to bad ionisation of the hydroxy group. However, off- and on-line SPE-LC-MS methods have been reported.

On-line SPE extraction of 100 mL river water (pH 3) using LiChrolut EN precolumns; elution with the wateracetonitrile-methanol gradient; LC-APCI-MS analysis; LOQ in SIM mode ~ 0.1 ng/L [5]

On-line SPE extraction of 10 mL river (pH 2.5) water using polymeric adsorbents; LC-APCI-MS analysis; LOQ in SIM mode \sim 5 ng/L [6]

Off-line SPE with PS-DVB membrane extraction disk of 500 mL tap water, elution with acetonitrile; LC-APCI-MS with post-column addition of diethylamine; SIM ions 263, 265, 267; LOQ ~ 20 ng/L [7].

Comments

The SPME-GC-MS and SPE-on-line-LC-MS methods have lower LOQ but are not standardized. Other derivatisation reagents such as pentafluorobenzoyl chloride can be used prior to GC analysis.

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Compound								
Benzo[Benzo[b]fl Benzo[g,l Benzo[k]fl Indeno[1,2		$\begin{array}{c} C_{20}H_{12} \\ C_{20}H_{12} \\ C_{22}H_{12} \\ C_{20}H_{12} \\ C_{22}H_{12} \\ C_{22}H_{12} \end{array}$						
	CAS	Number	Log K _{OW}		Water S	Water Solubility [mg/L]		
Benzo[a]pyrene Benzo[b]fluoroanthene Benzo[k]fluoroanthene Benzo[g,h,i]perylene Indeno[1,2,3-cd]pyrene	20 20 19	0-32-8 05-99-2 07-08-9 01-24-2 03-39-5	6.13 5.78 6.11 6.63 6.70	5.78 6.11 6.63 6.70		0.00162 0.0015 0.0008 0.00026 0.00019		
		AA-EQS [µg/L]			QS [µg/L]		
	Inland Wa		Other Surface Waters	Inl	and Surface Waters	Other Surface Waters		
Benzo[a]pyrene	0.0)5	0.05		0.1	0.1		
Benzo[b]fluoroanthene Benzo[k]fluoroanthene	$\Sigma =$	0.03	$\Sigma = 0.03$	no	t applicable	not applicable		
Benzo[g,h,i]perylene								
Indeno[1,2,3-cd]pyrene	$\Sigma = 0$	0.002	$\Sigma = 0.002$	no	t applicable	not applicable		
Determination of 15 polyc hydrocarbons (PAH) in wa fluorescence detection afte extraction <u>Matrix</u> Drinking, ground, water	ater by HPL er liquid-liqu	C with iid	<i>EN ISO 17993:2</i> performance liqu fluorescence dete polycyclic aroma	id chi	romatography (
<u>Sampling</u> <u>Pretreatment</u> <u>Storage</u>			Limit of Quantification (LOQ): Drinking and ground water: > 0.005 μg/L Surface water: > 0.01 μg/L					
<u>Method Validation</u>	study for spi	ked drinking	a water 1996 (Gorn	nan)	= <i>X</i> Total mean <i>s</i> _R standard deviat	<i>l</i> Number of laboratories <i>t</i> _{AP} percental rate of outliers after elimination of outliers ion between the laboratories icibility variation coefficient		
National interlaboratory s	study for spi		=	nan)	= <i>X</i> Total mear <i>s</i> _R standard deviat CV _R reprodu	<i>n_{AP}</i> percental rate of outliers after elimination of outliers ion between the laboratories icibility variation coefficient		
National interlaboratory s Substance		n _{AP} in %	= x in pg/L	nan)	= X Total mear $s_{\rm R}$ standard deviat $CV_{\rm R}$ reprodu $s_{\rm R}$ in pg/L	n_{AP} percental rate of outliers after elimination of outliers ion between the laboratories icibility variation coefficient CV_R in %		
National interlaboratory s	1		=	nan)	= <i>X</i> Total mear <i>s</i> _R standard deviat CV _R reprodu	<i>n_{AP}</i> percental rate of outliers after elimination of outliers ion between the laboratories icibility variation coefficient		

Benzo[g,h,i]perylene	32	6.2	25.21	5.941	23.6
Indeno[1,2,3-cd]pyrene	29	12	26.31	4.417	17.9

Other Analytical Methods

EPA 8270c [1] claims detection limit of 10 μ g/L which is obviously too high. Other analytical methods based on this standard using modern GC/MS equipment however may attain the required low LOQs.

Comments

The methods do not attain low enough LOQs and uncertainties for compliance checking with the AA-EQS for the sum of *Benzo[b]fluoroanthene and Benzo[k]fluoroanthene as well as the sum of Benzo[g,h,i]perylene and Indeno[1,2,3-cd]pyrene*. In addition, not enough validation data are available regarding the analysis of surface water samples in particular samples containing substantial amounts of SPM.

A new ISO standard for the determination of PAH in water using gas chromatography with mass spectrometry detection in under development (ISO/CD 28540)

References

[1] http://www.accustandard.com/asi/pdfs/epa_methods/8270c.pdf

Compound				CI				
Sime	ızine							
CAS Number		Log K	ow Water Solubility [mg/L]					
1912-24-9		~ 2.2	~ 6.2					
AA-EQ			MAC-EQS [µg/L]					
Inland Surface Waters 1	rs Other Surface Waters		Inland Su	urface WatersOther Surface Waters44				
	vailable Standard Method			<u>Description</u>				
 EN ISO 10695:2000 Determination of selected organic nitrogen and phosphorus compounds - Gas chromatography methods <u>Matrix</u> Drinking waters, ground waters, surface waters and waste waters containing up to 50 mg/L of suspended solids 			Liquid/liquid extraction with dichloromethane or liquid/solid extraction (SPE) on reversed-phase (RP)-C18 material or other adsorbent. Elution of the cartridges with e.g. methanol or acetone. After concentration, the sample extracts are analysed by gas chromatography, using a nitrogen-phosphorus or mass spectrometric detector.					
<u>Sampling</u>			1					
				Quantification (
<u>Pretreatment</u>	<u>Pretreatment</u>			Liquid/liquid extraction method: 0.5 µg/L Liquid/solid extraction method: 0.012 µg/L				
Storage			1			10		
Method Validation			L			<i>l</i> Number of laboratories n_{AP} percental rate of outliers		
Interlaboratory study 199	2 for liquid	halid artugation		s _R star	Total mean dard devia	n after elimination of outliers tion between the laboratories ucibility variation coefficient		
Matrix	5 jor iiquiu 1		=		ua/I			
	l	n _{AP} in %	x in µį	-	-	CV _R in %		
Drinking water	12	16.4	0.058	3 0.00)44	27.3		
Other Analytical Method	ls							
Other Analytical Methods Gas Chromatography - Mass Spectrometry GC-MS determination of the ions 201 and 186; LOQ ~ 1 ng/L (after SPE) [1, 2] (EPA method 525) GC-NPD EPA method 507 [3] GC-ECD EPA method 505; microextraction with hexane and GC-ECD analysis [32] Liquid Chromatography - Mass Spectrometry Identification and quantification of simazine (and other pesticides) by liquid chromatography coupled to (tandem) mass spectrometric detection (LC-MS-MS) using positive electrospray ionization (ESI). LC-MS fragment ions: m/z 202 and 124 [4] LC-MS-MS transition: 202 > 132 [5, 6]								
$LOQ \sim 1 \text{ ng/L}$ (depending	on the St E	entrennent ide						

- [1] T. D. Bucheli, F. C. Grüebler, S. R. Müller, R. P. Schwarzenbach, Simultaneous Determination of Neutral and Acidic Pesticides in Natural Waters at the Low Nanogram per Liter Level. *Anaytical Chemistry* 69, 1997, 1569-1576.
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<u>Compound</u> Tributyltin compounds					Sn+	^
CAS Number 688-73-3				Wat	ter Solubi ~ 2 m	l lity [mg/L] g/L
AA-EQS	5 [µg/L]			MAC	EQS [µg/	[L]
Inland Surface Waters 0.0002		face Waters .0002	Inland Surface Waters 0.0015		Other S	burface Waters 0.0015
Available Standard Meth EN ISO 17353:2005 Determination of selected Gas chromatographic meth <u>Matrix</u> Drinking, surface a containing not more than 2 material. Sampling Pretreatment Storage	organotin c nod and waste w	vaters	Method fo organotin Compoun Monobuty Dibutyltir Tributyltir Tetrabuty Organotin tetraethyl- The extrac concentra capillary g suitable sy determine an interna	vitin cation (MB cation (DBT) F n cation (TBT) I litin (TTBT) Bu ₄ compounds in v borate (NaBEt ₄) ct can be cleaned tion, the tetra-su gas chromatogra ystem (MS, FPD	/or cation: T) BuSn ³⁺ Bu_2Sn^{2+} Bu_3Sn^+ Sn water are () and extra d with silie bstituted (phy and d D, AED). T over the t ire.	s in water. ethylated with sodium acted with hexane. ca. After OTC are separated by etected with a 'he concentration is otal procedure using
<u>Method Validation</u> Interlaboratory study 1996	8 in Germa	ny			ndard deviat	<i>l</i> Number of laboratories A_{AP} percental rate of outliers = <i>X</i> Total mean ion between the laboratories cibility variation coefficien
Matrix	l	n_{AP} in %	x in ng	g/L s _R in	µg/L	CV_R in %
Spiked surface water	11	8.3	388.2	. 92.	16	23.7
Other Analytical Method LLE-EI-GC-MS 100 mL seawater, pH 5.4, GC-NCI-MS LOQ 0.1 ng/L [2].		on with NaBEt ₄	, hexane ext	raction; LOQ ~	0.8 ng/L	[1].

<u>Liquid phase microextraction (LPME) – GC-MS-MS</u> 4mL water sample; 4-fluorophenyl derivatisation; LOQ 0.36 ng/L [3].

LLE-GC-FDP

1 L sea water, pH 5.5, ethylation in aqueous phase with NaBEt₄, iso-octane/n-pentane extraction; LOQ 0.01 ng/L [4].

LLE-LC-MS

At pH 4 using hexane-ethylacetate (70:30); ion-trap LC-APCI-MS; m/z 323 and 307; LOD 35 µg/L for TBT [5].

GC-ICP-MS

Extraction of 1 L sample at clean room conditions. Derivatisation followed by GC-ICP/MS, LOQ ~0.01 ng/L [6,7]

Comments

Only tributyltin cation is required for WFD monitoring. In the environment different anions (OH-, Cl-, Br-, acetate) are associated with TBT. Analytical methods are assumed to derivatise all forms.

EQS values for TBT refer to the tributyl-cation, hence result shall be expressed in the same way.

Care has to be taken when comparing result with data from scientific literature because some authors express results as μg Sn /L.

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Compound Trichloro						
(all iso		Cl ₃				
CAS Number 12002-48-1	8			Wat	ter Solubi 6-19	lity [mg/L]
AA-EQS	5 [µg/L]	I		MAC-	EQS [µg/	L]
Inland Surface Waters 0.4	Other Su	rface Waters 0.4		rface Waters <i>oplicable</i>		urface Waters not applicable
EN ISO 15680 Water quality - Gas-chrom of a number of monocyclic naphthalene and several ch using purge-and-trap and th <u>Matrix</u> drinking water, gro seawater and (diluted) was <u>Sampling</u> ISO 5667-1, 566 <u>Pretreatment</u>	Available Standard Method EN ISO 15680 Water quality - Gas-chromatographic determination of a number of monocyclic aromatic hydrocarbons, naphthalene and several chlorinated compounds using purge-and-trap and thermal desorption Matrix drinking water, ground water, surface water, seawater and (diluted) waste water Sampling ISO 5667-1, 5667-2 and ISO 5667-3 Pretreatment Storage at 4 °C air tight and no direct sunlight, analysis within 5 days			Method DescriptionEN ISO 15680:2003 specifies a general method for the determination of volatile organic compounds (VOCs) is water by purge-and-trap isolation and gas chromatography (GC).Detection is preferably carried out by mass spectrometri in the electron impact mode (EI), but other detectors m be applied as wellSelected ions: 180, 182, 145Limit of Quantification (LOQ): n_{AP} percental rate of out s_R standard deviation between the laborato CV_R reproducibility variation coeffic $n.a. =$ not avail		
Matrix	l	n _{AP} in %	= x in ng	L $s_R in I$	ng/L	CV_R in %
Drinking water (0.2 μ g/L)	5	n.a.	n.a.	n.a	1.	27
Surface water (0.2 μ g/L)	4	n.a.	n.a.	n.a	1 .	35
Other Analytical Method ISO 6468:1996: Water qua	_			hlorine insectici	des, polvc	hlorinated hinhenvls

In water, trichlorobenzenes are likely to be adsorbed onto sediments and to bioconcentrate in aquatic organisms. Evaporation from water may be a significant removal process.

Compound Trichloro	methane							
CAS Number 67-66-3		Log K 1.97		Water	Solubilit 8.7	ty [mg/L]		
AA-EQS	[µg/L]			MAC-E(QS [µg/L]]		
Inland Surface Waters 2.5	Other Su	rface Waters 2.5		urface Waters applicable		Surface Waters		
Available Standard Meth	od		Method I	Description				
EN ISO 15680: 2003 Gas-chromatographic deten monocyclic aromatic hydro and several chlorinated cor and-trap and thermal desor	ocarbons, na npounds us	aphthalene	the determ (VOCs) in	5680:2003 speci nination of volat n water by purge natography (GC)	ile organi -and-trap			
<u>Matrix</u> drinking water, gro seawater and (diluted) was <u>Sampling</u> <i>ISO</i> 5667-1, 566 <u>Pretreatment</u>	te water		spectrome	is preferably can etry in the electro ectors may be app	on impact	t mode (EI), but		
Storage at 4 °C air tight an analysis within 5 days	d no direct	sunlight,	Limit of	Quantification ((<u>LOQ):</u> 0	0.01 µg/L		
Interlaboratory study Matrix	1	n _{AP} in %	=	g/L S _R in I	na/I	n.a. = not available CV_R in %		
			x in n	5.2	_			
Drinking water (0.2 μ g/L)	15	n.a.	n.a.	n.a		29		
Surface water (0.2 μ g/L)	13	n.a.	n.a.	n.a	a.	30		
 Other Analytical Methods EN ISO 10301:1997 specifies two methods for the determination of highly volatile halogenated hydrocarbons in water using gas chromatography with e.g. electron capture detector after: a) the extraction by an organic solvent or using, b) a head-space method (LOQ: 100 μg/L). The EPA Method 1624 is designed to determine the volatile organic pollutants in water amenable to purge and trap gas chromatography-mass spectrometry. Huybrechts et al. 2003 give a review of gas chromatography-based methods for analysis of volatile organic compounds in estuarine waters with special emphasis on monitoring. [1] 								
<u>Comments</u>								
References[1]T. Huybrechts, J.methods for analywith the river Sch	sis of anthr	opogenic volati	le organic c	compounds in est	tuarine w	aters, illustrated		

Compound Trifluralin				F ₃ C				
CAS Number 1582-0			L	og K _o ~ 5.3	w	Water	Solubility [mg/L] ~ 0.3	
	ля-8 A-EQS [µg	/L]		~ 3.5		MAC-EQ		
Inland Surface Wat			ace Water	rs	Inland S	urface Waters	Other Surface Waters	
0.03		C	0.03		not	applicable	not applicable	
Available Standard Method EN ISO 10695:2000 Determination of selected organic nitrogen and phosphorus compounds - Gas chromatography methods Matrix Drinking waters, ground waters, surface waters and waste waters containing up to 50 mg/L of suspended solids Sampling Pretreatment Storage				of	not applicable not applicable Method Description Liquid/liquid extraction with dichloromethane or liquid/solid extraction (SPE) on reversed-phase (RP)-C18 material or other adsorbent. Elution of the cartridges with e.g. methanol or acetone. After concentration, the sample extracts are analysed by gas chromatography, using a nitrogen-phosphorus or mass spectrometric detector. Limit of Quantification (LOQ): ~ 0.05 µg/L			
Interlaboratory study	y 1993 for l	iquid/so	lid extract	tion		s _R standard	I mean after elimination of outliers deviation between the laboratories eproducibility variation coefficient	
Matrix	l	n _A	_P in %	= x	in µg/L	s _R in µg/L	CV_R in %	
Raw water	14		8.6		0.296	0.0264	46.3	
Naw watch14 0.0 0.200 0.0204 40.3 Other Analytical MethodsGas Chromatography - Mass Spectrometry GC-MS determination of the ions 306, 264, 336 [1-5] LOQ ~ 13 ng/L (SPE extraction of 500 mL water) [1] LOQ ~ 5 ng/L (SPE extraction of 200 mL water) [4] LOQ ~ 1 ng/L (SPE extraction of 500 mL water) [5]EPA method 508.1 (GC-ECD) [6] SPE-GC- triple quadrupole-MS-MS C18-SPE, 100 mL, 306 > 264; LOQ ~ 25 ng/L [7]SPE-GC-NCI-MS C18-SPE, 100 mL, LOQ ~ 25 ng/L [7]Comments								

If released to water, trifluralin is expected to biodegrade under both aerobic and anaerobic conditions and to undergo direct photolytic degradation. It is expected to bioconcentrate in fish and aquatic organisms and adsorb strongly to sediment and suspended organic matter. It may also volatilize from water to the atmosphere. If released to the atmosphere, trifluralin is expected to undergo a rapid gas-phase photolysis.

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Carbontetrachloride	,	Tetrachloroethylene			Trie	Trichloroethylene		
			CI CI				CI CI	
CAS Number			L	og K _{OW}		Water So	lubility [mg/L]	
Carbontetrachloride	56-23-5	5		2.83			1160	
Tetrachloroethylene	127-18-4			3.4			100	
Trichloroethylene	79-01-6			2.42			1100	
		AA-E(QS [µg/]	L]		MAC-EO	QS [µg/L]	
	Inland S Wat			er Surface Waters		d Surface /aters	Other Surface Waters	
Carbontetrachloride Tetrachloroethylene	12 1()		12 10	not a	pplicable	not applicable	
Trichloroethylene	1()		10 Mathad Da	• •			
Available Standard Meth EN ISO 10301:1997	<u>100</u>			Method Des	scription	<u> </u>		
Determination of highly vo hydrocarbons - Gas-chrom <u>Matrix</u> Drinking, ground, <u>Sampling</u>	atographic	atographic methods determination of highly volatile halogenated hydrocarbons in water using gas chromatograph					logenated hromatography with : a) the extraction by	
Pretreatment	Limit of Quantification (LOQ [µg/L]):							
				<u> </u>		Solvent	Headspace	
				Carbon tetra	achloride	Solvent 0.01-0.1	Headspace 0.1	
				Tetrachloro	achloride ethylene	Solvent 0.01-0.1 0.1	Headspace 0.1 0.2	
<u>Storage</u>					achloride ethylene	Solvent 0.01-0.1	Headspace 0.1 0.2 0.2	
<u>Storage</u> <u>Method Validation</u> Interlaboratory study	,		9/	Tetrachloro Trichloroe	achloride ethylene thylene	Solvent $0.01-0.1$ 0.1 $0.05-0.1$ x Total mean astandard deviation CV_R reproduce	Headspace 0.1 0.2 0.2 <i>l</i> Number of laboratoria <i>up</i> percental rate of outlie after elimination of outlie on between the laboratoria ibility variation coefficient	
Storage Method Validation Interlaboratory study Wastewater	1	n _{AP} in	%	Tetrachloro	achloride ethylene thylene	Solvent 0.01-0.1 0.1 0.05-0.1 x Total mean a standard deviation	Headspace 0.1 0.2 0.2 0.2 l Number of laboratoric up percental rate of outlie after elimination of outlie on between the laboratoric	
Storage Method Validation Interlaboratory study Wastewater Solvent Extraction:			%	Tetrachloro Trichloroe = x in ng/L	achloride ethylene thylene	Solvent 0.01-0.1 0.1 0.05-0.1 $= n_A$ X Total mean a standard deviation CV _R reproduce a in $\mu g/L$	Headspace 0.1 0.2 0.2 l Number of laboratoring percental rate of outlie after elimination of outlie after elimination of outlie on between the laboratoring ibility variation coefficie CV_R in %	
Storage Method Validation Interlaboratory study Wastewater Solvent Extraction: Carbon Tetrachloride	18	0	%	Tetrachloro Trichloroe $= \frac{x in ng/L}{76.2}$	achloride ethylene thylene	Solvent $0.01-0.1$ 0.1 $0.05-0.1$ = x Total mean a standard deviation CV _R reproduce x Total mean a standard deviation CV _R reproduce x Total mean a x Total	Headspace 0.1 0.2 0.2 l Number of laboratoring percental rate of outlies after elimination of outlies on between the laboratoring ibility variation coefficies CV_R in % 9.4	
Storage Method Validation Interlaboratory study Wastewater Solvent Extraction: Carbon Tetrachloride Tetrachloroethylene	18 18	0 0	%	$= \frac{x in ng/L}{76.2}$	achloride ethylene thylene	Solvent 0.01-0.1 0.1 0.05-0.1 $=$ n_A = x Total mean a standard deviation CV_R reproduce $a in \mu g/L$ 7.2 6.4	Headspace 0.1 0.2 0.2 l Number of laboratori l_{P} percental rate of outlieafter elimination of outlieon between the laboratoriibility variation coefficie CV_R in % 9.4 7.8	
Storage Method Validation Interlaboratory study Wastewater Solvent Extraction: Carbon Tetrachloride Tetrachloroethylene Trichloroethylene	18	0	%	Tetrachloro Trichloroe $= \frac{x in ng/L}{76.2}$	achloride ethylene thylene	Solvent $0.01-0.1$ 0.1 $0.05-0.1$ = x Total mean a standard deviation CV _R reproduce x Total mean a standard deviation CV _R reproduce x Total mean a x Total mean a 	Headspace 0.1 0.2 0.2 l Number of laboratori I_P percental rate of outlie after elimination of outlie on between the laboratori ibility variation coefficie CV_R in % 9.4	
Storage Method Validation Interlaboratory study Wastewater Solvent Extraction: Carbon Tetrachloride Tetrachloroethylene Trichloroethylene Headspace:	18 18	0 0	%	$= \frac{x in ng/L}{76.2}$	achloride ethylene thylene	Solvent 0.01-0.1 0.1 0.05-0.1 $=$ n_A = x Total mean a standard deviation CV_R reproduce $a in \mu g/L$ 7.2 6.4	Headspace 0.1 0.2 0.2 l Number of laboratori l_{P} percental rate of outlieafter elimination of outlieon between the laboratoriibility variation coefficie CV_R in % 9.4 7.8	
<u>Storage</u> <u>Method Validation</u> <i>Interlaboratory study</i>	18 18 18	0 0 6	%	Tetrachloro Trichloroe $= \frac{x in ng/L}{76.2}$	achloride ethylene thylene	Solvent 0.01-0.1 0.1 0.05-0.1 $= n_A$ X Total mean a standard deviation CV _R reproduct A in $\mu g/L$ 7.2 6.4 7.3	Headspace 0.1 0.2 0.2 l Number of laboratorion μ_P percental rate of outlieafter elimination of outlieon between the laboratorionibility variation coefficie CV_R in %9.47.89.7	

EPA method 524.2 – Purge and Trap concentration with GC/MS analysis [2].

<u>Comments</u> Modern GC/MS instrumentation may allow the use of full scan mass spectra for identification and quantification of these substances at levels below EQS.

References

[1]	http://www.accustandard.com/asi/	/pdfs/epa	methods/502	2.pdf

http://www.accustandard.com/asi/pdfs/epa_methods/524_2.pdf [1]

ANNEX III: Existing certified reference materials¹

Table 1: Reference materials producers

BAM, Germany	http://www.bam.de/
CMI, Czech Republic	http://www.cmi.cz/
EUROFINS, Denmark	http://www.eurofins.dk/
GUM, Poland	http://www.gum.gov.pl/pl/site/
IAEA, Austria	http://www.iaea.org/programmes/aqcs/
IPO, Poland	http://www.ipo.waw.pl/
IRMM, European Commission	http://www.irmm.jrc.be
LGC, GB	http://www.lgcstandards.com/home/home_de.aspx
NIST, USA	http://www.nist.gov/
NRC-CNRC, Canada	http://inms-ienm.nrc-cnrc.gc.ca/
National Institute of Metrology, China	http://www.en.nim.ac.cn/
National Measurement Institute, Australia	http://www.measurement.gov.au/
SMU, Slovakia	http://www.smu.gov.sk/
SMU, Slovakia	http://www.smu.gov.sk/

¹ EAQC-WISE project, funded under the 6th RDT Framework Programme, European Commission

Table 2: Certified reference materials related to the WFD priority substancesP – Pure compounds or solutions

Priority substance	Matrix	CRM-Identifier	Producer	Certified value	Reference
Alachlor	Р	SRM-3070	NIST	24.0 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
Anthracene	Р	DPAC-1	NRC-CNRC, Canada	13 μg/g	http://inms-ienm.nrc-cnrc.gc.ca/calserv/crm_e.html
	Р	SIRM 10-2-30s	Q-chem Ltd., Slovakia	48.7 μg/g	http://www.comar.bam.de/ http://ts.nist.gov/MeasurementServices/ReferenceMaterial
	Р	SRM-1647d	NIST	20.77 μg/mL	s/232.cfm
	Р	SRM-2260a	NIST	3.231 µg/mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
Atrazine	Р	SRM-3070	NIST	39.2 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
Benzene	Р	7141-95M/1	GUP TSIKV	0.99 g/L	http://www.comar.bam.de/
	P P	7141-95M/2 7141-95M/3	GUP TSIKV GUP TSIKV	0.495 g/L 0.097 g/L	http://www.comar.bam.de/ http://www.comar.bam.de/
	Р	7141-95M	GUP TSIKV	99.87 %	http://www.comar.bam.de/
	Р	C03	SMU Tian Jin Institute of Metrological	99.9 %	http://www.comar.bam.de/
	Р	GBW 06104	Technology	99.95 %	http://www.comar.bam.de/
Cadmium and its compounds	P P	6690-93/1 7325-96	GUP TSIKV GUP TSIKV	1 g/L 0.0101 %	http://www.comar.bam.de/ http://www.comar.bam.de/
	Р	DMR-85c	CENAM, Mexico	1001.0 mg/L	http://www.comar.bam.de/
	Р	DMR-8i	CENAM, Mexico National Institute of Metrology,	1 mg/L	http://www.comar.bam.de/
	Р	GBW-08602	China National Institute of Metrology,	0.100 µg/kg	http://www.comar.bam.de/
	Р	GBW-08607	China	0.100 µg/g	http://www.comar.bam.de/
	Р	GBW-08608	National Institute of Metrology, China	10.0 µg/kg	http://www.comar.bam.de/
	Р	JK-0012 Cd 1000	CERI, Japan	0.1 %	http://www.comar.bam.de/
1,2-Dichloroethane	P	7332-96	GUP TSIKV, Russia	99.76 %	http://www.comar.bam.de/ http://ts.nist.gov/MeasurementServices/ReferenceMaterial
	Р	SRM-3012	NIST	0.010039 g/g	s/232.cfm
Endosulfan	Р	SRM-3069	NIST	4.66 mg/kg (I) 5.29 mg/kg (II)	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
	Р	SRM-2275	NIST	2.880 mg/kg (I)	http://ts.nist.gov/MeasurementServices/ReferenceMaterial
				2.943 mg/kg (II)	s/232.cfm http://ts.nist.gov/MeasurementServices/ReferenceMaterial
Hexachlorobenzene	Р	SRM-1492	NIST	308 µg/kg	s/232.cfm
	Р	SRM-2261	NIST	1.968 µg/mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
	Р	SRM-3069	NIST	4.39 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
Hexachlorocyclohexane	Р	SLRM-10-2-08	Slovak Institute of Metrology,	0.1 %	http://www.comar.bam.de/
gamma-isomer			Slovakia NIST		http://ts.nist.gov/MeasurementServices/ReferenceMaterial
	Р	SRM-1492	INIS I	310 µg/kg	s/232.cfm http://ts.nist.gov/MeasurementServices/ReferenceMaterial
	Р	SRM-2261	NIST	1.972 μg/mL	s/232.cfm
	Р	SRM-3069	NIST	4.22 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
	Р	SRM-8466	NIST	99.9 weight %	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
Lead and its compounds	Р	7012-93/1	GUP TSIKV	1.01 g/L	http://www.comar.bam.de/
	P P	DMR-63c DMR-8i	CENAM, Mexico CENAM, Mexico	1002.1 mg/L 10 mg/L	http://www.comar.bam.de/ http://www.comar.bam.de/
Naphthalene	Р	DPAC-1	NRC-CNRC, Canada	116 μg/g	http://inms-ienm.nrc-cnrc.gc.ca/calserv/crm_e.html
	Р	SIRM 10-2-30s	Q-chem Ltd., Slovakia	49.8 μg/g	http://www.comar.bam.de/ http://ts.nist.gov/MeasurementServices/ReferenceMaterial
	Р	SRM-1586-1	NIST	126.5 μg /g	s/232.cfm
	Р	SRM-1586-2	NIST	126.6 µg /g	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
	Р	SRM-1647d	NIST	20.13 µg /mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
	Р	SRM-2270	NIST	77.0 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterial
Nickel and its compounds	Р	DMR-8i	CENAM, Mexico	1 mg/L	s/232.cfm http://www.comar.bam.de/
Pentachlorophenol	Р	SRM-1584	NIST	15.4 μg/mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial
	Р	SRM-3066	NIST	100.7 mg/L	s/232.cfm http://ts.nist.gov/MeasurementServices/ReferenceMaterial
PAHs	Р	DPAC-1	NRC-CNRC, Canada	35 μg/g	s/232.cfm http://inms-ienm.nrc-cnrc.gc.ca/calserv/crm_e.html
benzo[a]pyrene	P	GBW-08701	Beijing Municipal Environmental	5.75 µg/g	http://www.comar.bam.de/
	P		Monitoring Centre, China Beijing Municipal Environmental	10.0 μg/g	1.
	P P	GBW-08702	Monitoring Centre, China		http://www.comar.bam.de/
		SIRM 10-2-30s SRM-1586-1	Q-chem Ltd., Slovakia NIST	49.4 μg/g 49.2 μg/g	http://www.comar.bam.de/ http://ts.nist.gov/MeasurementServices/ReferenceMaterial
	P			49.2 μg /g	s/232.cfm http://ts.nist.gov/MeasurementServices/ReferenceMaterial
	P	SRM-1586-2	NIST	44.1 μg /g	s/232.cfm http://ts.nist.gov/MeasurementServices/ReferenceMaterial
	P	SRM-1647d	NIST	4.91 μg /mL	s/232.cfm http://ts.nist.gov/MeasurementServices/ReferenceMaterial
	Р	SRM-2260a	NIST	4.07 μg/mL	s/232.cfm http://ts.nist.gov/MeasurementServices/ReferenceMaterial
× .	Р	SRM-2270	NIST	37.3 mg/kg	s/232.cfm
fluoranthene	P P	DPAC-1 DPAC-2	NRC-CNRC, Canada NRC-CNRC, Canada	116 μg/g 117 μg/g	http://inms-ienm.nrc-cnrc.gc.ca/calserv/crm_e.html http://inms-ienm.nrc-cnrc.gc.ca/calserv/crm_e.html
	r P	SRM-1647d	NIST	7.64 μg /mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial
	Р	SRM-2260a	NIST		s/232.cfm http://ts.nist.gov/MeasurementServices/ReferenceMaterial
				7.200 μg/mL	s/232.cfm
	Р	SRM-2269	NIST	62.6 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterial

Priority substance	Matrix	CRM-Identifier	Producer	Certified value	Reference
					s/232.cfm
benzo[b]fluoranthene	Р	BCR-048R	EC-JRC-IRMM	0.995 g/g	http://http://www.irmm.jrc.be/html/reference_materials_c atalogue/index.htm/html/reference_materials_catalogue/i ndex.htm
	Р	DPAC-1	NRC-CNRC, Canada	35 μg/g	http://inms-ienm.nrc-cnrc.gc.ca/calserv/crm_e.html
	Р	SRM-1647d	NIST	4.17 μg /mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
	Р	SRM-2260a	NIST	6.80 µg/mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
	Р	SIRM 10-2-30s	Q-chem Ltd., Slovakia	49.9 μg/g	http://www.comar.bam.de/
benzo[k]fluoranthene	Р	BCR-048R	EC-JRC-IRMM	0.995 g/g	http://http://www.irmm.jrc.be/html/reference_materials_c atalogue/index.htm/html/reference_materials_catalogue/i ndex.htm
	Р	SRM-1647d	NIST	4.72 μg /mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
	Р	SRM-2260a	NIST	2.979 μg/mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
	Р	SIRM 10-2-30s	Q-chem Ltd., Slovakia	49.9 μg/g	http://www.comar.bam.de/
indeno[1,2,3-cd]pyrene	Р	SIRM 10-2-30s	Q-chem Ltd., Slovakia	49.5 μg/g	http://www.comar.bam.de/
	Р	SRM-1647d	NIST	4.28 μg /mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
benzo[g,h,i]perylene	Р	BCR-052	EC-JRC-IRMM	0.990 g/g	http://http://www.irmm.jrc.be/html/reference_materials_c atalogue/index.htm/html/reference_materials_catalogue/i ndex.htm
	P P	DPAC-1 SIRM 10-2-30s	NRC-CNRC, Canada Q-chem Ltd., Slovakia	23 μg/g 48.8 μg/g	http://inms-ienm.nrc-cnrc.gc.ca/calserv/crm_e.html http://www.comar.bam.de/
	Р	SRM-1647d	NIST	3.68 µg /mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
	Р	SRM-2260a	NIST	4.904 μg/mL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
	Р	SRM-2270	NIST	35.34 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
Simazine	Р	SRM-3070	NIST	49.4 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm
Trichloromethane (Chloroform)	Р	7288-96	GUP TSIKV, Russia	99.88 %	http://www.comar.bam.de/
		SRM-1639	NIST	6235 ng/µL	http://ts.nist.gov/MeasurementServices/ReferenceMaterial s/232.cfm

Table 3: Certified reference materials related to the WFD priority substancesMatrix: S – Sediment; W – Water; A – Aquatic plant or animal

Priority substance	Matrix	CRM-Identifier	Producer	Certified value	Reference
Anthracene	s	EC-1	National Water Research Institute, Canada	1.2 μg/g	http://www.comar.bam.de
	s	EC-8	National Water Research Institute,	41 µg/kg	http://www.comar.bam.de
	s	IAEA-383	Canada IAEA Analytical Quality Control	0.03 mg/kg	http://www.comar.bam.de
	s	IAEA-408	Services IAEA Analytical Quality Control	0.0098 mg/kg	
			Services IAEA Analytical Quality Control		http://www.comar.bam.de
	s	IAEA-417	Services	630 ng/g	http://www.comar.bam.de
	s s	HS-3B HS-4B	NRC-CNRC, Canada NRC-CNRC, Canada	2.76 μg/g 0.46 μg/g	http://www.comar.bam.de http://www.comar.bam.de
	s	HS-5	NRC-CNRC, Canada	0.38 µg/g	http://www.comar.bam.de
	s	HS-6	NRC-CNRC, Canada	1.1 μg/g	http://www.comar.bam.de
	s	SRM-1941b	NIST	184 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	s	SRM-1944	NIST	1.77 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	SRM-1974b	NIST	0.527 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	А	IAEA-140/OC	IAEA Analytical Quality Control	0.014 mg/kg	http://www.comar.bam.de
	A	IAEA-432	Services IAEA Analytical Quality Control	1.5 ng/g	http://www.comar.bam.de
Benzene	S	PR 9584	Services RIZA QA and Interlaboratory studies	7.62 µg/kg	http://www.comar.bam.de/
benzene		110 7504	KIZA QA and internationatory studies	7.02 µg/kg	http://http://www.iomai.bail.dc/
Cadmium and its compounds	w	BCR-505	EC-JRC-IRMM	0.80 nmol/kg	ce_materials_catalogue/index.htm/html/ref erence_materials_catalogue/index.htm http://http://www.irmm.jrc.be/html/referen
	w	BCR-403	EC-JRC-IRMM	0.175 nmol/kg	ce_materials_catalogue/index.htm/html/ref erence_materials_catalogue/index.htm http://http://www.irmm.jrc.be/html/referen
	w	BCR-609	EC-JRC-IRMM	0.164 µg/kg	ce_materials_catalogue/index.htm/html/ref erence_materials_catalogue/index.htm
	w	BCR-610	EC-JRC-IRMM	2.94 µg/kg	http://http://www.irmm.jrc.be/html/referen ce_materials_catalogue/index.htm/html/ref erence_materials_catalogue/index.htm
	w	BCR-713	EC-JRC-IRMM	5.1 µg/L	http://http://www.irmm.jrc.be/html/referen ce_materials_catalogue/index.htm/html/ref erence_materials_catalogue/index.htm
	w	BCR-714	EC-JRC-IRMM	19.9 µg/L	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	w	BCR-715	EC-JRC-IRMM	40 µg/L	http://www.irmm.jrc.be/html/reference_ma terials catalogue/index.htm
	w	QC LL2	EUROFINS (DK)	2 μg/L	http://www.eurofins.dk
	W W	CASS-4	NRC-CNRC, Canada	0.026 µg/L	http://inms-ienm.nrc-cnrc.gc.ca
	w	LGC-6016 LGC-6017	LGC-Promochem LGC-Promochem	101 μg/kg 0.13 μg/L	http://www.comar.bam.de/ http://www.comar.bam.de/
	W	LGC-6019	LGC-Promochem	0.11 µg/L	http://www.comar.bam.de/
	W W	NASS-5 SLEW-3	NRC-CNRC, Canada NRC-CNRC, Canada	0.023 μg/L 0.048 μg/L	http://inms-ienm.nrc-cnrc.gc.ca http://inms-ienm.nrc-cnrc.gc.ca
	w	SLRM-12-3-10	Research Institute for Irrigation,	0.005 µg/g	http://www.comar.bam.de
	w	SLRS-4	Slovakia NRC-CNRC, Canada	0.012 μg/L	http://inms-ienm.nrc-cnrc.gc.ca
	w	SRM-1640	NIST	22.79 μg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	w	SRM-1643e	NIST	6.408 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	w	TM-23.2	NRC-CNRC, Canada	2.6 µg/kg	http://www.comar.bam.de/
	W W	TM-24 TM-26.2	NRC-CNRC, Canada NRC-CNRC, Canada	12.5 μg/kg	http://www.comar.bam.de/
	w	TM-20.2 TM-27	NRC-CNRC, Canada	6.8 μg/kg 1.0 μg/kg	http://www.comar.bam.de/ http://www.comar.bam.de/
	w	TM-27.2	NRC-CNRC, Canada	1.2 µg/kg	http://www.comar.bam.de/
	W W	TM-28 TMRAIN-51.2	NRC-CNRC, Canada NRC-CNRC, Canada	1.2 μg/kg 25.1 μg/kg	http://www.comar.bam.de/ http://www.comar.bam.de/
	w	TMRAIN-52.2	NRC-CNRC, Canada	91.4 μg/kg	http://www.comar.bam.de/
	W	TMRAIN-53.2	NRC-CNRC, Canada	122 µg/kg	http://www.comar.bam.de/
	W W	TMRAIN-54.2 TMRAIN-95	NRC-CNRC, Canada NRC-CNRC, Canada	185 μg/kg 0.48 μg/kg	http://www.comar.bam.de/ http://www.comar.bam.de/
	S	GBW-07314	NRCCRM, China	0.20 μg/g	http://www.comar.bam.de/
	s	GBW-08301	Institute of Environmental Chemistry, China	2.45 µg/g	http://www.comar.bam.de/
	s s	HISS-1 IAEA-SL-1	NRC-CNRC, Canada IAEA, Austria	0.024 mg/kg 0.26 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	s	MESS-3	NRC-CNRC, Canada	0.24 mg/kg	http://www.comar.bam.de/ http://inms-ienm.nrc-cnrc.gc.ca
	s	NIES-2	NIES, Japan	0.82 µg/g	http://www.comar.bam.de/
	s s	NIES-9 PACS-2	NIES, Japan NRC-CNRC, Canada	0.15 μg/g 2.11 mg/kg	http://www.comar.bam.de/ http://inms-ienm.nrc-cnrc.gc.ca
	s	SRM-1646a	NRC-CNRC, Canada NIST	0.148 mg/kg	http://ts.nist.gov/MeasurementServices/Ref
	s	SRM-1944	NIST	8.8 mg/kg	erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
	s	SRM-1944 SRM-2702	NIST		erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
				0.817 mg/kg	erenceMaterials/232.cfm
	A A	DORM-2 LUTS-1	NRC-CNRC, Canada NRC-CNRC, Canada	0.043 mg/kg 2.12 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca http://inms-ienm.nrc-cnrc.gc.ca
	Α	DOLT-3	NRC-CNRC, Canada	19.4 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	Α	TORT-2	NRC-CNRC, Canada Institute of Environmental Chemistry,	26.7 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	Α	GBW-08571	China	4.5 μg/g	http://www.comar.bam.de/
	Α	LGC-7160	LGC-Promochem	1.85 mg/kg	http://www.comar.bam.de/

Priority substance	Matrix	CRM-Identifier	Producer	Certified value	Reference
	Α	SLRM-12-2-02	Institute of Radioecology and Applied Nuclear Techniques, Slovakia	44.8 µg/kg	http://www.comar.bam.de/
	А	SRM-1566b	NIST	2.48 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	SRM-2976	NIST	0.179 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	SRM-2977	NIST	0.82 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	ERM-CE278	EC-JRC-IRMM	0.348 mg/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	Α	BCR-279	EC-JRC-IRMM	0.274 mg/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	Α	BCR-060	EC-JRC-IRMM	2.20 mg/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	Α	BCR-414	EC-JRC-IRMM	0.383 mg/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm http://www.irmm.jrc.be/html/reference_ma
	A	BCR-422	EC-JRC-IRMM National Water Research Institute,	0.017 mg/kg	terials_catalogue/index.htm
Hexachlorobenzene	s	EC-2	Canada National Water Research Institute,	200.6 µg/kg	http://www.comar.bam.de
	s	EC-3	Canada National Water Research Institute,	279 µg/g	http://www.comar.bam.de
	s	EC-8	Canada	98 μg/kg	http://www.comar.bam.de
	s	SRM-1941b	NIST	5.83 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	s	SRM-1944	NIST	6.03 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	SRM-1945	NIST	32.9 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	А	SRM-1946	NIST	7.25 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	А	BCR-598	EC-JRC-IRMM	55.7 µg/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	Α	IAEA-432	IAEA Analytical Quality Control Services	0.2 ng/g	http://www.comar.bam.de
	A	SRM-1588	NIST	157.8 μg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
Hexachlorobutadiene	s	EC-2	National Water Research Institute, Canada National Water Research Institute,	21.3 µg/kg	http://www.comar.bam.de
	s	EC-3	Canada National Water Research Institute,	61 µg/g	http://www.comar.bam.de
	s	EC-8	Canada	21 µg/kg	http://www.comar.bam.de http://ts.nist.gov/MeasurementServices/Ref
Hexachlorocyclohexane	A	SRM-1945	NIST IAEA Analytical Quality Control	32.9 μg /kg	erenceMaterials/232.cfm
gamma-isomer	S	IAEA-408	Services IAEA Analytical Quality Control	0.00019 mg/kg	http://www.comar.bam.de
	S	IAEA-417	Services	0.54 ng/g	http://www.comar.bam.de http://www.irmm.jrc.be/html/reference ma
	A	BCR-598	EC-JRC-IRMM	23.0 µg/kg	terials_catalogue/index.htm http://ts.nist.gov/MeasurementServices/Ref
	A	SRM-1588 SRM-1945	NIST NIST	24.9 μg/kg	erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
	A A	SRM-1945	NIST	3.30 μg/kg 1.14 μg/kg	erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
Lead and its compounds	W	7272-96	GUP TSIKV	0.103 g/L	erenceMaterials/232.cfm http://www.comar.bam.de/
Licula and its compounds	w	BCR-403	EC-JRC-IRMM	0.117 nmol/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	w	BCR-609	EC-JRC-IRMM	1.63 µg/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	w	BCR-610	EC-JRC-IRMM	7.78 µg/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	w	BCR-713	EC-JRC-IRMM	47 µg/L	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	w	BCR-714	EC-JRC-IRMM	145 µg/L	http://www.irmm.jrc.be/html/reference_ma terials catalogue/index.htm
	w	BCR-715	EC-JRC-IRMM	0.49 µg/L	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	W W	QC LL2 CASS-4	EUROFINS (DK) NRC-CNRC, Canada	20 μg/L 0.0098 μg/L	http://www.eurofins.dk http://inms-ienm.nrc-cnrc.gc.ca
	w	NASS-5	NRC-CNRC, Canada	0.008 µg/L	http://inms-ienm.nrc-cnrc.gc.ca
	W	SLEW-3	NRC-CNRC, Canada	0.0090 µg/L	http://inms-ienm.nrc-cnrc.gc.ca
	W W	SLRS-4 GBW-08601	NRC-CNRC, Canada NRCCRM, China	0.086 μg/L 1.0 μg/g	http://inms-ienm.nrc-cnrc.gc.ca http://www.comar.bam.de/
	W	GBW-08607	National Institute of Metrology, China	1.0 µg/g	http://www.comar.bam.de/
	W W	GBW-08608 LGC-6016	National Institute of Metrology, China LGC-Promochem	50 μg/kg 196 μg/kg	http://www.comar.bam.de/ http://www.comar.bam.de/
	W	LGC-6017	LGC-Promochem	1.0 µg/L	http://www.comar.bam.de/
	W	LGC-6019	LGC-Promochem	5.2 µg/L	http://www.comar.bam.de/
	W	ERML-CA010a	LGC Research Institute for Irrigation,	95 mg/L	http://www.lgcpromochem.com
	w w	SIRM 12-3-10 SLRM-12-3-10	Slovakia Research Institute for Irrigation,	0.029 μg/g 0.029 μg/g	http://www.comar.bam.de/
	w	SRM-1640	Slovakia NIST	22.79 μg/kg	http://ts.nist.gov/MeasurementServices/Ref
	w	SRM-1643e	NIST	19.45 μg/kg	erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
	w	TM-23.2	NRC-CNRC, Canada	3.8 µg/kg	erenceMaterials/232.cfm http://www.comar.bam.de/
	W	TM-24	NRC-CNRC, Canada	7.3 µg/kg	http://www.comar.bam.de/
	W W	TM-26.2 TM-27	NRC-CNRC, Canada NRC-CNRC, Canada	9.9 µg/kg 4.9 µg/kg	http://www.comar.bam.de/ http://www.comar.bam.de/
	W	TM-27.2	NRC-CNRC, Canada	3.2 µg/kg	http://www.comar.bam.de/
	W W	TM-28 TMRAIN-51.2	NRC-CNRC, Canada NRC-CNRC, Canada	3.0 µg/kg 72.9 µg/kg	http://www.comar.bam.de/ http://www.comar.bam.de/
	w	TMRAIN-51.2 TMRAIN-52.2	NRC-CNRC, Canada NRC-CNRC, Canada	72.9 μg/kg 368 μg/kg	http://www.comar.bam.de/
		10000-02.2	mice Critice, Callaua	200 hB/kB	http://www.comar.bam.uc/

Priority substance	Matrix	CRM-Identifier	Producer	Certified value	Reference
	W	TMRAIN-53.2	NRC-CNRC, Canada	360 µg/kg	http://www.comar.bam.de/
	W W	TMRAIN-54.2 TMRAIN-95	NRC-CNRC, Canada NRC-CNRC, Canada	531 μg/kg 0.29 μg/kg	http://www.comar.bam.de/ http://www.comar.bam.de/
	s	GBW-07314	NRCCRM, China	25 μg/g	http://www.comar.bam.de/
	s	GBW-08301	Institute of Environmental Chemistry,	79 μg/g	http://www.comar.bam.de/
	s	HISS-1	China NRC-CNRC, Canada	3.13 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	S	MESS-3	NRC-CNRC, Canada	21.1 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	s s	PACS-2 IAEA-SL-1	NRC-CNRC, Canada IAEA, Austria	183 mg/kg 37.7 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca http://www.comar.bam.de/
	S	NIES-2	NIES, Japan	105 µg/g	http://www.comar.bam.de/
	s	NIES-9	NIES, Japan	1.35 µg/g	http://www.comar.bam.de/
	s	SRM-1646a	NIST	11.7 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
	s	SRM-1944	NIST	330 mg/kg	erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
	S A	SRM-2702 BCR-060	NIST EC-JRC-IRMM	132.8 mg/kg	erenceMaterials/232.cfm http://www.irmm.jrc.be/html/reference_ma
	A	BCR-000	EC-JRC-IRMM	63.8 mg/kg 13.48 mg/kg	terials_catalogue/index.htm http://www.irmm.jrc.be/html/reference_ma
	A	BCR-414	EC-JRC-IRMM	3.97 mg/kg	terials_catalogue/index.htm http://www.irmm.jrc.be/html/reference_ma
	А	BCR-422	EC-JRC-IRMM	0.085 mg/kg	terials_catalogue/index.htm http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	А	ERM-CE278	EC-JRC-IRMM	2.00 mg/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	Α	DOLT-3	NRC-CNRC, Canada	0.319 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	A A	DORM-2 GBW-08571	NRC-CNRC, Canada Institute of Environmental Chemistry,	0.065 mg/kg 1.96 μg/g	http://inms-ienm.nrc-cnrc.gc.ca http://www.comar.bam.de/
	A	SLRM-12-2-02	China Institute of Radioecology and Applied	1.23 μg/g	http://www.comar.bam.de/
	A	SRM-1566b	Nuclear Techniques, Slovakia NIST	0.308 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	А	SRM-2976	NIST	1.19 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	А	SRM-2977	NIST	2.27 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	LUTS-1	NRC-CNRC, Canada	0.010 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
Mercury and its compounds	A W	TORT-2 8004-93/1	NRC-CNRC, Canada GUP TSIKV	0.35 mg/kg 1.01 g/L	http://inms-ienm.nrc-cnrc.gc.ca http://www.comar.bam.de/
stereury and its compounds	w	BCR-579	EC-JRC-IRMM	1.85 ng/kg	http://www.irmm.jrc.be/html/reference_ma
	w	ORMS-3	NRC-CNRC, Canada		terials_catalogue/index.htm
	w	SIRM 12-3-10	Research Institute for Irrigation, Slovakia	12.6 pg/g 0.0011 μg/g	http://inms-ienm.nrc-cnrc.gc.ca http://www.comar.bam.de/
	w	SRM-1641d	NIST	1.590 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	w	QC LL3	EUROFINS (DK)	5 µg/L	http://www.eurofins.dk
	w	QC LL3A	EUROFINS (DK)	0.5 μg/L	http://www.eurofins.dk http://www.irmm.jrc.be/html/reference_ma
	s	ERM-CC580	EC-JRC-IRMM	132 mg/kg	terials_catalogue/index.htm
	s	GBW-07314	NRCCRM, China Institute of Environmental Chemistry,	0.20 µg/g	http://www.comar.bam.de/
	s	GBW-08301	China	0.048 µg/g	http://www.comar.bam.de/
	S	MESS-3	NRC-CNRC, Canada	0.091 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	s s	PACS-2 SRM-2702	NRC-CNRC, Canada NIST	3.04 mg/kg 0.4474 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca http://ts.nist.gov/MeasurementServices/Ref
	s	WQB-1	NRC-CNRC, Canada	1.09 µg/g	erenceMaterials/232.cfm http://www.comar.bam.de
	s	WQB-3	NRC-CNRC, Canada	2.75 µg/g	http://www.comar.bam.de http://www.irmm.jrc.be/html/reference_ma
	Α	BCR-060	EC-JRC-IRMM	0.34 mg/kg	terials_catalogue/index.htm http://www.irmm.jrc.be/html/reference_ma
	A	BCR-414	EC-JRC-IRMM	0.276 mg/kg	terials_catalogue/index.htm http://www.irmm.jrc.be/html/reference_ma
	A	BCR-422	EC-JRC-IRMM	0.559 mg/kg	terials_catalogue/index.htm http://www.irmm.jrc.be/html/reference_ma
	A	BCR-463 DOLT-3	EC-JRC-IRMM	2.85 mg/kg	terials_catalogue/index.htm http://inms-ienm.nrc-cnrc.gc.ca
	A A	DORM-2	NRC-CNRC, Canada NRC-CNRC, Canada	3.37 mg/kg 4.64 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	Α	ERM-CE278	EC-JRC-IRMM	0.196 mg/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	А	ERM-CE464	EC-JRC-IRMM	5.24 mg/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	Α	GBW-08571	Institute of Environmental Chemistry, China	0.067 µg/g	http://www.comar.bam.de/
	Α	LGC-7160	LGC-Promochem	0.096 mg/kg	http://www.comar.bam.de/
	Α	SRM-1566b	NIST	0.0371 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm http://tc.nist.gov/MagsurementServices/Ref
	A	SRM-1946	NIST	0.433 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
	A	SRM-1974b	NIST	17.0 μg/kg	erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
	A A	SRM-2976 TORT-2	NIST NRC-CNRC, Canada	61.0 μg/kg 0.27 mg/kg	erenceMaterials/232.cfm http://inms-ienm.nrc-cnrc.gc.ca
Naphthalene	s	EC-8	National Water Research Institute,	10 μg/kg	http://www.comar.bam.de
- apiniment	s	IAEA-383	Canada IAEA Analytical Quality Control	0.096 mg/kg	http://www.comar.bam.de
	s	IAEA-408	Services IAEA Analytical Quality Control Services	0.027 mg/kg	http://www.comar.bam.de
	s	HS-3B	NRC-CNRC, Canada	2.14 μg/g	http://www.comar.bam.de
	S	HS-4B	NRC-CNRC, Canada	0.22 µg/g	http://www.comar.bam.de
	s s	HS-5 HS-6	NRC-CNRC, Canada NRC-CNRC, Canada	0.25 μg/g 4.1 μg/g	http://www.comar.bam.de http://www.comar.bam.de
	š	SRM-1941b	NIST	848 μg/kg	http://ts.nist.gov/MeasurementServices/Ref

Priority substance	Matrix	CRM-Identifier	Producer	Certified value	Reference
	s	SRM-1944	NIST	1.65 mg/kg	erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
	А	IAEA-140/OC	IAEA Analytical Quality Control Services	0.017 mg/kg	erenceMaterials/232.cfm http://www.comar.bam.de
	А	SRM-1974b	NIST	2.43 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
Nickel and its compounds	W	7272-96	GUP TSIKV, Russia	0.102 g/L	http://www.comar.bam.de/
	w	8001-93/1	GUP TSIKV, Russia Research Institute for Irritotion	1.00 g/L	http://www.comar.bam.de/
	w	SIRM 12-3-10	Research Institute for Irrigation, Slovakia	0.061 µg/g	http://www.comar.bam.de/
	w	BCR-403	EC-JRC-IRMM	4.4 nmol/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	w	BCR-505	EC-JRC-IRMM	24.1 nmol/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	w	BCR-713	EC-JRC-IRMM	30 µg/L	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	w	BCR-714	EC-JRC-IRMM	108 µg/L	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	w	BCR-715	EC-JRC-IRMM	1.20 µg/L	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	w	QC LL1	EUROFINS (DK)	15 μg/L	http://www.eurofins.dk
	w	SRM-1643e	NIST	60.89 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	w	GBW-08607	National Institute of Metrology, China	0.500 µg/g	http://www.comar.bam.de/
	W W	GBW-08608 LGC-6016	National Institute of Metrology, China LGC-Promochem	60 μg/kg 186 μg/kg	http://www.comar.bam.de/ http://www.comar.bam.de/
	w	LGC-6017	LGC-Promochem	1.6 μg/L	http://www.comar.bam.de/
	W	LGC-6019	LGC-Promochem	2.6 µg/L	http://www.comar.bam.de/
	W	ERML-CA010a	LGC	48 mg/L	http://www.lgcpromochem.com
	W W	CASS-4 SLEW-3	NRC-CNRC, Canada NRC-CNRC, Canada	0.314 μg/L 1.23 μg/L	http://inms-ienm.nrc-cnrc.gc.ca http://inms-ienm.nrc-cnrc.gc.ca
	w	SLRS-4	NRC-CNRC, Canada	0.67 µg/L	http://inms-ienm.nrc-cnrc.gc.ca
	w	NASS-5	NRC-CNRC, Canada	0.253 µg/L	http://inms-ienm.nrc-cnrc.gc.ca
	W	TM-23.2	NRC-CNRC, Canada	5.3 µg/kg	http://www.comar.bam.de/
	W W	TM-24 TM-26.2	NRC-CNRC, Canada NRC-CNRC, Canada	3.5 µg/kg 9.9 µg/kg	http://www.comar.bam.de/ http://www.comar.bam.de/
	w	TM-27	NRC-CNRC, Canada	2.7 µg/kg	http://www.comar.bam.de/
	w	TM-27.2	NRC-CNRC, Canada	2.5 µg/kg	http://www.comar.bam.de/
	W	TM-28	NRC-CNRC, Canada	19.3 μg/kg	http://www.comar.bam.de/
	W W	TMRAIN-51.2 TMRAIN-52.2	NRC-CNRC, Canada NRC-CNRC, Canada	66.7 μg/kg 268 μg/kg	http://www.comar.bam.de/ http://www.comar.bam.de/
	w	TMRAIN-53.2	NRC-CNRC, Canada	319 µg/kg	http://www.comar.bam.de/
	W	TMRAIN-54.2	NRC-CNRC, Canada	325 µg/kg	http://www.comar.bam.de/
	w s	TMRAIN-95 SRM-1944	NRC-CNRC, Canada NIST	0.80 μg/kg 76.1 mg/kg	http://www.comar.bam.de/ http://ts.nist.gov/MeasurementServices/Ref
	s	SRM-2702	NIST	75.4 mg/kg	erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
	s	GBW-07314	NRCCRM, China	34.3 μg/g	erenceMaterials/232.cfm http://www.comar.bam.de/
	S	HISS-1	NRC-CNRC, Canada	2.16 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	S S	IAEA-SL-1 MESS-3	IAEA, Austria NRC-CNRC, Canada	44.9 mg/kg 46.9 mg/kg	http://www.comar.bam.de/ http://inms-ienm.nrc-cnrc.gc.ca
	Š	NIES-2	NIES, Japan	40 µg/g	http://www.comar.bam.de/
	S	PACS-2	NRC-CNRC, Canada	39.5 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	S A	WQB-3 DOLT-3	NRC-CNRC, Canada NRC-CNRC, Canada	52.0 μg/g 2.72 mg/kg	http://www.comar.bam.de http://inms-ienm.nrc-cnrc.gc.ca
	A	DORM-2	NRC-CNRC, Canada	19.4 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	Α	LUTS-1	NRC-CNRC, Canada	0.2 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	Α	TORT-2	NRC-CNRC, Canada Institute of Environmental Chemistry,	2.50 mg/kg	http://inms-ienm.nrc-cnrc.gc.ca
	A	GBW-08571	China	1.03 µg/g	http://www.comar.bam.de/ http://www.irmm.jrc.be/html/reference ma
	A	BCR-414	EC-JRC-IRMM	18.8 mg/kg	terials_catalogue/index.htm http://ts.nist.gov/MeasurementServices/Ref
	A A	SRM-2977 LGC-7160	NIST LGC-Promochem	6.06 mg/kg 0.23 mg/kg	erenceMaterials/232.cfm http://www.comar.bam.de/
Pentachlorobenzene	S	EC-2	National Water Research Institute,	48.6 μg/kg	http://www.comar.bam.de
r chuchor obchieche	s	EC-3	Canada National Water Research Institute,	65 µg/g	http://www.comar.bam.de
	s	EC-8	Canada National Water Research Institute,	30 μg/kg	http://www.comar.bam.de
Pentachlorophenol	s	BCR-530	Canada EC-JRC-IRMM	0.47 mg/kg	http://www.irmm.jrc.be/html/reference_ma
PAHs	s	BCR-535	EC-JRC-IRMM	1.16 mg/kg	terials_catalogue/index.htm http://www.irmm.jrc.be/html/reference_ma
benzo[a]pyrene	s	EC-1	National Water Research Institute,	5.3 μg/g	terials_catalogue/index.htm http://www.comar.bam.de
	s	EC-2	Canada National Water Research Institute, Canada	1.21 μg/g	http://www.comar.bam.de
	s	EC-3	National Water Research Institute, Canada	386 µg/g	http://www.comar.bam.de
	s	EC-5	National Water Research Institute, Canada	449 µg/kg	http://www.comar.bam.de/
	s	EC-8	National Water Research Institute, Canada	207 µg/kg	http://www.comar.bam.de
	s	IAEA-383	IAEA Analytical Quality Control Services	0.12 mg/kg	http://www.comar.bam.de
	s	IAEA-408	IAEA Analytical Quality Control Services	0.048 mg/kg	http://www.comar.bam.de
	s	IAEA-417	IAEA Analytical Quality Control Services	2800 ng/g	http://www.comar.bam.de
	S	HS-3B	NRC-CNRC, Canada	5.80 µg/g	http://www.comar.bam.de
	S	HS-4B HS-5	NRC-CNRC, Canada	1.55 μg/g	http://www.comar.bam.de
	S S	HS-5 HS-6	NRC-CNRC, Canada NRC-CNRC, Canada	1.7 μg/g 2.2 μg/g	http://www.comar.bam.de http://www.comar.bam.de
	s	SRM-1941b	NIST	2:2 µg/kg 358 µg/kg	http://ts.nist.gov/MeasurementServices/Ref
				rob	erenceMaterials/232.cfm

Priority substance	Matrix	CRM-Identifier	Producer	Certified value	Reference
	s	SRM-1944	NIST	4.30 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	IAEA-140/OC	IAEA Analytical Quality Control Services	0.02 mg/kg	http://www.comar.bam.de
	А	IAEA-432	IAEA Analytical Quality Control Services	0.9 ng/g	http://www.comar.bam.de
	А	SRM-2977	NIST	8.35 μg/g	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	SRM-1974b	NIST	2.80 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
fluoranthene	s	EC-1	National Water Research Institute, Canada	23.2 µg/g	http://www.comar.bam.de
	S	EC-2	National Water Research Institute, Canada	3.55 µg/g	http://www.comar.bam.de
	s	EC-3	National Water Research Institute, Canada National Water Research Institute,	558 µg/g	http://www.comar.bam.de
	s s	EC-8 HS-3B	Canada NRC-CNRC, Canada	462 μg/kg 25.33 μg/g	http://www.comar.bam.de http://www.comar.bam.de
	s	HS-4B	NRC-CNRC, Canada	3.33 μg/g	http://www.comar.bam.de
	S	HS-5	NRC-CNRC, Canada	8.4 μg/g	http://www.comar.bam.de
	S	HS-6	NRC-CNRC, Canada IAEA Analytical Quality Control	3.54 μg/g	http://www.comar.bam.de
	s s	IAEA-383 IAEA-408	Services IAEA Analytical Quality Control	0.29 mg/kg 0.084 mg/kg	http://www.comar.bam.de http://www.comar.bam.de
	s	SRM-1941b	Services NIST	651 μg/kg	http://ts.nist.gov/MeasurementServices/Ref
	s	SRM-1944	NIST	8.92 mg/kg	erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
	A	IAEA-140/OC	IAEA Analytical Quality Control	0.088 mg/kg	erenceMaterials/232.cfm http://www.comar.bam.de
	A	IAEA-432	Services IAEA Analytical Quality Control	12 ng/g	http://www.comar.bam.de
	А	SRM-1974b	Services NIST	17.1 µg/kg	http://ts.nist.gov/MeasurementServices/Ref
	А	SRM-2977	NIST	38.7 µg/g	erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
benzo[b]fluoranthene	s	BCR-535	EC-JRC-IRMM	2.29 mg/kg	erenceMaterials/232.cfm http://www.irmm.jrc.be/html/reference_ma
{-}	s	IAEA-383	IAEA Analytical Quality Control	0.15 mg/kg	terials_catalogue/index.htm http://www.comar.bam.de
	s	IAEA-408	Services IAEA Analytical Quality Control Services	0.046 mg/kg	http://www.comar.bam.de
	s	IAEA-417	IAEA Analytical Quality Control Services	4100 ng/g	http://www.comar.bam.de
	s	SRM-1941b	NIST	453 μg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	s	SRM-1944	NIST	3.87 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	s	EC-8	National Water Research Institute, Canada	208 µg/kg	http://www.comar.bam.de
	s	EC-1	National Water Research Institute, Canada	7.9 µg/g	http://www.comar.bam.de
	s	EC-2	National Water Research Institute, Canada	2.48 µg/g	http://www.comar.bam.de
	S	HS-5	NRC-CNRC, Canada	2.0 μg/g	http://www.comar.bam.de
	s	HS-6	NRC-CNRC, Canada IAEA Analytical Quality Control	2.8 μg/g	http://www.comar.bam.de
	A	IAEA-432 SRM-1974b	Services NIST	4.8 ng/g	http://www.comar.bam.de http://ts.nist.gov/MeasurementServices/Ref
	A A	SRM-19740 SRM-2977	NIST	6.46 μg/kg	erenceMaterials/232.cfm http://ts.nist.gov/MeasurementServices/Ref
benzo[k]fluoranthene	S	BCR-535	EC-JRC-IRMM	11.01 μg/g 1.09 mg/kg	erenceMaterials/232.cfm http://www.irmm.jrc.be/html/reference_ma
oonzojajnuotunuone	s	IAEA-383	IAEA Analytical Quality Control	0.073 mg/kg	terials_catalogue/index.htm http://www.comar.bam.de
	s	IAEA-408	Services IAEA Analytical Quality Control	0.046 mg/kg	http://www.comar.bam.de
	s	IAEA-417	Services IAEA Analytical Quality Control	2000 ng/g	http://www.comar.bam.de
	s	SRM-1941b	Services NIST	225 μg/kg	http://ts.nist.gov/MeasurementServices/Ref
	s	EC-8	National Water Research Institute, Canada	294 μg/kg	erenceMaterials/232.cfm http://www.comar.bam.de
	s	EC-1	Canada National Water Research Institute, Canada	4.4 μg/g	http://www.comar.bam.de
	s	EC-2	Canada National Water Research Institute, Canada	1.93 μg/g	http://www.comar.bam.de
	s	HS-5	NRC-CNRC, Canada	1.0 µg/g	http://www.comar.bam.de
	s s	HS-6 SRM-1944	NRC-CNRC, Canada NIST	1.43 μg/g 2.30 mg/kg	http://www.comar.bam.de http://ts.nist.gov/MeasurementServices/Ref
	A	IAEA-432	IAEA Analytical Quality Control	2.50 mg/kg 1.9 ng/g	erenceMaterials/232.cfm http://www.comar.bam.de
	A	SRM-1974b	Services NIST	3.16 μg/kg	http://ts.nist.gov/MeasurementServices/Ref
indeno[1,2,3-cd]pyrene	s	BCR-535	EC-JRC-IRMM	1.56 mg/kg	erenceMaterials/232.cfm http://www.irmm.jrc.be/html/reference_ma
C A Market Street	s	EC-1	National Water Research Institute,	5.7 μg/g	terials_catalogue/index.htm http://www.comar.bam.de
	s	EC-2	Canada National Water Research Institute,	1.55 µg/g	http://www.comar.bam.de
	s	EC-8	Canada National Water Research Institute,	34 μg/kg	http://www.comar.bam.de
	s	IAEA-417	Canada IAEA Analytical Quality Control	2700 ng/g	http://www.comar.bam.de
	S	HS-5	Services NRC-CNRC, Canada	1.3 µg/g	http://www.comar.bam.de

Priority substance	Matrix	CRM-Identifier	Producer	Certified value	Reference
	S	HS-6	NRC-CNRC, Canada	1.95 μg/g	http://www.comar.bam.de
	s	SRM-1941b	NIST	341 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	s	SRM-1944	NIST	2.78 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	А	IAEA-140/OC	IAEA Analytical Quality Control Services	0.033 mg/kg	http://www.comar.bam.de
	А	SRM-1974b	NIST	2.14 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	SRM-2977	NIST	4.84 µg/g	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
benzo[<i>g,h,i</i>]perylene	s	EC-1	National Water Research Institute, Canada	4.9 µg/g	http://www.comar.bam.de
	s	EC-2	National Water Research Institute, Canada	1.47 µg/g	http://www.comar.bam.de
	s	EC-8	National Water Research Institute, Canada	176 µg/kg	http://www.comar.bam.de
	s	IAEA-383	IAEA Analytical Quality Control Services	0.19 mg/kg	http://www.comar.bam.de
	s	IAEA-408	IAEA Analytical Quality Control Services	0.038 mg/kg	http://www.comar.bam.de
	s	IAEA-417	IAEA Analytical Quality Control Services	2300 ng/g	http://www.comar.bam.de
	S	HS-3B	NRC-CNRC, Canada	3.88 µg/g	http://www.comar.bam.de
	S	HS-4B	NRC-CNRC, Canada	1.23 μg/g	http://www.comar.bam.de
	S	HS-5	NRC-CNRC, Canada	1.3 μg/g	http://www.comar.bam.de
	S	HS-6	NRC-CNRC, Canada	1.78 μg/g	http://www.comar.bam.de
	s	SRM-1941b	NIST	307 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	s	SRM-1944	NIST	2.84 mg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	SRM-1974b	NIST	3.12 µg/kg	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	SRM-2977	NIST	9.53 μg/g	http://ts.nist.gov/MeasurementServices/Ref erenceMaterials/232.cfm
	Α	IAEA-140/OC	IAEA Analytical Quality Control Services	0.02 mg/kg	http://www.comar.bam.de
Tributyltin compounds	s	BCR-462	EC-JRC-IRMM	54 ug/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	s	BCR-646	EC-JRC-IRMM	480 µg/kg	http://www.irmm.jrc.be/html/reference_ma terials_catalogue/index.htm
	S	HIPA-1	NRC-CNRC, Canada	78 ng/g	http://inms-ienm.nrc-cnrc.gc.ca
	S	PACS-1	NRC-CNRC, Canada	890 ng/g	http://inms-ienm.nrc-cnrc.gc.ca
	S	PACS-2	NRC-CNRC, Canada	0.890 mg/kg (Tri-)	http://inms-ienm.nrc-cnrc.gc.ca
	s	SOPH-1	NRC-CNRC, Canada	125 ng/g	http://inms-ienm.nrc-cnrc.gc.ca
T-d-hlh	Α	NIES-11	NIES, Japan	1.3 μg/g	http://www.comar.bam.de/
Trichlorobenzenes (1,2,4-trichlorobenzene)	s	EC-2	National Water Research Institute, Canada	80.7 µg/kg	http://www.comar.bam.de
	s	EC-8	National Water Research Institute, Canada	67 µg/kg	http://www.comar.bam.de

Table 4: Certified reference materials related to other pollutantsP - Pure compounds or solutions

Priority substance	Matrix	CRM-Identifier	Producer	Certified value	Reference
DDT p,p'-DDT	Р	SRM 1492	NIST	302 µg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	Р	SRM 2261	NIST	3.004 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	Р	SRM 2273	NIST	2.862 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	Р	SRM 2275	NIST		http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
Dieldrin	Р	SRM 1492	NIST	307 µg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	Р	SRM 2261	NIST	3.012 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
Endrin	Р	SRM 2275	NIST	2.908 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
Carbontetrachloride	Р	SRM 3006	NIST	0.010099 g/g	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
Tetrachloroethylene	Р	SRM 3010	NIST	0.009772 g/g	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm

Table 5: Certified reference materials related to other pollutantsMatrix: S – Sediment; W – Water; A – Aquatic plant or animal

Priority substance	Matrix	CRM-Identifier	Producer	Certified value	Reference
DDT p,p'-DDT	s	IAEA-408	IAEA Analytical Quality Control Services	0.0014 mg/kg	http://www.comar.bam.de
	s	SRM-1944	NIST	199 µg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	Α	SRM-1974b	NIST	3.91 µg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	А	IAEA-140/OC	IAEA Analytical Quality Control Services	0.0022 mg/kg	http://www.comar.bam.de
	Α	SRM-1945	NIST	245 µg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	Α	SRM-1946	NIST	37.2 µg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	А	SRM-1588b	NIST	570 µg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	Α	SRM-2977	NIST	1.28 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	Α	BCR-598	EC-JRC-IRMM	179 µg/kg	http://www.irmm.jrc.be/html/reference_materials_catalogue/index.htm
Dieldrin	s	IAEA-408	IAEA Analytical Quality Control Services	0.0003 mg/kg	http://www.comar.bam.de
	Α	IAEA-140/OC	IAEA Analytical Quality Control Services	0.0017 mg/kg	http://www.comar.bam.de
	Α	SRM-1588b	NIST	156 µg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	Α	SRM-2977	NIST	6.04 mg/kg	http://ts.nist.gov/MeasurementServices/ReferenceMaterials/232.cfm
	Α	BCR-598	EC-JRC-IRMM	59 μg/kg	http://www.irmm.jrc.be/html/reference_materials_catalogue/index.htm

ANNEX IV: Case Studies

Background information

Title/Name of case study:

PESTICIDES IN SURFACE WATER BODIES FROM AGRICULTURAL SOURCES (PESTICIDE PROGRAM).

Type of case study:

Monitoring to check the chemical and ecological status compliance (operational and investigative monitoring).

Reporting Institution:

Ministry of the Environment (Spain).

Web-Link: <u>www.mma.es</u>

Main sources for further information; literature:

"Analysis of pesticides and metabolites in Spanish surface waters by isotope dilution gas chromatography/mass spectrometry with previous automated solid-phase extraction. Estimation of the uncertainty of the analytical results".

Planas et al. Journal of Chromatography A, 1131 (2006) 242-252.

Objective of case study - background information:

Development of the methodology for monitoring the pollution caused by pesticides from agricultural sources.

Analysing pesticides is necessary to check the good chemical and ecological status compliance. The pesticides included in the monitoring program must be all the priority substances discharged and "other pesticides" discharged in significant quantities. It is not easy to select the "other pesticides" to analyse due to high number of possible compounds, changes in pesticides use, pesticide fate, etc.

To solve this, the surveillance of pesticide pollution from agricultural activities may combine 2 types of analytical methodologies. Type 1: Standardized techniques with high level of QA/QC to monitor EQS compliance (legally binding EQSs or calculated EQSs). Type 2: MS characterization to determine new pesticides not included in common lists in order to incorporate them in the selected compounds to monitor in the future.

This combined methodology is used to monitor pesticides in water bodies potentially at risk of failing to meet the pesticides EQS due to pressure from agricultural sources.

Contribution to...

Specific contribution linked to WFD monitoring programmes

Operational and investigative monitoring design.

Description

Monitoring points: water bodies potentially at risk of pesticides pollution from agricultural sources.

Matrix: Water

Frequency: 8 sampling/year

Methodology and substances

Using Isotope dilution GC/MS with previous automated SPE

- 10 pesticides from Priority Substances List,
- 04 pesticides from List II with national legally binding EQSs
- 05 metabolite pesticides
- 13 pesticides commonly used or detected in waters

Using MS characterization of the all the pollutants present in the sample

Unknown pesticide presents in the sample

Experiences gained - Conclusions - Recommendations Experience gained:

Determination of the level of pollution from pesticides of water bodies at risk due to agricultural pressures. Development of a Methodology with high QA/QC data.

Determination of new pollutants to be included in the Pesticides Program.

Conclusion:

A method based on isotope dilution GC/MS with automated SPE extraction was developed for the analysis of 32 pesticides and metabolites in surface water samples. Trueness was in the range 80-120% for 29 pesticides, precision below 15% for 25 compounds, method detection limit ranged from 1 to 9 ng/g and expanded uncertainties were < 40% for 24 pesticides.

93 Spanish surface water samples collected during summer and autumn 2004. Highest concentration and occurrence were found for atrazine, simazine, alachlor, terbutylazine and metoachlor included in the Priority List and/or Spanish Relevant List.

New pesticides were detected using MS characterization technique, the pollutant molinate and imazalil and are included in the future pesticide program.

Pesticides concentrations and occurrence are higher in the summer than in the autumn period. In summer, four pesticides were found in more than 50% of the analysed samples and four compounds were detected above the concentration level of 1 μ g/l (atrazine, terbutylazine, 3,4-dichloroaniline and fenitrothion), while in autumn percentage of detection was bellow 50% for all pesticides, only one compound (terbutylazine) exceeded 1 μ g/l.

Recommendations:

The surveillance of pesticides in water from agricultural activities needs the combination of 2 types of techniques. Standardized analytical methods with a high level of QA/QC to monitor specific pesticides, at least all the pollutants included in the Priority List and/or other National Relevant List in order to monitor the EQS compliance, and hence Chemical status. And the application of screening techniques to detect new pesticides from diffuse sources not included in common Lists in order to monitor ecological status compliance.

Outlook - Next steps - Accessibility of results/information

Title/Name of case study:

Conversion of pollutant concentrations measured in suspended particulate matter (SPM) into total concentrations in the whole water sample.

Type of case study:

Routine operation since 1990 in the water quality monitoring program of the international Rhine Commission (ICPR) for compliance checking of annual data of lipophilic pollutants with water quality targets.

Reporting Institution:

International Commission for the Protection of the Rhine (ICPR)

Web-Link:

http://www.iksr.de/fileadmin/user_upload/Dokumente/Berichte/IKSR_Bericht_Nr_143d.pdf

Objective of case study - background information -

In surface waters a number of the priority substances are adsorbed to SPM from 50 close to 100 percent. The water quality targets of the ICPR (analogous the EQS) for organic priority substances are expressed as total concentrations, that is the dissolved portion plus adsorbed portion of the substance in the whole water sample. For some organic priority substances the AA-EQS are very low, and the respective detection limits of the recommended analytical methods are insufficient. The objective is to support compliance checking with whole water EQS (or ICPR water quality targets) by conversion of SPM determinand concentration.

Contribution to <u>support</u> compliance checking with EQS

Specific contribution linked to WFD monitoring programmes

Surveillance monitoring design; specific monitoring of pollutants adsorbed to suspended particulate matter; compliance checking of SPM determinand concentration with whole water EQS.

Characterisation

For organic micropollutants like PCB, TBT, PAH or hexachlorobenzene, the ICPR had developed water quality targets for whole water. For reasons of the analytical method, surveillance of the quality targets was performed by sampling and analyzing these pollutants in suspended particles.

Sampling by centrifuge allows collecting the suspended material from several 1,000 litres of water within a few hours. Thus, a sufficient amount of SPM can be gained for chemical analysis, what corresponds to an enrichment from several 100 (or 1,000) litres of water. The amount of water centrifuged is recorded.

Course of procedure:

The concentration of contaminants is determined in $\mu g/kg$ dw and converted to whole water by means of the SPM content (in mg/L):

 $C_{Ti} = (Si \times C_{si}) \times 10^{-6}$

 C_{Ti} = total contaminant content on the day of sampling in μ g/L

- Si = SPM content on the day of sampling in mg/L
- C_{si} = contaminant content in SPM on the day of sampling in $\mu g/kg$.

This applies to substances that are adsorbed at SPM by more than 90 %. For substances that are adsorbed by 50 % at SPM, the value is multiplied by the factor 2:

 $C_{Ti} = 2 (Si \times C_{si}) \times 10^{-6}$

Other factors might be selected for a particular substance, if the partition coefficient is known.

The mean value (50- or 90-percentile in ICPR procedure) is calculated from the C_{Ti} values. Values below the limit of detection (limit of quantification) are included in the mean-value calculation (arithmetic mean) by the numerical value of the limit of detection. Then, the value of the arithmetic mean is indicated as "less than".

Experiences gained - Conclusions - Recommendations

Experience gained:

The ICPR member states have successfully applied this method at selected sampling sites since 1990. The error that results from the arbitrary definition of the conversion factor for adsorption between 50 and 90 percent is negligible against the other errors in trace analyses. But for EQS compliance checking procedure it is possible to define the adsorbed portion for each priority substance in 10-percent steps (50, 60, 70, 80, 90 or 100 percent).

Conclusion:

The collection of the SPM from several 100 (or 1,000) litres of surface water allows the compliance checking of the EQS for priority substances, which are partially or not dissolved in the water because of their hydrophobic and lipophilic properties. Sampling by centrifuge is time- and labour-intensive and should be applied in surveillance monitoring only at selected sampling sites at large rivers. But it is no problem to meet the minimum performance criteria for the quality of analytical results – also for low EQS values (e.g. tributyl tin). The results of the contaminant concentrations in SPM can be used for comparisons with the EQS, and - after a hydrological interpretation - they are also suitable for trend analyses.

Recommendations:

- For selected priority substances and for selected monitoring sites at large rivers the described procedure is suitable for compliance checking with EQS and for trend analyses. Special attention should be given to the following micropollutants: Pentabromodiphenylether, C10-13 chloroalkanes, Fluoranthene, Hexachlorobenzene, Pentachlorobenzene, PAH (Benzo(a)pyrene, Benzo(b)fluoranthene, Benzo(k)fluoranthene, Benzo(g,h,i)perylene, Indeno(1,2,3-cd)pyrene), Tributyltin and Trifluralin.
- the result of each measurement (spot sample) is converted into the total determinand concentration by means of the SPM content of the water and according to the percent factor of the adsorbed portion of contaminants;
- No further effort is necessary for analytical techniques to obtain a limit of quantification half of the EQS.

Outlook - Next steps - Accessibility of results/information

Title/Name of case study:

Comments Concerning the National Swedish Contaminant Monitoring Programme in Marine Biota

Type of case study:

Monitoring activities within the Swedish contaminant programme in marine biota

Reporting Institution:

Environmental Protection Agency (Sweden)

Web-Link: <u>www.naturvardsverket.se</u>

Main sources for further information; literature:

Comments Concerning the National Swedish Contaminant Monitoring Programme in Marine Biota, 2006

Objective of case study - background information:

The data of concern in this report represent the bioavailable part of the investigated contaminants i.e. the part that has virtually passed through the biological membranes and may cause biological effects. The main objectives of the monitoring program in marine biota could be summarised as follows:

• to estimate the levels and the normal variation of various contaminants in marine biota from several representative sites, uninfluenced by local sources, along the Swedish coasts. The goal is to describe the general contaminant status and to serve as reference values for regional and local monitoring programmes

• to monitor long term time trends and to estimate the rate of found changes.

• to estimate the response in marine biota of measures taken to reduce the discharges of various contaminants

• to detect incidents of regional influence or widespread incidents of 'Chernobyl'- character and to act as watchdog monitoring to detect renewed usage of banned contaminants.

• to indicate large scale spatial differences

• to explore the development and regional differences of the composition and pattern of e.g. PCB's, HCH's and DDT's as well as the ratios between various contaminants.

Contribution to...

Specific contribution linked to WFD monitoring programmes

Surveillance monitoring design and operational monitoring design as soon as EQS are developed for biota.

Description

<u>Substances monitored:</u> Metals, for example Hg, Cd, Pb and Cu and organic substances, for example PCB, DDT, Lindane, brominated flameretardants and dioxins.

<u>Sampling area</u>: The sampling sites are located in areas regarded as locally uncontaminated and, as much as possible, uninfluenced by major river outlets or ferry routes and not too close to heavy populated areas.

<u>Collected specimens:</u> For many species adult specimens are less stationary than sub-adults. To increase comparability between years, young specimens are generally collected. Only healthy looking specimens with undamaged skin are selected. The collected specimens are placed individually in polyethene plastic bags, deep frozen as soon as possible and transported to the sample preparation laboratory.

<u>Number of samples and frequency:</u> In general 20 individual specimens from the Baltic sites (reported to HELCOM) and 25 from the Swedish westcoast sites (reported to OSPARCOM) are analysed annually from each site/species. For guillemot eggs and perch, 10 individual specimens are analysed. Organo-chlorines in blue mussels are analysed in pooled samples containing about 50 individual specimens in each pool. Since 1996, samples from 12 individual specimens are analysed which is proposed in the revised guidelines for HELCOM and OSPARCOM.

Experience gained:

Continuous development of design for both a spatial and temporal monitoring programme and also increased knowledge of choice of matrix. The importance of quantifying objectives.

Conclusion:

Herring is the most commonly used indicator species for monitoring contaminants in biota within the BMP (Baltic Monitoring Programme) in the HELCOM convention area and is sampled by Finland, Estonia, Poland and Sweden. Herring muscle tissue is fat and thus very appropriate for analysis of fatsoluble contaminants i.e. hydrocarbons.

Cod is among the 'first choice species' recommended within the JAMP (Joint Assessment and Monitoring Programme) and BMP (Baltic Monitoring Programme). The cod liver is fat and organic contaminants are often found in relatively high concentrations. For that reason, it is also a very appropriate matrix for screening for 'new' contaminants.

Mussels are one of the most common used organisms for monitoring contaminants in biota. Adult mussels are sessile and hence it is easier to define the area the samples represent, compared to fish. Blue mussel is among the 'first choice species' recommended within the JAMP (Joint Assessment and Monitoring Programme).

Recommendations:

It is very important that the objectives of the monitoring are quantified before designing a monitoring programme. When the objectives are defined the choice of sampling location, matrix, sampling method and analytical procedure could cause problems if the proper guidelines are not followed.

Outlook - Next steps – Accessibility of results/information

Title/Name of case study:

SCREENING OF INDUSTRIAL AND CONSUMER CHEMICALS IN FINLAND (VESKA 1)

Type of case study:

Screening to select sites and substances for surveillance monitoring of WFD

Reporting Institution:

Finnish Environment Institute SYKE

Web-Link: http://www.ymparisto.fi

Main sources for further information; literature:

Objectives of case study:

To study the occurrence and concentrations of the substances chosen by risk assessment in the aquatic environment close to emission sources

To produce information for the purpose of emission source identification.

To develop the analytics used to examine harmful substances and to build up cooperation network between laboratories.

To develop a risk assessment method that can be used to estimate the circulation of the studied compounds in nature and to minimize the expenses needed for their monitoring.

To set a criterion that could be used to preclude or add substances to a more detailed monitoring plan.

Contribution to...

Specific contribution linked to WFD monitoring programmes

Supply data for national surveillance monitoring programme 2006-2008

Description

The study was carried out in the vicinity of twelve industrialized cities in Finland, in water systems downstream municipal sewage treatment plants. Concentrations were measured in sludge and waste water samples (1-3 occasions) and surface water (1-3 occasions), sediment (once, surface 2-3 cm) and fish (4-10 specimen pooled, Northern pike, *Esox lucius*) muscle samples.

Organotin compounds (both butyl and phenyl –forms) were found in all matrices, often exceeding EQS or "benchmark" values. This will trigger more research, monitoring and management activities MBT and DBT were found in sediments also below inland sewage treatment plants, indicating sources other than antifouling paints (stabilizer in plastics). TPhT indicated bioaccumulation: found less than TBT in sediments but more in fish

Alkylphenols, phthalates, organotin compounds, HCH, HCB and VOC-compounds were found in treated sewage water. Out of these, only DEHP and alkylphenols were found in excess of surface water EQS proposals.

In surface waters, only nonylphenol ethoxylates were found at the (national) EQS-level. Several substances were not detected (other alkylphenols, chlorobenzenes and volatile organic compounds). PAHs, PBDE, organotins and phthalates were, however, not measured in surface water phase. In sediments, organotins, PAHs, HCHs and dibutylphthalate were most commonly found.

Experience gained:

Conclusion:

Simultaneous screening of several substance groups might not be cost-efficient for industrial and household chemicals with several, poorly known sources.

On national level, surveillance monitoring will be started for alkylphenols, phthalates and PAHs in water (10 sites/year). First year (mid 2007- mid 2008) results will determine the need of continuation (rotation, cessation). Sediment and biota monitoring continues (5-15 sites, 2-6 yr rotation) for lipophilic substances.

Recommendations:

In addition to analytical problems at low concentrations, natural conditions in Northern Europe (lakerichness, low temperature, low particulate matter, low degradation, low population/water volume) would favour sediment and/or biota over water, in monitoring many industrial and household chemicals presently on the Priority Substances list (e.g. PBDE, HCB, SCCP, some PAHs, DEHP, TBT).

Effect-based monitoring should be developed and brought into the guidance for impact monitoring locations.

Outlook - Next steps – Accessibility of results/information

New groups of substances will be screened in sewage effluents and waste deposit leachates. The impact monitoring of selected compounds will start gradually along source identification and the renewal process of the permissions (industrial and municipal STPs) with statutory monitoring.

Title/Name of case study:

PESTICIDE SCREENING IN FINNISH SURFACE WATERS (VESKA 2)

Type of case study:

Screening to select sites and substances for surveillance monitoring of WFD

Reporting Institution:

Finnish Environment Institute (SYKE)

Web-Link: http://www.ymparisto.fi/download.asp?contentid=49817&lan=EN

Main sources for further information; literature:

Objectives of case study:

To study the occurrence and concentrations of agricultural pesticides in Finnish surface waters.

To produce information for identification of emission source.

To develop a risk assessment method that can be used to minimize the expenses of later monitoring.

Contribution to...

Specific contribution linked to WFD monitoring programmes Supply data for national surveillance monitoring programme 2006-2008

Description

Monitoring points: water bodies (streams and rivers) potentially at risk of pesticide pollution from agricultural sources

Matrix: surface water and the top of bottom sediment (0-1 cm)

Duration: 2004 - 2005

Frequency

- ➤ at one intensive site: weekly (May-Oct 2004) + monthly in winter (2004-2005)
- intermediate sites: monthly (Apr/May-Sep/Oct)
- areal screening sites: twice (Jun/Jul + Jul/Aug in 2005).

Analysed substance:

- 100 compunds from water samples using :multi-residue methods (GC-MS and LC-MS-MS) (including all pesticides in the EU Priority list and four of the six pesticides in the national list of pesticides)
- Tribenuronmethyl (a low-dose herbicide in the national list) from part of water samples using a specific analysing method
- The 14 pesticides analysed from sediment samples were selected basing on partitioning coefficients (from the EU and national lists of priority substances and other potential pesticides).

Number of samples: 190 water samples + 31 sediment samples

The study was carried out in two parts. The objective of the pilot year 2004 was to find the best timing for surface water sampling and to assist in the selection of the sampling sites for a survey for following year. In the second year (2005) the main objective was areal representativeness, but most of the sites were sampled only twice at the time of expected high concentrations.

In 2004 the focus was in upstream sites (catchment areas: 1-100 km²) close to fields. Surface water was collected in an intensive site and in 6 other agricultural streams. Additional samples were collected once from 6 rivers (Sep/Oct). The pilot year indicated that agricultural land use percent was more important factor to explain pesticide detections than catchment area.

In 2005, sampling sites were selected from watershed register using agricultural land use percent as the

main criterion: 35 agricultural sites and five reference sites from non-agricultural areas (catchment areas of the sampling points: .50 - 37 000 km²). In addition, six rivers and two of the smaller streams sampled during the pilot year were sampled monthly (May-Oct) in 2005.

Pesticides in water samples (70 (in 2004) + 120 (in 2005) + quality assurance samples): The number of different pesticides detected was 24 in 2004 and 46 in 2005. Detected concentrations were mainly low (traces). In 2005, one or more herbicides were detected in 90% of water samples. Phenoxy acid herbicides were most frequently found compounds. This may be explained by the volume of they use in Finland. Only following few exceedings of EQS values were observed during the screening, although the sampling was focused in the expected concentration peaks and in the areas of high pesticide loadings:

- Endosulfansulphate was detected in one site (3 detections /3 sampling times) (max $0.02 \mu g/l =$ quantification limit), while suggested EQS-value is $0.005 \mu g/l$.
- The detected peaks of MCPA concentration exceed national EQS value (1.6 µg/l) in several sites (max 8.8 µg/l), but concentrations were still lower than MacQS (15 µg/l).
- In single samples the concentrations of low-dose herbicides (thifensulfuronmethyl, tribenuronmethyl) were higher than Swedish "target values". However, the Finnish national EQS of tribenuronmethyl was not exceeded.
- Insecticides were detected seldom, which is inline with the sold amounts of them. Pirimicarb was found in one site and its concentration was higher than Norwegian limit value.

Pesticides in sediment samples (31 + a parallel sample):

In 2005, the top of bottom sediment (in all sites where possible) was sampled and 14 substances were analysed and 6 detected from sediment samples samples. Atrazine (forbidden in 1991) was detected in more than half of the sediment samples, prochloraz in every third sample, and the others in single samples. Currently there are no confirmed sediment EQS values for pesticides, but the earlier proposed values were exceeded in case of atrazine, prochloraz, endosulfan and its metabolite endosulfansulphate.

Conclusion:

Simultaneous screening of several substances was cost-efficient for pesticides. High number of different pesticide compounds, mainly herbicides and their metabolites were detected. Concentrations exceeded seldom levels of potential environmental threat.

The time and site of sampling have significant effects on the results. No watershed specific information about pesticide usage is available. Agricultural land use percent was a good tool in selecting sampling sites, but further information about the type of agriculture (e.g. typical crops) would help in finding the risky sites.

The intensive agricultural land use fraction is usually rather low in big Finnish catchments and pesticides are diluted in water from non-agricultural areas in big rivers while pesticides may occasionally pose local problems in some small streams close to fields.

Outlook - Next steps – Accessibility of results/information

For year 2007 the statutory pesticide monitoring is planed to be performed in 8 rivers and in one stream (10-12 samples/site). Later (in 2008-) it will be reasonable to include some upstream sites to monitoring program.

Title/Name of case study:

FRENCH NATIONAL MONITORING NETWORK (RNO)

Réseau National d'Observation de la qualité du milieu marin

Type of case study:

A 30 years Chemical monitoring network (surveillance and operational monitoring)

Reporting Institution:

IFREMER (Institut français de recherche pour l'exploitation de la mer)

French Research Institute for Exploitation of the Sea

Web-Link: www.ifremer.fr and www.ifremer.fr/envlit/surveillance/rno.htm

Main sources for further information; literature:

Annual bulletin may be downloaded at : www.ifremer.fr/envlit/surveillance/rnopublis.htm

Objective of case study - background information:

1. Assessment of the levels and trends of the chemical contamination of the French coastal seas.

2. To meet OSPAR and Barcelona Conventions monitoring obligations

Contribution to...

Specific contribution linked to WFD monitoring programmes Surveillance and Operational Monitoring

Description

Chemical contaminants monitored :

- metals : Ag, Cd, Cr, Cu, Hg, Ni, Pb, V, Zn
- organochlorinated compounds: DDT compounds, HCH isomers, PCB congeners
- PAH's : 16 US-EPA PAH's and some alkylated PAH's

Strategy of monitoring :

The RNO is largely based on bivalve molluscs (mussels and oysters) which are used as quantitative indicators of contamination. Some 90 sampling points are sampled twice a year for metals and once a year for organic contaminants. Analyses are carried out at a single laboratory, the Ifremer Center of Nantes. An aliquot of the samples is systematically archived for possible controls at a later date or the retrospective search of other contaminants (the bank contains more than 8000 samples, since 1981).

The contaminants are also measured in the sediments. The first centimetre of the surface sediments can integrate several years of contamination. A yearly sampling cruise is carried on 1/10 of the French maritime frontage, the whole French littoral being covered every 10 years. The contaminants measured are the same than in biota, in addition to descriptive and normalization parameters (grain size, organic carbon, carbonates, aluminium, iron, lithium and manganese). Some sediment cores are collected to reconstruct the history of the contamination over several decades.

Experience gained:

Direct monitoring in water is not used any longer because it is too expensive and not reliable. Indeed, the operational difficulties to collect valid samples (i.e. uncontaminated and representative) for analyses of substances present in water at trace levels are too great. First, collecting uncontaminated samples needs a care difficult to reach in routine monitoring. Secondly, the spatial and temporal representativeness of samples collected in Water Bodies affected by tides and currents, are too low. This is our first-hand

experience of direct measurements of contaminants in water which was carried out between 1979 and 1984 and abandoned for the reasons above. This is coherent with the OSPAR Convention monitoring programme which is also based on biota and sediment.

Conclusion:

To assess the levels of contamination by metals and organic hydrophobic substances, it is highly preferable to use accumulative matrixes that present higher levels and allow temporal integration of the natural water variability.

Recommendations:

Within the frame of the WFD, Surveillance Monitoring should be advantageously carried out in sediments for the metals and hydrophobic substances. One survey by Management Plan (6 years) would be sufficient to assess the quality of the Water Bodies.

Operational Monitoring needs to assess temporal trends of the contamination and should be carried out in biota (mussels as a first choice). One survey per year (synchronous with biological life cycle of the biota) would make possible to evaluate the trends.

Outlook - Next steps – Accessibility of results/information

If remer developed a Web site including a broad section devoted to marine monitoring (<u>www.ifremer.fr/envlit/surveillance</u>). It is possible to view and to download the data of contaminants in molluses for each sampling point.

Title/Name of case study:

What concentrations of hazardous substances do we find in the environment? Results from the Swedish Screening Programme 2003-2004 (Abstract)

Type of case study:

Screening

Reporting Institution:

Environmental Protection Agency (Sweden)

Web-Link: <u>www.naturvardsverket.se</u>

Main sources for further information; literature:

What concentrations of hazardous substances do we find in the environment? Results from the Swedish Screening Programme 2003-2004 Report 5524, February 2006.

Objective of case study - background information:

For the purpose of obtaining information regarding the concentrations of newly discovered persistent organic pollutants (POPs), as well as other potential problem substances used in society, the national environmental monitoring scheme was supplemented by a screening programme a few years ago. In this programme, one or more selected substances are measured on one or more occasions during a single year and in different media, such as sewage, fish or air.

Screening was initiated on a small scale in 1996-97 and has gradually increased in scope since then. The reason for including a substance in the screening programme may be that it is used on a large scale, that it has been prioritized in various international contexts, or that it has attracted national attention for other reasons.

Contribution to...

Specific contribution linked to WFD monitoring programmes

Data for substances those are not included in the regular monitoring programme.

Description

The screening studies have been modelled on the **DPSIR** strategy. This strategy analyses environmental problems based on **D**rivers such as those created by industry, **P**ressures on the environment such as polluting discharges that harm the **S**tate of the environment, which in turn results in an **I**mpact on human health and the environment, whereby we try to find **R**esponses or actions to deal with the problems.

A screening study consists of the following closely-connected parts:

Choice of substance Preparatory theoretical study Measurement study Evaluation

Experience gained:

The primary purpose of the screening is not to support research on new POPs, but to comply with requirements on reporting of certain substances in various EU directives and international conventions. Many of these substances are of no relevance for Sweden, and the screening could help to show this so that resources do not have to be wasted on measurement of these substances.

Draw attention to new environmental pollutants. Informing the public without frightening them is important, but perhaps even more important is furnishing data to those who work with chemical risk reduction, i.e. various public authorities.

Conclusion:

Further studies should be made of the following substance groups Organophosphates Bisphenol A α - and β -endosulfan Siloxanes Chloro- and bromostyrenes

The following substance group should be subjected to another screening study in a few years Adipates

The following substance groups do not have to be followed regularly Chlorinated paraffins Limonene Mirex Isocyanates

Recommendations:

Will be based on the results obtained for each substance/group.

Outlook - Next steps – Accessibility of results/information

Data are available on the Internet from the environmental monitoring scheme's data host for screening. http://www.ivl.se/english/ivlstartpage/rightmenu/environmentaldata.4.360a0d56117c51a2d30800064209.html

Title/Name of case study:

MONITORING USING PASSIVE SAMPLING DEVICES TO IMPROVE TRACE METAL-RELATED RISK ASSESSMENTS

Type of case study:

Complementary monitoring by *in-situ* deployment of passive sampling to help reduce the uncertainty associated with infrequent grab sampling for compliance monitoring

Reporting Institution:

University of Portsmouth (UK), Bureau de Recherche Géologique et minière (Fr)

As part of the EU-funded FP6 project *Screening Methods for Water Data information in support of the implementation of the WFD (SWIFT-WFD)*

Web-Link: none

Main sources for further information; literature:

"Evaluation of the performance of the Chemcatcher and DGT passive sampling devices for monitoring heavy metals in water" Allan IJ, Knutsson J, Guigues N, Mills GA, Fouillac A-M and Greenwood R, (in preparation) (2007)

Objective of case study - background information:

Demonstration of the applicability of passive sampling to increase confidence in measures of water quality provided by infrequent spot (bottle) sampling campaigns.

Grab or bottle sampling followed by filtration at $0.45\mu m$ is to be used for compliance checks (AA-EQS & MAC-EQS).

Whilst monthly spot samples provide a precise estimate of concentrations of contaminants at the time of sampling, there is uncertainty concerning conditions prevailing in the periods between sampling events. This is of particular concern where concentrations are known to fluctuate or where there are significant or potential natural or anthropogenic pressures. One possible solution to this problem is to deploy passive samplers to provide time-weighted average (TWA) concentrations of concentrations of contaminants in the period between grab samples. This may confirm or contradict the data from the routine monitoring campaigns, and should help to reduce the possibility of making erroneous decisions in risk assessments required in the implementation of the WFD.

In this application the TWA concentrations of metals estimated using DGT and Chemcatcher samplers were compared with estimates based on spot sampling in the Meuse River (Eijsden, The Netherlands). In this trial 2 different grab sampling procedures were used at relatively high frequencies (one to three times per week) and metal analyses conducted in two different laboratories in a pilot-scale inter-organisational comparison that incorporated both the analytical determination and the sampling step.

Contribution to...

Specific contribution linked to WFD monitoring programmes

Surveillance and operational monitoring tasks

Description

Matrix: Water

Passive sampling: Use of the Diffusive Gradient in Thin Film (DGT) and Chemcatcher sampling devices following guidelines provided in the BSI Publicly Available Specification 61, and analysis of sampler extracts by ICP-MS in an university research laboratory.

Sampler exposure: Consecutive and/or overlapping 7, 14, 21 and 28 day periods

Grab sampling protocol 1: Routine weekly sampling, transport of the sample to the laboratory followed by filtration (0.45 μ m) and ICP-MS analysis.

Grab sampling protocol 2: Sampling every two or three days, on-site filtration (0.45 µm) and analysis by

ICP-OES in an accredited laboratory.

Experience gained:

There was good agreement between the results for cadmium, copper, nickel and zinc provided by the two types of samplers, despite differences in their working principles. High and fluctuating concentrations of cadmium (between proposed MAC-EQS and AA-EQS thresholds) were detected by passive sampling during the first 14 day exposure period. This was confirmed by the relatively high frequency of grab sampling used in this trial, but may have been missed if only conventional sampling rates had been used. Extra information, on speciation of the metals, was provided by the samplers, and this was consistent with predictions obtained using equilibrium speciation modelling with visual MINTEQ (NICA-Donnan model).

Conclusion:

Reliability and consistency of TWA metal concentrations measured by passive sampling were shown when compared with 2 distinct grab sampling protocols in a procedure that included uncertainty both on the sampling and analytical measurement steps. This provided representative information on average concentrations. On the other hand an estimate based on a single grab sample could have provided misleading information since for instance levels of Cd varied over a factor of five during the trial.

Recommendations:

Passive sampling devices should be deployed following BS PAS 61 guidelines and manufacturer's specifications for period up to 14 to 20 days to achieve representative sampling. This should be combined with grab sampling to reduce monitoring uncertainty to manageable levels.

Outlook - Next steps – Accessibility of results/information

These results will be published in detail in a peer-reviewed scientific journal.

A desirable extension of this work would be to increase the duration of the trial to two or three years in order to provide a reliable, long-term comparison between the average concentrations of contaminants estimated by routine grab sampling, and passive sampling. This would establish the utility of the two methods either on their own or in combination to quantify trends in trace metal concentrations over time. Since the two methods measure different (operationally defined, and water body specific) fractions of metals, relationships between the concentrations in filtered bottle samples and those measured by passive samplers need to be established.

Other future developments may include testing and optimising combinations of grab and passive sampling in order to improve sampling representativeness while ensuring a reduction in monitoring costs.